1	- VOLUME 2 -
2	IN THE UNITED STATES DISTRICT COURT
3	IN AND FOR THE DISTRICT OF DELAWARE
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5	DAD DUADMACEUMICAI INC CIVII ACMION
6	PAR PHARMACEUTICAL, INC., : CIVIL ACTION PAR STERILE PRODUCTS, LLC, : and ENDO PAR INNOVATION :
7	COMPANY, : Plaintiffs, :
8	: : :
9	vs. :
10	EAGLE PHARMACEUTICALS INC., : : NO. 18-823-CFC-JLH
11	Defendant. : (Consolidated)
12	PAR PHARMACEUTICAL, INC., : CIVIL ACTION PAR STERILE PRODUCTS, LLC, :
13	and ENDO PAR INNOVATION : COMPANY, LLC, :
14	: Plaintiffs, :
15	:
16	vs. :
17	AMNEAL PHARMACEUTICALS OF : NEW YORK, LLC, et al., :
	:
18	Defendants. : NO. 18-2032-CFC-CJB
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20	milainata p
21	Wilmington, Delaware Thursday, July 8, 2021
22	8:40 o'clock, a.m.
23	
	BEFORE: HONORABLE COLM F. CONNOLLY, Chief Judge
24	
25	Valerie J. Gunning Official Court Reporter

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1 PROCEEDINGS 2 3 (Proceedings commenced in the courtroom, 4 beginning at 8:40 a.m.) 5 THE COURT: All right. Good morning. Please be 6 7 seated. All right. Anything I need to attend to before we 8 get started? 9 MS. WACKER: We're going to start with Dr. Park 10 He's doing noninfringement and then invalidity. now. 11 are a couple of exhibits and slides that Par has objected to 12 on the invalidity section, so I'm not sure, Your Honor, if 13 you want us to go forward. 14 We can do noninfringement and deal with those objections before we move to invalidity or we can deal with 15 16 it as we get to the slides, whatever you prefer. 17 THE COURT: Maybe it's cleaner to do it all at 18 once. Mr. Black, what do you think? 19 The main issue I'm interested in is MR. BLACK: 20 testimony on lot 788435, for which there's no support. can address it when we get to that slide. 21 22 THE COURT: Okay.

MS. WACKER: I thought there were two. That

I'm not sure if there are any

BLACK:

MR.

objections I'm actually aware of.

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makes my life easier.

THE COURT: Did you all look at the issue of the subject matter jurisdiction? I want to make sure. Is it the reformulated version that is the listed drug?

MR. BLACK: Mr. Goldberg is going to address jurisdiction under 271(e)(1) and I will address jurisdiction under 271(a) and (b).

THE COURT: Okay. Now, are you prepared to talk about this? I mean, you don't think there's an issue with jurisdiction?

MS. WACKER: I don't think -- we've analyzed this and we don't think there is an issue based on our assessment as well.

THE COURT: Okay. So do you agree that the listed drug is a reformulated version?

MS. WACKER: It is.

THE COURT: Oh.

MS. WACKER: My understanding is that because the reformulated drug was filed under the same NDA, that there was a requirement that Eagle certify the patents that were listed even though we were certifying on the earlier formulation. That's just a factor of the Hatch-Waxman statute that required us to do that.

THE COURT: Mr. Goldberg, did your client certify to the FDA that the original Vasopressin read on the

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two asserted patents in this case? MR. GOLDBERG: No, Your Honor. We listed the patents for the NDA and the NDA covered its current product, and we assert that the patents-in-suit cover the current product. THE COURT: But when you filed the NDA, the product that was at issue when you filed was the original Vasopressin. Is that right? MR. GOLDBERG: The original NDA was for original Vasopressin in 2014. THE COURT: You're saying the same NDA then was revised? MR. GOLDBERG: Exactly. THE COURT: To cover the reformulated Vasostrict? MR. GOLDBERG: Yes. It can be amended and supplemented, and when you do that, you submit your supplement. Some can include listing new patents. THE COURT: And when you supplement the NDA to effectively change the product, I mean, that's what was done here. Right? It changed from the original Vasostrict to the reformulated Vasostrict? MR. GOLDBERG: Yes, in some of the supplements. THE COURT: Do you actually say explicitly to the FDA, we're changing the product, or is it, no, it's the

same product, because it's the same NDA number?

MR. GOLDBERG: Yes. There are regs that explain the different types of supplements you submit. There are some more major than others. There are regs that govern that.

In general, you submit your supplement with the data you would need in order to have that product change.

In this case it was the reformulated version and the supplement was submitted. I forget exactly what type of amendment that was, but there was an amendment to the NDA saying we're changing the formulation.

I can also add that in terms of jurisdictional issue for 271(e)(2), it really doesn't depend on the RLD. You know, what Eagle identified as the RLD. The statute just says it shall be an act of infringement to submit an application under 505(j), that's an ANDA of the Federal Food, Drug & Cosmetic Act or described in Section 505(b)(2) of such act for a drug claimed in a patent for the use of which is claimed in a patent.

So here, Par filed a complaint against Eagle alleging that Eagle infringed the '785 and '209 patent by submitting its ANDA and nothing more is required. The case that's controlling here is Vanda Pharmaceuticals versus Westward and the cite is 887 F.3d. 1117, and I'm looking at the page 1124. And it just says in that case, here, Vanda

in its complaint alleged that Westward infringed '610 the patent under 35 U.S.C. 271 by filing the ANDA.

The same situation here. Nothing more was required to establish the District Court's subject matter jurisdiction pursuant to 28 U.S.C., 1338(a).

I have a copy of the case if Your Honor would like it as well.

THE COURT: No. You are saying it doesn't matter, I mean, when during the course of the ANDA the patent was listed in the Orange Book?

MR. GOLDBERG: No, it does not.

MR. BLACK: As it happens, Your Honor, these patents issued after the formulation change, so the formulation changed and the product was launched and then the patents issued.

The patents were listed in the Orange Book. The FDA required Eagle to send us a paragraph 4 certification.

They sent it. The filing of the ANDA is an act of infringement. It creates jurisdiction.

There's also jurisdiction separately and independently under 271(a) and (b) because we're now at the point where Eagle has said if they get approval, they will launch.

So the Court has jurisdiction to decide the declaration of what would constitute infringement. It

usually doesn't come up in the ANDA cases because if you have a finding of infringement under 271(e), the statute requires that the FDA convert to tentative approval until the expiration of the patent. That's mandatory relief. But if we fail on that point, on 271(a) and (b), the Court has jurisdiction and will ask for sort of a lesser included offense, if you will, a declaration that if they do certain things that would constitute infringement, like if they make a product in the top end of the range because the evidence is that that will float into the range during the product life.

That wouldn't give us an automatic right of injunctive relief, but that's an issue we have to deal with if they actually did launch. But we would be entitled to that declaration. And Your Honor would be discharging the Article III duties to determine the dispute between effectively there was a conflict between the regulations governing patent law and the FDA law.

So with my overlapping circles, if the PTO has issued a claim which would be infringed by an FDA approval, 3.64, you would have the authority, and I would ask that you issue a declaration, and that would constitute infringement to the extent upward drift would create an infringement during the life of the patent.

THE COURT: All right. And just also I think

what you are saying also is that the product for the NDA, although it started out as original Vasostrict, it morphed into reformulated Vasostrict.

MR. BLACK: Right, Your Honor. Just as you have seen their ANDA -- you file a document with the FDA. It's not a document, it's a gigantic, gigantic filing with many, many pieces to it and it's a living document. Both during the ANDA process they keep changing their specs and reporting to the FDA.

At some point along the way the FDA is saying not good enough, not good enough. Your impurity data is not good enough. It's not the same as the original Vasostrict. You need to deal with these issues. And then eventually, maybe they'll get an approval, and then the approval sticks, but they may make changes along the way.

They've said, for instance, that after they get approval, they are going to make -- they're going to file a separate approval to get these batches made on a different product line because the line that they've been planning to make it on is being taken up by Covid vaccines and that's going to require separate approvals and who knows what will happen when those batches are actually made. But it's a living document.

THE COURT: It's a living document, and I use, in terms of figuring out what the relevant product is, I use

1 the product that is most current? 2 MR. BLACK: Yes. The law is clear that you have 3 to consider the ANDA as it stands. 4 THE COURT: Well, and then this is both the NDA 5 and ANDA? MR. BLACK: Well, no, because there's no 6 7 relevance to the NDA to infringement or validity. You don't have to make --8 9 THE COURT: Hold on. There's no relevance. 10 thought there was relevance because it has got to be --11 you're listing the patent. You're asserting that the ANDA 12 product reads, that the NDA product reads on the patent. 13 MR. BLACK: Well, we're not required to prove 14 that. I mean, the --15 THE COURT: Isn't that -- I thought that's the 16 whole triggering device. In order to get ANDA --17 MR. BLACK: Right. But we don't have to prove 18 it. We listed our product on the Orange Book. 19 Well, I'm not saying you have to THE COURT: 20 come in and prove it here in court. 21 MR. BLACK: Yes. THE COURT: I'm saying to have jurisdiction, the 22 23 starting point is that you list these patents in the FDA and say that our product, the NDA, reads on these patents. 24 25 That's what gives rise to Hatch-Waxman jurisdiction in the

1 first instance. Right? 2 MR. BLACK: Yes. Yes. 3 THE COURT: Okay. I just wanted to make sure. I'm just saying they've been 4 MR. BLACK: Yes. 5 conflating the infringement, validity and NDA filing evidence for rhetorical purposes, but for purposes of how 6 7 this is going to be reviewed in the Federal Circuit, the only issues are does their product infringe under standard 8 infringement analysis, and, B, is the patent valid and 9 10 enforceable? Neither of those questions entail any analysis 11 of the NDA. It's their product. 12 So the only thing is, and this THE COURT: 13 is what strikes me, is that you're saying that your 14 product, when I figure out what your product is for the NDA, it's the most current version that has been approved by 15 the FDA. 16 17 MR. BLACK: Right. So when I look at the ANDA 18 THE COURT: Right. 19 that is at issue here and figure out what the product is, I 20 guess I ought to be looking at what's the most current 21 version of the product that's before the FDA. 22 Well --MR. BLACK: 23 THE COURT: That seems to be a little bit at 24 odds with what your experts were saying.

MR. BLACK:

I don't think so.

I don't think so

at all, Your Honor. The issue for infringement, I'm not trying to be -- maybe I'm being a little obtuse today. I just don't quite understand where you're coming from.

The infringement analysis is if they make the product that they proposed in the ANDA, their current ANDA specs, not what they may have filed at the beginning, but their current ANDA specs, if they make that product and sell it, will it infringe our patent? That's the infringement analysis.

THE COURT: Right. As I understand it, it's undisputed as a factual record, and please correct me if I'm wrong, that the current version of the product is made pursuant to the optimized manufacturing process. Right?

MS. WACKER: Yes.

MR. BLACK: That is what --

THE COURT: Which doesn't cover SVA1. Right?

MR. BLACK: That is not right. I don't agree with that at all, Your Honor.

THE COURT: Okay.

MR. BLACK: They're still representing that as the exhibit batch, as the registration batch, the three critical batches they had to file to get this approved, and the data from SVA1 is still part of the process, if they meet the release spec on SVA1, they can sell it tomorrow.

	Park - direct
1	THE COURT: Okay.
2	MR. BLACK: If it expired
3	THE COURT: We'll just have argument on that.
4	MS. WACKER: Okay.
5	THE COURT: We'll flesh it out.
6	MS. WACKER: I disagree with those statements.
7	THE COURT: Let's then go forward.
8	MS. WACKER: Okay. Eagle calls Dr. Park.
9	DR. KINAM PARK, having been duly
10	sworn/affirmed as a witness, was examined and testified as
11	follows
12	THE COURT: All right.
13	THE WITNESS: Good morning.
14	MS. WACKER: Your Honor, may I proceed? Your
15	Honor, we have some binders that we need to bring up.
16	THE COURT: To the witness? Sure. Thank you.
17	DIRECT EXAMINATION
18	BY MS. WACKER:
19	Q. Good morning, Doctor.
20	A. Good morning.
21	MS. WACKER: Your Honor, is it okay to proceed?
22	THE COURT: Please.
23	BY MS. WACKER:
24	Q. Could you please introduce yourself to the Court.
25	A. My name is Kinam Park. I'm currently a Showalter

1 distinguished professor at Purdue University.

- Q. Can you briefly tell the Court what you will be testifying about today?
- A. I'll be testifying today that Eagle's product does not infringe the claims of the asserted patents.
 - Q. And did you help prepare --
- A. I'm sorry. I'm not done. The asserted patents are invalid and the 2014 Vasostrict label is material to the patents.
- 10 Q. And did you help prepare any demonstratives to assist with your testimony today?
- 12 A. Yes, I did.

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- Q. Can you briefly provide background of your overview of your educational background?
 - A. Yes. I obtained my B.S. in pharmacy from Seoul
 National University in Seoul, Korea, and after serving in
 the Korean military as a lieutenant, I came to the United
 States, the University of Wisconsin, Madison, and obtained
 my PhD in pharmaceutics in 1983, and I did two years of
 post-doctoral training in chemical engineering before I go
 to become a professor.
 - Q. Can you briefly tell the Court about your career after receiving your Ph.D.?
- A. I went to Purdue University in 1986. Promoted to full professor in 1994. Since 1998, I have been appointed to the

department of biomedical engineering.

In 2006, I was promoted to the Showalter distinguished professor of biomedical engineering. In 2001, I started a company called Akina Incorporation, and presently, I serve as the president of the company.

- Q. And what is Akina Incorporation?
- A. Akina Incorporation is a company that sells specialty polymers that are used for a variety of formulations and biomedical devices and currently, we have contracts with the FDA and many other companies.
- Q. We've heard yesterday that this case is about peptide formulations. What experience do you have with peptide formulations?
- A. Yes. Over the years I have made many peptide formulations, protein formulations, especially for long-acting injectable formulations. For peptide formulations, I made leuprolide, goserelin, octreotide, triptorelin. Protein formulations, I made insulin formulations, botulinum toxin, which is commonly known as botox, Concanavalin A, lysozyme, fibrinogen and transferrin.
- Q. In your binder, you have DTX-282-A. Is this a current and accurate copy of your CV?
- A. Yes except a few more publications I added in.

25 MS. WACKER: Eagle tenders Dr. Park as an expert

Park - direct 1 in pharmaceutics and chemical development. 2 MR. BLACK: No objection. 3 THE COURT: All right. BY MS. WACKER: 4 5 Let's turn to noninfringement. What do you understand to be the asserted claims in this case? 6 7 The '209 patent, claims 1, 4, 5 and 7, and the '785 Α. patent, claims 1, 5 and 8. 8 9 And here we have the defendants' claims from the '209 10 and '785 patent. Can you just generally give an explanation 11 of what those claims are directed to? 12 Well, the '209 patent is related to a method of 13 treatment using specific Vasopressin formulations having a 14 specific impurity profile and '785 patent relates to pharmaceutical compositions of Vasopressin formulation with 15 16 their profile. 17 And why is that an X next to the pH 3.7 to 3.9 18 formulation element here? 19 As we heard yesterday, Eagle made a formulation having Α. 20 3.4 to 3.6, so Eagle does not make a formulation having pH 21 3.7 to 3.9. And what legal framework did you apply in coming to 22 23 your opinions in this case with respect to infringement?

I applied the legal framework that follows. Direct

infringement, accused direct infringer must literally

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perform each step of the claimed methods. This is the '209 patent and an accused direct infringer must make, use, sell, offer to sell, or import claimed compositions in the U.S., the '785 patent.

Eagle is not making formulation with the '209. For indirect infringement, induced infringement, normally, direct infringement by a third party. Knowingly induce direct infringement by a third party.

- Q. Okay. Now, did you hear the testimony of Dr. Kirsch yesterday with respect to infringement?
- 11 A. Yes, I did.

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- Q. And do you agree with his opinion that Eagle's products, if approved and sold, will infringe the asserted claims in this case?
- 15 A. No, I do not.
 - Q. All right. So let's turn to the first part of your opinion. What is your first opinion with respect to noninfringement?
- A. My first opinion is that Eagle's ANDA requires a noninfringing pH, so it will not go to 3.7 to 3.9.
- Q. And here on slide 10 we have DTX-131. What is DTX-131?
 - A. This is Eagle's ANDA submission to the FDA.
- Q. And if we look at this ANDA submission, on the left-hand column, the first one that's highlighted that says

Vasopressin injection, USP, what do you understand that to be representing?

- A. This is Eagle's proposed generic product to the Vasopressin injection.
- Q. And then next to that where it states Vasostrict,

 Vasopressin injection, initial NDA, approved April 17, 2014,

 what is that representing?
- 8 A. I now understand this is a reformulated Vasostrict.
- 9 0. The middle column?

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- 10 A. Middle columns is Original Vasostrict.
- 11 Q. Sorry. Let me make sure we have this accurate for the
 12 record. So the middle column is highlighted Vasostrict
 13 injection, initial NDA approved April 17th, 2014. What is
 14 that column representing?
 - A. This column represents the RLD, which is known as reference listed drug, and Eagle is making copy of this original Vasostrict.
- Q. So the middle column, can we agree that's original
 Vasostrict?
 - A. Yes. Again, we talk about either original Vasostrict or 2014 Vasostrict.
- Q. Okay. Are there any differences here between Eagle's proposed Vasopressin injection product and the original Vasostrict product that was approved in April 2014?
- A. There was no difference. That's the point of this

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ANDA. As we can see, condition of use is exactly the same. Active ingredients, exactly the same. Inactive ingredients, water for injection, acetic acid, chlorobutanol, exactly the That's the whole point of the product. same. You mentioned RLD. What is an RLD? 0. RLD is again reference listed drug. Whenever new drug applications are approved, that drug automatically becomes RLD because then generic companies will make a copy of the That's why they call it reference listed drug. Okay. And now on the right-hand column, the one that's not highlighted, Vasostrict, Vasopressin injection, supplement three, approved March 21, 2016, what do you understand that to be representing? That is again reformulated Vasostrict, which Par Α. revised from the original Vasostrict. And is Eagle using that reformulated formulation as an RLD? Again, at the bottom highlighted, Eagle specifies Α. the original version of Vasostrict, approved April 17, 2014, as the RLD. It's clear. 0. All right. And does the reformulated Vasostrict have the same pH information as the original Vasostrict and Eagle's product? No. We can compare at the bottom, second from the

bottom row. I can see that acetic acid should be adjusted

1 to 3.4 to 3.6 original Vasostrict. The right side says

- 2 sodium acetate buffer to adjust pH to 3.8. There's a
- difference in pH. Also, reformulated Vasostrict does not
- 4 contain chlorobutanol.
- 5 Q. All right. Slide 11, we have an excerpt from DTX-327
- at page one. What is this document, Dr. Park?
- 7 A. This is, again, ANDA documentation to the FDA.
- 8 \ Q. Okay. And what is being shown here on page 1 of
- 9 | DTX-327?
- 10 A. This shows that the stability is 3.4 to 3.6.
- 11 Q. This is for Eagle's proposed ANDA product?
- 12 A. Yes, it is.
- 13 Q. And what is a release specification?
- 14 A. Again, it clearly indicates 3.4 to 3.6.
- 15 | Q. What is a release specification?
- 16 A. Release specification is pH before they release the
- 17 product.
- 18 Q. And what is a stability specification?
- 19 A. A stability specification is the specification that
- 20 should be met during the shelf life.
- 21 Q. And so based on these specifications, what is your
- 22 | understanding of what Eagle is telling the FDA about its
- 23 proposed ANDA products, the pH of its proposed ANDA products
- 24 | through its shelf life?
- 25 A. Well, Eagle is reporting to the FDA that Eagle's

product will maintain pH of 3.4 to 3.6 throughout its shelf
life.

- Q. And based on the information you have reviewed about Eagle's product, would you expect Eagle's product to meet these specifications?
- A. Yes. All the data we have from yesterday, all the pH we have seen, yes, they are between 3.4 and 3.6, so I say that Eagle's product will maintain pH between 3.4 and 3.6.
- Q. At the bottom of the slide, there's also reference to DTX-678, page 2, and PTX-1427 at page 1. Do you understand that these are just supplemental ANDA submissions that contain the same specification?
 - A. Oh, yes. Eagle submitted the revised updated version all the time, but this table, the information remained the same.
 - Q. All right. Now, did you look at the data for Eagle's, the pH data for Eagle's proposed product?
- 19 A. Yes, I have.

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- Q. Okay. Did you look at all of the pH data that is available for the batches that have been made of Eagle's ANDA product?
- A. Yes. I think everybody saw it, too. We have pH data from batch 1 all the way to batch 17.
- 25 Q. And do you agree with Dr. Kirsch, that the data that

1 Eagle has on the batches that have been made would establish 2 infringement of the asserted claims? 3 I don't. We all saw the data. Data does not support Α. that. 4 5 And approximately how many pH measurements have been Q. 6 taken? 7 Α. Roughly, I think around 350. I want to talk a little bit about what batches have 8 9 been made, and so here on slide 14, you describe a number of 10 the batches and what they are called in the FDA documents, 11 so maybe we could talk about that a little bit. 12 On the left-hand side here at the top, there is 13 a discussion of registration batches, SVA1 through 3. 14 do you understand those to be and when were they manufactured? 15 16 Those are the batches that Eagle or any company used 17 to report to the FDA that this is a registration batch, 18 that we're going to prove that we can present the final 19 product. 20 And so the registration batches, were those made according to the current proposed commercial process? 21 Actually, that's not. That's why I have a 22 No. 23 dotted line there to indicate the manufacturing changes. 24 So they actually at the time submitted to the FDA, it was.

Now that the manufacturing process changes, especially for

1 pH, the batch 1 does not represent pH of the final product. 2 And below that, you have characterization batches, 3 SVA4 through 6. What are the characterization batches and when were they manufactured? 4 5 Well, Eagle still made those representative batches. FDA has some question, so Eagle made new characterization 6 7 batches to provide answers to the FDA. And when were those manufactured? 8 Q. 9 Those were manufactured in March 2019. 10 And then you have a dotted line here where it says 11 manufacturing change. Were those two, the six batches, 12 registration and characterization batches, made according to 13 the new optimized manufacturing process? 14 Right. So as I mentioned, optimization batches were Α. 15 made by using new manufacturing changes so that the pH can 16 be more represented between 3.4 and 3.6. 17 So in your opinion, would batches 1 through 6 that 18 were made with the old manufacturing process be 19 representative of the pH of Eagle's proposed ANDA products? 20 No. Batch, registration batches still representative 21 of some properties of the product, but not pH. So pH is represented by optimization batches. 22 23 Okay. So now below the manufacturing line change,

there's optimization batches 7 through 9. When were those

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manufactured?

A. They were made in July and August of 2019.

Q. And were those made with the optimized manufacturing process?

- A. Exactly. That is why it says optimized batches.
- Q. The PPQ batches, 11 through 13, when were they manufactured?
 - A. November and December 2020.

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- Q. Were those manufactured with the optimized process?
- 9 A. Yes. They are all prepared using manufacturing changes.
- 11 Q. And in your opinion, would batches 7 through 9 and 11
 12 through 13 be representative of the pH of Eagle's proposed
 13 ANDA product?
 - A. Oh, yes. As I mentioned before, pH of the final product is represented by the optimization batches or since then.
 - Q. All right. Now, on the right-hand side of the slide, there's some information about different parameters that are used for the stability testing. Could you explain what is listed here?
 - A. Yes. For many studies, many samples are collected, samples in vials. One is at room temperature. The other is refrigeration temperature. The others, the bridging study, the refrigeration temperature followed by room temperature.

In each storage condition, they are stored in

upright and inverted storage. Inverted storage is necessary because solution has contact with the rubber stopper. You want to make sure that rubber stopper does not release anything toxic. So this is to study nothing released from the rubber stopper.

- Q. Now, are you aware of other batches that Eagle made that were not placed on stability?
- A. Yes. They are placed on the slide, batch 10, 14, 15, 16 and 17 were made.
 - Q. Okay. And for the batches that were rejected or aborted, do you have an understanding as to whether they were rejected or aborted for reasons relating to pH?
 - A. No. Some of them was because the API was not pure, so they were rejected. The fact that they were rejected or aborted means Eagle is using good manufacturing process.

 There's a good quality control in place. So if the batch does not meet the criteria of each step of the process, they
 - Q. All right. So on slide 16 here, we have some information from DTX-993, which is the summary exhibit on page 5 of that. First of all, can you just look at DTX-993 in your binder?
 - A. Yes.

24 Q. And what is this document?

are not going to release the batch.

25 A. This is a table containing all the pH measurements,

all three different conditions, upright and inverted conditions.

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- Q. Okay. And did you examine -- there's a number of DTX numbers cited on the pages of this DTX-993. Do you have an understanding of what those are?
- A. Yes. Those are the reports on the pH and report, characterization report of each batch at each time point and there are several laboratory notebooks, so all of these numbers indicate that particular characterization, particular batches, particular time and laboratory notebooks.
- 12 Q. And did you review those documents and the data
 13 contained in DTX-73?
- A. Yes. I compared the actual pH data on the document and I confirmed that it's correct.
 - Q. Going back to slide 18, what data is being shown here?
- A. This is all pH measurement data of the room temperature data.
- Q. Okay. In your opinion, based on the room temperature data, would any of that establish infringement of the asserted claims?
- A. Well, as we all can see, all of the pH values are between 3.4 and 3.6 and none of them goes to the claimed range of pH 3.7 to 9.
- Q. Did you hear any testimony from Dr. Kirsch yesterday

that any of the room temperature data for Eagle's proposed product would establish infringement?

A. I don't believe so.

- Q. And looking at the optimized batches at the bottom

 here, the pH data, do you see any sort of trend with respect

 to this data?
 - A. Well, if you plot it on a graph paper, you may see it a little more clearly, but from this table, in the beginning you start with about pH 3.50 and you look going down to 12 months, about 3.37, 3.39. So it looks like a trend is literally going downward.
- Q. All right. So slide 17, there's more information from 993, pages 7 and 9. This is the bridging study data. Can you explain what a bridging study is?
 - A. Again, bridging study is initially vials are stored in a refrigeration temperature for a period of time and moving to room temperature. That's called bridging study.
 - Q. And in your opinion, would any of the data relating to the bridging study establish infringement of the asserted claims?
 - A. It was to see the data. None of the data shows outside 3.4 to 3.6.
 - Q. Did you hear any testimony from Dr. Kirsch yesterday regarding any of the bridging study data and that would establish infringement of the asserted claims?

A. I don't think so.

Q. And I think I forgot to ask this. On these slides and on some of the demonstratives that we looked at yesterday, the batches are indicated in red and blue. What does the red and blue represent?

- A. Again, I think all the tables show the red means batches were prepared before manufacturing change and blue highlighted batches are batches prepared after manufacturing change.
- Q. All right. So let's take a look at slide 18.

 It has data from 993, pages 1 and 13. What is being shown
 here?
 - A. This is pH measurements of the samples through the refrigeration temperature.
- Q. On the top, the top batch here is upright; is that right?
 - A. Yes. That's the one upright position.
 - Q. The top right-hand corner, there's the box that has the highlighted entries. What do you understand those to be?
 - A. All right. When they measured the pH of the samples, the initial pH measurement was 3.69, so that was out of specification, so at the time measured one more time the same sample and that was 3.75. Again, that was out of specification. They pulled another five vials and measured

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1 pH again. That was 3.68. That's why we have three values.

- Q. Focusing on the top, the batches in red, are there any other pH measurements that are out of the 3.4 to 3.6 specification for those batches?
- A. No. I think what I'd like to highlight here is that even without the manufacturing change, all the pH values remain between 3.4 and 3.6. Only one, batch 1, upright position at 24 months at the end of expiry shows one out of specification pH measurement.
 - Q. And now looking at the data in blue, the optimization and PPQ batches, what is your opinion with respect to that data?
 - A. I think the data clearly indicates one thing clear. First of all, before manufacturing changes, if you look in data at the red highlighted batches, the pH there is between 3.4 and 3.6 except the one at 24 months, but the values are around, literally fluctuate a lot, because as you can see from 3.44 to 3.64, et cetera, but if you look at the blue highlighted batches, all the data around 3.5, 3.51, 3.52. None of the data is actually as high as the one we observed in the batch 1.
 - Q. One quick thing. On the bottom, PPQ batch, the initial 3.52 value.
- 24 A. I'm sorry. Which one?
- 25 Q. For the initial value, for SVA11. Do you see that?

- A. Yes.
- 2 0. You understand there were six measurements that were
- 3 made for SVA11, six pH measurements that were taken for
- 4 SVA11?

- 5 A. Six pH.
- 6 Q. Can we get DDX-7-1 on the screen? I can't read that
- one. Can we get DDX-7-3 on the screen?
- 8 A. Oh, that one? Yes, I'm sorry.
- 9 Q. Okay. And so for the initial measurements for SVA11,
- 10 you understand there were six measurements that were taken
- 11 for the release testing and the initial pH?
- 12 A. Yes. First three measurements were done on
- 13 November 30th and another technician at AMRI measured the
- 14 same again. That's the second one.
- 16 | including. Do you have an understanding as to why that 3.52
- 17 is included here?
- 18 A. Because all six measurements were for the same sample,
- 19 so they averaged all six measurements. That is 3.52.
- 20 | That's the actual value.
- 21 Q. Is that the value that Eagle and AMRI have recorded
- 22 for the initial measurements for that batch?
- 23 A. Yes, that's correct.
- 24 \ Q. And how long does the data, how long has the stability
- 25 data been taken for SVA7 through 9?

1 A. Up to 18 months by now.

- Q. All right. How long do we have stability data for SVA11 through 13?
- 4 A. Up to six months.
- 5 Q. So based on all of the refrigerated data we have here,
- 6 what is your opinion as to whether or not some vials of
- 7 Eagle's ANDA products will have a pH in the infringing range
- 8 3.7 to 3.9?

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- 9 A. I can rely only on data. As you can see, data
- 10 | indicates all the values around 3.50. 3.52 or 3.51. So
- 11 | none of the data go out of their range, so I expect that pH
- will remain between 3.4 and 3.6.
- 13 Q. And on slide 19, the pH data, can you explain what's
- 14 being represented here?
- 15 A. This is a graphical representation of the pH measured
- 16 for refrigerated batches before and after manufacturing
- 17 change.
- 18 Q. This is the refrigerated data?
- 19 A. Yes.
- 20 \parallel Q. At the top, there's a plot of the registration and
- 21 characterization batch data, so that's SVA1 through 6.
- 22 A. That's right. SVA1 through 6.
- 23 \ \Q. And on the bottom, there's a plot of the data for the
- optimization and PPQ batches 7 through 9 and 11 through 13?
- 25 A. Yes.

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Q. Okay. And what is your opinion? Can you explain what this data shows?

A. Yes. Previously we saw the table. The table is sometimes difficult to see the trend, but the figures clearly indicate the difference between batches prepared before and after manufacturing changes.

Now, at top, the red highlighted one, before manufacturing changes. As you can see, again, except the one at the end of 24 months, all the values remain between 3.4 and 3.6. You can see data points are a little widely fluctuating. On the other hand, if you look at the pH at the bottom, all the pH values are closely around 3.50, and that is exactly the goal that Eagle had. What they reported to FDA are to make around 3.50. That's what the result shows, that Eagle meant.

- Q. All right. So let's talk about that optimized manufacturing process. First of all, here on the screen we have DTX-331, page 9. When that data point for SVA1 was obtained, what do you understand Eagle did, Eagle and AMRI did in response to that?
- A. Well, when Eagle and AMRI received the one data pH reading, out of specification, they had to investigate, they had to understand why that particular time point was out of specification. So they did, as you highlight there, out of specification investigation was promptly conducted.

Q. Okay. And what was the cause, the root cause determined for that?

- A. Again, that's Eagle reported to the FDA. They went through all the potential factors. It's not the API itself. It's not acetic acid. So they concluded the specification, conclude that the root cause of the out of specification was determined to be batch SVA001 was released at the upper limit of the pH specification, which is the release value of 3.64.
- Q. And did Eagle do anything in response to the determination?
- A. As soon as they found out the root cause, they did manufacturing changes to correct the problem, and that's why they further described that in order to provide greater confirmation of consistent product quality through the proposed expiry period, the manufacturing in-process controls were subsequently optimized to assure tighter control of pH. Here, trying to control pH during manufacturing.
- Q. All right. So let's talk about those changes.

What is the demonstrative here generally? What is being shown here on slide 22.

- A. This is a schematic presentation of Eagle's manufacturing process, so that everyone can understand.
- 25 Q. And so on the left-hand column, the one we can see

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that's not shaded out, what's that step? What's being represented here?

A. Well, the left step is the beginning of the compounding and compounding is a process where you mix ingredients to make a final formulation. That's called compounding.

The second is pH adjustment. The reason we call it adjustment, we start with the water. Start with the water and add Vasopressin and chlorobutanol. To make a pH 3.4 to 3.6, we had to adjust the pH. That's why we add acetic acid. Acetic acid is vinegar. Vinegar mixes with water very well. You just add acetic acid to 35-liter tank and stir to make it homogeneous pH around the tank.

- Q. On the right-hand side there's a 95 percent QS and a 100 percent QS. What does that mean?
- A. So this batch starts with 35 liters. 35 liter as counsel mentioned in opening statement about Gatorade with the cooler, it's about one foot diameter and about two feet high. So it's quite a big size, nine gallons. So 35 liters, you fill the water up to 95 percent, so 95 QS, quantity sufficient to fill 95 percent, and then add acetic acid and adjust the pH, pH adjustment to make a pH around 3.42 to 3.49.
- Q. All right. And then once -- let's start with the 95 percent QS. Once the QS has been adjusted to 3.42 to

1 3.49, you have arrows, you're going to stabilization? 2 This is Eagle's proposed manufacturing change to 3 ensure how to control the manufacturing, Eagle implemented 4 so-called new pH stabilization step. It's just basically 5 continuing mixing, so pH remains between 3.42 and 3.50. Is this all done again at the 100 percent QS? 6 7 Right. If the pH is met, then you add more water to Α. make 100 percent QS, meaning 35-liter, hundred percent 8 volume, and then measure the pH again, the pH adjustment, 9 10 stabilization step again. At the bottom of this slide there's reference to 11 12 DTX-324, the proposed commercial manufacturing batch 13 Is that where you obtained your information to 14 prepare this? The batch manufacturing record shows all the 15 Yes. 16 diagrams. I just put into some figure so we can easily 17 understand. All right. Let's talk in a little bit more detail 18 19 about the adjustment and stabilization steps. On the 20 left-hand side here, at zero time, that's the adjustment 21 step you were talking about where acetic acid is added? That's the beginning when you add acetic acid. 22 23 As I mentioned, acetic acid is vinegar which mixes with 24 water very well. Mix less than ten minutes to have

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homogeneous mixing.

- 1 Q. How is it being mixed?
- 2 A. Sorry?

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- 3 Q. How is it being mixed?
- A. Oh, every solution in this quantity, you have to have a propeller used to put from the top to stir.
 - Q. And then so it's mixing at ten minutes and then what happens at time 20 minutes?
 - A. So you, after adjusting pH, and at 20 minutes of mixing, you collect the sample from the bottom of the tank and measure the pH again, and then continue to mix for another ten minutes. After mixing, you collect the sample again from the bottom and measure the pH again.
 - Q. Why would you collect the sample from the bottom of the tank?
 - A. Acetic acid is on the top. The top has enough acetic acid where you can stir to make that acetic acid go down to the bottom and homogeneously mix.
- Q. And so when the measurement is taken at 20 minutes, what pH range does that need to be in?
- A. As shown at the top, should have pH reading between 3.42 and 3.50.
- Q. And then what pH range does the sample at 30 minutes maintain?
- A. So when we measure pH at 20-minute sample and 30-minute sample, between 3.42 to 3.50 and also two pH

It happened three or four times.

record.

Q. And did they go back and do the pH adjustment steps?

A. Right. That's what the whole process is. If pH is out of spec, you go all the way back to pH adjustment and start all over again.

Q. But what happens if you have a measurement at time 20 minutes that is 3.42 and then at time 30 minutes, it's 3.49. So they are both within the 3.42 to 3.49 specification but they are not within .03 of each other?

A. As long as they are meeting the specification, the two readings are not within .03. You just continue to stir. So in that case you go back to second step, not less than ten-minute stirring. So you continue to stir another ten minutes and measure again the samples and another ten-minute

Q. And what would be the scientific objective of doing this process, of stabilizing the formulation?

interval to see the difference in pH reading.

A. Complete homogeneous mixing. As I mentioned, acetic acid is water soluble, so you can mix very well, but still to ensure we can mix, continue mixing for hours.

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You have a 35-liter coffee and you add the You just completely mixing for hours. I can always quarantee the sugar dissolves and sugar dispersed throughout the 35-liter coffee container. That's what it's trying to do. So in your opinion, what is the impact on Eagle's proposed ANDA product of having a pH stabilization step, an adjustment step as set out here? Again, the goal was to have a narrower tighter pH That's exactly what they did. Again, acetic acid, reading. water soluble, that Eagle implemented let's mix more. Let's mix hours so that we have a full complete mixing of acetic acid throughout the whole tank. That's what Eagle has done. That's the result we have seen. So in your opinion, would Eagle's product be homogeneous? I cannot think otherwise. Have you looked at any, of the batches 7 through 9 and actually all of the batches that have been made with the optimized process to see approximately how long this process has taken for Eagle for those batches? I received the documents, batch 7 through 17, and some batches you had to go through a step twice or three times, sometimes adjustment step. So overall, average

time to finish adjustment, stabilization, it was about three

1 hours.

- 2 Q. Okay. So you have DTX numbers cited on the slide.
- 3 Are DTX-333, 334, 845, 867, 889, 913, 938 and 941, are those
- 4 the batch records?
- 5 A. Yes. Those are batch records I saw where they discuss
- 6 how much time it took for pH adjustment step. If the
- 7 specification was not met, they had to do it again and
- 8 establish those steps, too.
- 9 | Q. You mentioned in your opinion this would generate a
- 10 \parallel homogeneous solution. Does that mean that any pH
- 11 measurements that are take even for stabilization for these
- 12 products will be exactly the same?
- 13 A. No. It's a human thing. Human, also mechanical
- 14 thing. Even if you measure same sample, it's not going to
- 15 be exactly the same, but around the same value.
- 16 \ Q. All right. So now going back to the overall
- commercial manufacturing process, what is being shown here
- 18 in the middle section?
- 19 A. Now, if the pH adjustment and pH stabilization samples
- 20 | are done successfully, then you go next to the example for
- 21 pre-filtration. And filtration, here is the filtration to
- 22 point to the filter to remove all of the bacteria and just
- 23 make it clearer.
- 24 \ \Q. And what is the specification for the pre-filtration
- samples measured here on the bottom on slide 24?

A. 3.42 to 3.54.

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Q. And how is that sample taken?

milliliter as a reserve.

- A. Oh, again, the sample is taken from the bottom of the tank.
- Q. And then on the right-hand side, what is being shown here?
- A. So once you measure the pH before filtration and go through the filtration steps and then the solution is filled into individual vials.
- 10 Q. Approximately how many vials are filled from this tank
 11 of one batch?
- 12 A. 35 liter, they make around 26,000 vials, because even
 13 though it's a one ml sample, they actually add in 1.16 ml.
 14 So it is about 26,000 bottles and they reserve about 5,000
 - Q. And then below this, you have post filtration samples.

 Are those samples of the vials?
- A. Yes. They are all samples. At this point everything is filled into the vials and the vials are collected and all the samples are measured from the vials.
- Q. Okay. And so let's take a look at the next slide.

 That would be helpful. Where do you understand the post
 filtration samples to come from?
- A. Again, after filtration, you start filling the vials and in the process, first 90 vials are collected to

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measure post filtration pH measurements. And you can have a random sampling, 90 for samples, but if the stability study is necessary, then you can select random samples from the beginning, the middle and end of the capping process. And this is slide 25. And all of these vials, are they all filled at the same time? They are filled not exact same time, but around Α. Yes. the same time, same day, the same place. So you are not like leaving this tank full of stuff and taking a measurement from it six months later for stability, 18 months for stability, that's all done in the vials? You collect the samples of the vials. means vials and store it in the refrigerated temperature or room temperature and collect those five vials at each time point and measure, and once you measure, obviously, you cannot use it again, you discard. The next time point, you collect another five vials for measurement. You mentioned that you select five vials for measurement. Why would five vials be selected for measurement? Again, as I understand, the pH probe is actually The one I have been using in all of my research. large. So one ml vial is small, so opening, sometimes it cannot go in. It's routine to pull different vials with sufficient volume

so the pH probe. pH probe is basically glass. You have to submerse enough to have an accurate reading of a pH.

- Q. And then so what is done with those samples or those vials? Are they used for the next pH sample test?
- A. No. One measurement, that's the end. You use another new vial for next measurement.
- Q. All right. So let's go to slide 26. Can you generally describe what's being shown here?
 - A. This is just to compare the difference in manufacturing process between registration batches, batch 1 through 3, and the proposed commercial product, which is optimization batches.
- Q. And focusing on the first part, registration batches 1 through 3, what was the pH specification for the adjustment stage for the acetic acid added to the tank?
- 16 A. Right. The pH was 3.4 to 3.6.
- 17 \ Q. And what was the target?
- 18 A. The target is 3.5.

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- Q. And now for the proposed commercial products, what is the pH adjustment specification?
 - A. Now, here, as we discussed before, when Eagle was out of specification of pH reading at the end of 24 months, they conducted an investigation and the root cause was incomplete mixing. So they started making sure. So they adjust the pH to 3.4 to 3.49 and target 3.45, and then they conducted a

new pH stability step.

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- Okay. So was there a pH stabilization step in the registration batch manufacturing?
- That is the point of slide 26, because it was not Α. No. 5 there and that's why pH might have been fluctuating so much so that Eagle implemented a new manufacturing process that 6 7 includes pH stabilization data.
- And what is the pH specification for the stabilization 8 Q. 9 test?
 - Α. 3.42 to 3.50, but as you can see, it's much lower and narrower than the pH specification of the registration batch.
 - All right. On slide 28 we have excerpts from DTX-323 at pages 5 and 9 through 10. Is this where you obtained the specification information for the, that was used for the registration batches and that is in the proposed commercial batch?
 - This is the same information I used. Yes.
 - All right. And looking on the right-hand side here, Q. pages 9 through 10, is there any rationale or specification that Eagle provided to the FDA as to why they changed their pH specification and added the stabilization step?
 - Well, yes. We went through the slide already, but here again, Eagle informed the FDA, the exact reason is to assure tighter pH control throughout manufacturing process.

Q. Now, going down to the next in-process specification, the first one, in-process pre-filtration. Now, that was that middle column where the sample is taken after acetic

acid is no longer added and before the vials are filled; is

5 that right?

- 6 A. Right.
- 7 Q. Okay?
- 8 A. Pre-filtration, yes.
- 9 Q. So that was that middle column of the tank?
- 10 A. That's right.
- 11 Q. All right. And what was in-process pre-filtration
- 12 specification for the registration batches?
- 13 A. Specification was 2.5 to 4.5 with a target of 3.4 to
- 14 **3.6.**
- 2. What is the in-process pre-filtration proposed specification for the proposed commercial product?
- 17 A. Obviously, much narrower. 3.42 to 3.54.
- 18 Q. Below that, the in-process post filtration. Now,
- 19 that's testing of the vials; is that right?
- 20 A. Yes, that's right.
- Q. And what was the in-process post filtration for the registration batches?
- 23 | A. Again, 2.5 to 4.5 with a target of 3.4 to 3.6.
- Q. And what is the in-process post filtration
- 25 investigation for the proposed commercial product?

A. 3.42 to 3.54.

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Q. And on slide 30 you have again an excerpt from DTX-323, pages 12 to 13. This is where you obtained the specification information from the previous slide with respect to the registration batches and the intended commercial product batches?

A. That's correct.

- Q. Now, did you look at any of the in-process pH measurements for SVA1?
- 10 | A. Yes.
 - Q. Here, on the top of the slide is that same excerpt from 323 that we just had and on the bottom is DTX-134, which is the certificate of analysis for SVA1 at page 1.

What was the pre-filtration measurement for SVA1.

- A. As you can see, 3.7, that meets the specification for batch 1 at the time.
- 18 Q. And what was the post-filtration pH measurement?
- 19 A. 3.6.
 - Q. So would both of those pH measurements have met the specification in place at the time SVA1 was manufactured?
- 22 A. Right, at the time.
- Q. And let's start with the first one. Would the
 pre-filtration measurement of 3.7 have met the specification
 of the intended commercial production batches?

A. Clearly not, because the intent is 3.42 to 3.54, so this would not meet the qualification.

- Q. Okay. And would the post filtration measurement of 3.6 have met the intended commercial production batch pH
- 6 A. Again, not.

specification?

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- Q. So in your view, is SVA1 representative of the pH of the proposed commercial product?
- A. No. Again, I mentioned a few times before, we went through, batch 1 represents certain properties of the final product but not pH. pH is represented by batches 7 to 9, optimization batches, after manufacturing using new manufacturing process.
 - Q. All right. Let's turn to the last section, release and stability. At the time SVA1 through 3 were manufactured, what were the release and stability pH specifications?
- A. Well, this is the one we have gone through, 2.5 to 4.5.
- 20 Q. And for the proposed commercial products, what are the release and stability products?
- 22 | A. **3.4 to 3.6.**
- 23 Q. And here on slide 33 at the top, do we have an excerpt
 24 from DTX-2 at page 1? Is this where you obtained the
 25 release and stability specification for SVA1?

1 A. Yes. As you can see, the lot number, SVA001.

Q. And that was manufactured in March of 2017?

A. Yes.

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- 4 Q. And your understanding, that was before the patents in
- 5 this case issued?
- 6 | A. Yes.
- \bigcirc Q. And on the bottom, there's an excerpt from DTX-327 at
- 8 page 1. This is where you obtained the release and
- 9 stability information for Eagle's products?
- 10 | A. Yes.
- 11 \ \Q. And we're also citing to DTX-78 and DTX-1427 at page
- 12 | 1, the chart, which are the same position and updated ANDA?
- 13 A. Yes. I mentioned they are updating the information,
- 14 but this information remained the same.
- 15 \ Q. Have you seen any information in the batch records
- 16 that indicate that the post-filtration and release testing
- samples were taken at the same time and is the same place?
- 18 A. Yes. As I show on the slide, you can see that batches
- 19 7, 8 and 9, they are collected from the same testing area
- 20 | capping area, which means you fill the vials and cap,
- 21 | capping area. You can see the same day, on the same day
- 22 around the same time and the same place.
- 23 | Q. I didn't mean to interrupt you. This is slide 34 and
- 24 on the top, DDX-2-34, that's the proposed batch records
- 25 we've been discussing?

1 A. Yes.

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Q. SVA7, 8 and 9 were the information maintained for those batch records?

A. Yes.

- Q. Why in your opinion does it matter that these vials, these samples are selected at the same time from the same place?
- A. Well, because post-filtration and release samples and release samples, they are collected at the same time, so basically, we label them post filtration and samples, but samples are collected the same day, same place, same time.

 So they actually represent the whole batch.
- 13 Q. All right. So did you hear Dr. Kirsch yesterday opine 14 that in his opinion, some vials may be released at 3.64?
- 15 A. I heard that.
- 16 Q. And do you agree with that?
 - A. Again, based on the manufacturing process we have gone through and all the data we have here for batch 7 and on, no data indicates that the pH will go to 3.64.
 - Q. So based on Eagle's optimized manufacturing process that we've just gone through, would you expect to see large variability between different vials of the batch of Eagle's product?
- A. No. Again, as we have seen, all the data, pH
 measurements for the new manufacturing process batches, the

pH values are all around 3.50, 3.51, 3.52, so I don't expect they will be above 3.64.

Q. Can I get DDX-7-3.

Now, this is -- this is showing all of the post-optimization data?

A. That's right.

- Q. In your opinion, does this data demonstrate very wide variability within Eagle's batches?
- A. No. I think -- let me be a little more specific here.

 Dr. Kirsch and Par was talking about batch 11, and they had

 two data points, 3.47 and 3.57. So there's a .1

 variability. It's true, it's there, but it is good to

 remember that all this variation occurs around 3.52.

If you have average of 3.57 and 3.52, 3.57 and 3.47, all six measurements, 3.52. So this variability is indicating it's around 3.52. So variation occurs, yes, but it's around 3.52. It's not going to go to 3.64.

- Q. And did you hear Dr. Kirsch opine that you would take that .1 variability and then you could apply that to the top of any of the pH values that you obtain here? Do you agree with that?
- A. No. As I mentioned, pH 3.47, pH 3.5, already variable, but it's there. You cannot input the variability and assume they occur again on top of it. What that means is that actual variability is .2, so up and down, above

1 3.52.

So he can -- he cannot apply .1 variation already occur on top of the top value again. That data just does not show that.

Let me, let me talk about batch 7. If you look at batch 7, data point of 6 months, 3.55. 18 months, 3.46. Variation is 0.09. Variation occurs around 3.50. You cannot say that another .1 variation on top of 3.5. That is pure speculation. Data just does not support that.

- Q. If there were vials of product that were in the 3.7 to 3.9 range, would you expect to see higher pH values in any of the stability testing that has been done?
- A. No. Again, can you go back to showing all the values from batch 1?
- \mathbb{Q} . Can we get DDX-7-2?
 - A. Right. If that's the case, we should see the pH readings like in batch 1 or before manufacturing changes.

 You can see the values are between 3.4 and 3.6, but the value for 3.4 to 3.7. Right? That is the kind of variation we will see. But after manufacturing changes, the data just does not show such a big change.
 - Q. So you're saying after Eagle has made the optimized manufacturing changes, you are not seeing that same type of variability?
- 25 A. Right. Here, before manufacturing, the variation up

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to 03.1, but after manufacturing, change the maximum variation of 03.1, which occurs around 3.5. Can we go back to the slide 36. Here on slide 36, we have another section of 331 at page 26. And what were the pH results again for the optimization/confirmation batches 7, 8 and 9? Well, this is the data we have seen and Eagle, Α. according to the FDA, exactly what we have discussed. There were release pH results for the confirmation batches were 3.50 for seven. 3.52 for batch 8 and 3.48 for batch 9. batches met the target midpoint of the pH specification as designed. We have seen the data, all the data we have gone through varies around 3.50. All right. So now based on all the data you've seen in the optimized process, in your opinion, would Eagle's product fall into the claimed range? Again, based on the manufacturing changes, based on the data, we can only rely on data. Based on the data, I'm confident that Eagle's proposed product would not have pH outside of 3.4 to 3.6. All right. And we heard some testimony yesterday from Dr. Kirsch that Eagle will be unable to maintain a pH

Do you agree with that?

specification, that the optimization process isn't working.

1 Α. I heard that. 2 Do you agree with that? Q. 3 No. I don't agree. Α. And here on slide 38, we have DTX-800 at page 2. What 4 0. 5 is being shown here? Well, this is the one that Dr. Kirsch has been talking 6 7 about and also we went through this chart. 8 Now, the left column, blue highlight, is the one 9 that was measured on November 30th. Now, pH beginning, 10 middle and ends of the sample collection was 3.54, 3.56 and 3.57, average 3.56. 11 12 First of all, these values are very close to 13 each other, so measurement is precise. Two days later, the 14 measurements were 3.51, 3.49, 3.47. Again, values are close 15 enough, very precise. 16 So because of the same samples, they pulled the 17 six values and averaged 3.52. That was reported. 18 So Dr. Kirsch is saying 3.47 and 3.57, there's 19 .1 variation. Yes, he did. That's the data. But the 20 problem here is that you cannot add a .1 value already 21 happened and to the top of the maximum value of 3.54. That data just does not support it. 22 23 I just want to make sure I understand. So the data is highlighted in blue here, 3.54, 3.56 and 3.57. Was that all 24

taken from the same set of vials?

A. Right. As we discussed, when POSAs speak of stability study, you collect the random samples on the same day, same place, so they are the same pool of samples. You select five random vials.

Q. Sorry. I need to clarify my question a little bit more. The beginning measurement, middle measurements and -- the beginning measurement, is that taken with five vials?

- A. Yes. Sorry.
- Q. The middle measurement, is that five different vials?
- 10 | A. Yes.

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- 11 Q. All right.
- A. I think, yes, to me as a scientist, it's so obvious,
 but let me clarify. Beginning, you collect the sample from
 the beginning of the process. You collect the five samples
 and five samples collect from the middle of the filling
 process and select the five samples from the end, so this is
 a result of pulling five vials.
 - Q. Okay. And then those measurements in blue were taken on November 30th, you mentioned?
 - A. Yes.
- Q. All right. And the measurements on the right that are highlighted in yellow, 3.51, 3.49 and 2.47 were taken a few days later?
- 24 A. Two days later, yes.
- 25 Q. By a different lab technician?

A. Yes.

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- Q. Were there any other PPQ batches similar to 11 that were done at, that were manufactured by Eagle?
- 4 A. Yes. As we move on, there are more batches. Batch
- 5 | 11, 12, 13, 14, 15, 16 and 17.
- Q. Those are all shown here on slide 39, which we have excerpts from DDX-2-39?
- 8 A. Yes.
- Q. I just want to ask you a couple questions based on
 Dr. Kirsch's testimony yesterday. Dr. Kirsch was asked
 whether in his opinion batches 1 and 13 are the same. Do
 you believe batches 1 and 13 are the same?
- 13 A. I'm not sure how batch 1 is the same as batch 13.
- 14 Q. And how do you think they're different?
 - A. Batch 1 was prepared before manufacturing changes and batch 13 was made after manufacturing changes. Once again, as I mentioned a few times, batch 1 still represents certain properties of the final product, but not pH. Batch 13 was after manufacturing process changes, so batch 1 is not the same as batch 13.
 - Q. All right. Let's turn to your final opinion on inducement and what do you understand -- can you explain why you believe Eagle does not directly infringe the '209 and '785 patents?
- 25 A. I was about to do -- Eagle will have pH around 3.4 to

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3.6, will not -- I don't expect it will go to 3.7 to 3.9, so it will not infringe the '785 patent.

It will not infringe the '209 patent, because Eagle will not itself performed the claimed method of the treatment, treating patients.

- Q. All right. And on slide 42, we have an excerpt from DTX-133 at page 21. What is this document?
- A. Again, this is ensuring to the FDA that Eagle has implemented the new manufacturing process to ensure the pH remains within the established range during finished product manufacturing and through the proposed shelf life.
- Q. And in your opinion, has this goal, has this worked?
- A. Yes. I can only rely on data. As you can see, data shows that pH remains within 3.4 and 3.6.
 - Q. All right. On slide 43 we have a call-up from DTX-341 from page 34 from Eagle's ANDA. And what are they stating here to the FDA about the optimized process?
 - A. Again, the top highlighted portion indicates that what, the target was meeting pH to middle of the specification, which is 3.50, and bottom confirmed that these steps were implemented to provide greater assurance all future Vasopressin injection batches will remain within the proposed stability specifications through the end of shelf life, 24 months, for all labeled storage conditions.
 - Q. And taking a look again at the plot of the

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post-optimization data, what does this demonstrate to you with respect to whether or not Eagle's products will remain within specification? Now, the top shows that all the pH values are within 3.4 and 3.6. In particular, around 3.50 up to 18 months. Bottom shows that the pH values stored at room temperature up to 12 months and you can see in the beginning, pH is around 3.50. You can see a trend, the pH going down below 3.50, but above 3.40. So the data indicates to me that it is very likely that Eagle's product will maintain its pH between 3.4 to 3.6 throughout its shelf life. So then in your opinion, would Eagle's proposed commercial products infringe the asserted claims? It doesn't go to the claimed pH range of 3.7 to Α. No. 3.9, so I don't think it will infringe the claims. MS. WACKER: Your Honor, that's the end of noninfringement. I have some questions for Dr. Park on I'm going to move on to invalidity. invalidity. I don't know if you want to pause here. THE COURT: Did you figure out how you want to Do you want to do it in cross? do cross? I think we need to do it MR. BLACK: No. altogether. Maybe this would be a good time for a break. THE COURT: You said very likely that it will

1	remain, the range will remain between 3.4 and 3.6. That was
2	your testimony?
3	THE WITNESS: Yes.
4	THE COURT: What's very likely?
5	THE WITNESS: I cannot put any number, Your
6	Honor, but data would indicate to me all the data, pH
7	measurement between 3.4 and 3.6, around 3.5. So it did not
8	happen. I think it's not likely to happen again.
9	THE COURT: So it is very likely that it will
10	remain. Better than a 50 percent chance it's going to
11	remain between 3.4 to 3.6?
12	THE WITNESS: Personally, I think it's more than
13	that. Fifty percent is yes or no, either way.
14	THE COURT: Yes.
15	THE WITNESS: So very likely it didn't happen.
16	For example
17	THE COURT: You think it's greater than
18	50 percent?
19	THE WITNESS: Much greater.
20	THE COURT: Much greater. Okay. All right.
21	Do you want to take a break? Would you like a
22	break?
23	THE WITNESS: I'm fine, Your Honor, if you are
24	okay.
25	THE COURT: I'm okay.

1 MR. BLACK: I'd like to have a short break. 2 THE COURT: Mr. Black would like to have a 3 We're going to have a break. Let's try to make it break. 4 short, you know. Let's try to be back at 10:00. You may 5 step down. 6 THE WITNESS: Thank you. 7 (Short recess taken.) 8 9 (Proceedings resumed after the short recess.) 10 THE COURT: All right. Please be seated. 11 right. 12 MS. WACKER: May I get DDX-7-0? I forgot one 13 question on infringement. I apologize. 14 BY MS. WACKER: You heard Dr. Kirsch testify yesterday that the PhD of 15 SVA1 could have risen from anywhere between 18 months and 16 17 24 months. 18 Is there any 21-month refrigeration data 19 available for that batch? 20 Α. No. 21 On the stability study, on the bridging study, is there 21 month data eligible for SVA1? 22 23 Bridging study, yes. 21 months. Α. 24 And what is the ph measurement there? Q. 25 3.47 and 3.45, 3.49, 3.45, 3.44. Α.

Q. What is the 21-month stability measurement of the refrigerated bridging study for SVA1?

A. 3.6.

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- Q. All right. Going back to the slide, did you consider the claims from the perspective of a person of ordinary skill in the art?
- A. Yes, I did.
- 8 Q. And what definition did you apply?
 - A. A person of ordinary skill in the art is someone who has a master's degrees or a pH degree in pharmaceutical sciences or related skill with several years of experience in developing pharmaceutical dosage forms, including stable aqueous peptide formulations and more experience may substitute for lower level of education and vice-versa.

Also, a person can have access to and collaboration with persons having drug formulation experience as well as pharmacologists, chemists, biologists or clinicians basically can work as a team.

- Q. And did you consider the definition of a POSA that Dr. Kirsch applied?
- 21 A. Yes.
- Q. Would applying his definition change any of your opinions in this case?
- A. Practically, it was not that different, so I don't expect so.

Q. And what priority date did you apply to the patents-in-suit?

- A. February 7, 2017.
- 4 | Q. These dates are not disputed; is that right?
- 5 A. No.

- Q. And what asserted claims did you consider for Eagle and Amneal for purposes of invalidity?
- 8 A. For the '209 patent, 1, 4, 5, 7, and against Amneal,
- 9 1, 2, 4, 5, 6, 7, 8. For the '785 patent against Eagle,
- 10 both Eagle and Amneal, claims 1, 5 and 8.
- 12 Q. And are you relying on any other expert with respect to any of the limitations in the asserted patents?
- 13 A. Yes. For the method of treatment, I rely on Dr.
- 14 Cross.
- 15 Q. Now, both of these, all of the claims have a

 16 requirement of a pH of 3.7 to 3.9. How would you understand

 17 Dr. Kirsch and Par to be reading when that pH limitation is
- 18 | met?
- A. Dr. Kirsch and Par said that if claimed pH will occur any time during shelf life, even for a few minutes, and that
- 21 will infringe the claimed pH.
- 22 Q. So for a product under that definition, would a
- product that has a pH of 3.63 for its entire shelf life that
- rises to 3.65 for five minutes, would that infringe this
- 25 claim under their reading?

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A. Yes. That's what Dr. Kirsch and Par interpreted and that's why they rely on batch 1, which has a pH out-of-specification at the end of 24 months.

Q. And now have you applied that broad construction of this limitation in, that broad interpretation in an analysis -- sorry. Let me start over.

Have you applied that broad interpretation in analyzing the prior art in this case?

- A. Yes. In my analysis, although I don't agree with that interpretation, I applied their interpretation.
- Q. All right. And what limitations are added by the dependent claims in this case?
 - A. Dependent claims are about the impurity levels and each dependent claim is related to a specific impurity.
 - Q. So let's talk a little bit about Vasopressin and its history. On slide 51 here on the left-hand side we have an excerpt from DTX-38 at page 5. What do you understand that document is?
 - A. This is a document that JHP submitted to FDA in 2010 as a pre-NDA letter.
- 21 Q. And JHP, how are they affiliated with Par?
- 22 A. JHP was purchased by Par as I recall in 2014.
- Q. Okay. And what did Par and JHP tell the FDA in this submission?
- 25 A. They indicated numerous times that they are making

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Pitressin formulations, which was -- I can just read the letter. It's probably clearer. Pitressin currently marketed by JHP is the same product as the Pitressin drug product previously marketed by Parke-Davis since pre-1938. Okay. And on the right-hand side, this is a section from DTX-25 at pages 9 through 10. What do you understand that document to be? This is another letter to the FDA I think submitted in 2011, and to describe the same information, especially it is a pre-1938 drug as the grandfathered product manufactured and marketed by various companies without the approval of FDA. Does it indicate some of the companies that were marketing Vasopressin products? Yes, because it was sold without approval, that NDA, Α. numerous companies sold, and this letter, they describe it as two, American Regent and also APP as an example. Now, given that there were these unapproved products on the market at the time, are you aware of whether there were any standards in place for Vasopressin injection products? Α. Yes. What standards were in place? 0. Each drug product sold on the market, USP has a certain monograph, so it describes information, dosage form,

identity, purity, strength, all of this information about
specific formulations in the monogram.

Q. And what is USP?

A. USP is the United States Pharmacopeia that collect all

of the information of each product, so USP itself does not have legal authority to enforce what is described here, but FDA relies on USP. So all the formulation scientists, companies follow USP guidelines.

- Q. All right. So here on slide 52, we have an excerpt from DTX-135 at page 3 to 4. This is the USP from 2009?
- 11 A. Yes.

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- Q. What does it provide with respect to the pH for Vasopressin injection products?
- 14 A. Here it indicates that pH between 2.5 and 4.5.
- 15 \ \Q. So where does the USP get this information from?
 - A. Usually USP get this information to make a monograph from the company, usually. It may not be always. Usually, they get the information from the largest manufacturer of their particular product at the time.
 - \mathbb{Q} . So what would a POSA understand by the -- about the Vasopressin injection products by the USP providing 2.5 and 4.5?
 - A. Well, USP simply means that it indicates that if pH is maintained between 2.5 and 4.5, the product should be good for the shelf life.

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THE COURT: Ms. Wacker, can you repeat your question and can you slow down a little bit? I think both the court reporter and I are having a hard time keeping up. MS. WACKER: I'm sorry. THE COURT: That's all right. I know we have an answer, but I think just so the record is clear, it might be good. Would a POSA understand what? MS. WACKER: I will rephrase. THE COURT: You don't have to rephrase. might want to actually ideally repeat what you said, but slower, so we can make sure we got it. BY MS. WACKER: What would a POSA understand by the USP providing the pH range of 2.5 to 4.5 for a Vasopressin injection product? Whenever a POSA makes a formulation, they rely on Α. information like a USP monograph, so it describes information. Monograph describes identity, purity, purity level, how to measure pH, what is pH range, all the things I mentioned. So a USP monograph indicates pH between 2.5 and 4.5, then a POSA would understand if pH is maintained between 2.5 and 4.5, the formation must be good enough during the shelf life. Now, was there any information available regarding the

Pitressin product that was on the market that was unapproved

1 by the FDA?

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- A. Information, yes.
- Q. And here on the left-hand side, PTX-178 at page 2, what does the Pitressin label discuss about Vasopressin
- 5 injection products?
- A. Well, this is a Pitressin formulation label. It
 indicates, as you can see on the right side, the left side,
 it shows that how pH, pH -- how the Pitressin, that it's
 standardized to contain 20 USP units per ml. And the
 solution contains .5 percent chlorobutanol as a preservative
 - Q. And is that the same, are those the same ingredients that are in original Vasostrict and Eagle's product?

and the acidity of the solution is adjust with acetic acid.

- A. Yes. Exactly same.
- Q. And on the right-hand side, we have an excerpt from PTX-145 at page 302. What information does this provide with respect to the pH of the earlier Pitressin product?
- A. Well, you can see the third from the top, acetic acid,

 pH adjustment, QS to pH 3.4 to 3.6 and currently marketed

 unapproved formulation, QS to pH 3.6.
 - Q. What does QS to 3.6 mean?
- A. As we have gone through QS, it means quantity
 sufficient. In this case, add a sufficient quantity to make
 a pH between 3.4 to 3.6.
 - Q. That's the adjustment step where the pH is added to

1 | the tank of liquid?

A. Yes.

- Q. All right. Let's talk about your opinion with respect to original Vasostrict. What are you considering as the original Vasostrict prior art?
 - A. Oh, original Vasostrict is the product that was sold with a label and Eagle is making a copy of that particular product as a reference drug.
 - Q. And as of the priority date, what would a POSA in the art have looked to when determining what formulation of the Vasopressin injection product to make?
 - A. There's only one FDA approved product. That's original Vasostrict. So Eagle is making a copy of the approved drug, so naturally, you are looking to original Vasostrict.
 - Q. Were there any differences between the Pitressin product that was on the market prior to original Vasostrict product?
 - A. They are all the same except one minor difference. I think that JHP applied to the mixture Pitressin formulation, they're going to have an NDA approval, but then FDA, as I recall, said that why don't you remove the overage and have the same formulation. So Pitressin formulation then, original Vasostrict, almost the same formulation.
- Q. And here on slide 55 we have an excerpt from DTX-26 at

1 page 3. What is this document?

- A. It's a document subject to the -- document to the FDA.
- 3 Q. What is Par telling FDA about the Pitressin product?
- 4 A. Again, Par, as we discussed before, they keep citing
- 5 the fact that the Pitressin formulation was sold before,
- 6 almost hundred years, indicating they are safe and
- 7 effective. And the bottom they show that three, there is
- 8 no -- sorry. So basically, they are saying that Pitressin
- 9 product will be the original Vasostrict.
- 11 A. Yes. As I mentioned, FDA recommends to remove the
- 12 overage, so that's the only difference.
- 13 \ Q. And what are overages?
- 14 A. Before NDA application, the Pitressin formulations had
- 15 | seven percent additional Vasopressin, so that is called
- 16 overage.

- 17 \ Q. And you also, at the bottom of this section, it
- 18 discusses NDA registration batches. Are those the batches
- 19 that Par relied on to support its NDA for the original
- 20 Vasostrict product?
- 21 $\|$ A. Yes. Again, by definition, registration batches are
- 22 one they rely on for the product.
- 23 \ Q. And do you understand when original Vasostrict was
- 24 | first sold?
- 25 A. According to sales records, it will be November 2014.

Q. And these sales records are in DTX-86, page 1 here on slide 56?

A. Yes.

Q. Is original Vasostrict still sold today?

A. No. Par withdrew it from the market and they started selling so-called reformulated Vasostrict.

Q. Let's take a look at the claims of the patents and on the left-hand side here, we have DTX-132. This is the -- actually, can you tell us what DTX-132 is?

A. This is a March 2015 Vasostrict label.

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- 11 Q. All right. Is this the label that was sold with the 12 product when original Vasostrict was being sold?
- 13 A. Yes. The label, it's coming with the product.
- Q. Okay. And I believe you stated you're relying on Dr.
- 15 Cross with respect to these limitations of the claim?
- A. Yes. Although as I can read it, that the indication is to increase blood pressure, but I still rely on Dr.

 Cross, because he's a medical doctor.
 - Q. And how does the formulation of the original Vasostrict product as described in the March 2015 Vasostrict label, DTX-132.5, compare to the formulation requirements in the '209 patent?
 - A. I think the label made it clear that one ml solution contains Vasopressin, 20 units per ml and pH was 3.4 to 3.6.

 Now, 20 units per ml can be converted to milligram per ml

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concentration, because at bottom of the label it said one milligram is equivalent to 530 units.

So 20 units per ml is equal to .0377-milligram per ml, which is within the range of .01 to .07 milligrams per ml in the patent.

- Q. Now, in looking at the impurity limitation from the claims, how did you assess those limitations?
- A. Now, limitation is the sequence homology of SEQ ID number one, which we heard yesterday is Vasopressin itself.

 So total level in the claims said that impurities, it combined it.
 - Q. And what about for the dependent claim limitations?

 How do you assess those?
 - A. Right. Each dependent claim is identified by SEQ ID number and also it comes with another name, Gly9, et cetera, that Dr. Kirsch explained yesterday. So those individual claims indicated and that's why we can calculate it.
 - Q. Going back to the independent claim, what does sequence homology mean?
 - A. Again, SEQ ID Number 1 is Vasopressin. As Vasopressin undergoes degradation or chemical changes, some portion may change, but the whole structure remains similar. That's sequence homology.
 - Q. Now, in calculating the impurities of the prior art, did you include unidentified impurities in the total

1 value? 2 Unidentified is not identified in the individual No. 3 claims, so we do not count it. 4 MS. WACKER: I believe we're going to have a 5 dispute. MR. BLACK: Your Honor, with respect to the 6 7 original Vasostrict lot and the slide to follow, there's 8 almost no disclosure in his reports, which I think we have 9 seven. 10 This lot was manufactured after reformulated Vasostrict had been on the market. The data relating to the 11 12 product was provided during discovery. They have this 13 document during the discovery period. Amneal, there was no 14 question about whether it was sold. It was manufactured after the product had already been on the market. 15 16 THE COURT: There's no question that it was 17 sold? 18 MR. BLACK: Yes, it was sold. 19 THE COURT: Yes. 20 MR. BLACK: They were aware that it was sold 21 because the dates of the manufacture are on the slide. In the Amneal case, Amneal asked for sales data from 2015 22 23 forward and we gave it to them, and then when the cases were consolidated, Eagle said, well, we want to rely on that now 24

that we know for sure it was sold, and rather than bring the

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matter to the Court, we said, fine, it's going to be in the case anyway.

But he then gave a report with his opinion, as is required, and the report mentioned this lot number, but there's no data relating to the lot number in the report.

None of these things which are highlighted here, calculation of the homologous impurities, any of this stuff is in the report and we were shocked and surprised to receive a, to see it in opening and in his expert testimony here today.

There's a second lot, the same situation, 788436 that they had knowledge about, was on the same list, and it was sold after the -- after reformulated had been launched, and we said if you want to adopt Dr. -- he had no opinions on it.

And we said, if you want to adopt -- he had no opinions on it. We said if you want to adopt Dr. Winter's opinion to move the trial along, we'll let you do that, and he gave us a report, which I thought was a very reasonable compromise rather than bringing the matter to the Court.

And the bottom line is that we're being repaid for our largess here with clearly out-of-scope testimony, which they refused to back off on this lot 788435.

It is not in his report and we are surprised by this. They've had this for a long time. And this is really

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a situation where we're being sandbagged, and I would ask that they show where in the report the information that is on the next slide. THE COURT: Okav. MS. WACKER: Okay. Can I get paragraph 8 of Dr. Park's opinion report. THE COURT: Actually, do I have the report? MS. WACKER: I think so. There are quite a few in the binder. You don't, Your Honor. I apologize. THE COURT: That's all right. You don't have to apologize. And I only need the one -- well, I will just take what you give me. So where are we? All right. MS. WACKER: Paragraph 8. THE COURT: Which tab? MS. WACKER: It is the February 2nd, 2021, report. THE COURT: On, February 2nd, 2021? MS. WACKER: Correct. THE COURT: Okay. I'm looking at it. MS. WACKER: And so just for some context, in Dr. Park's opening report, he provided an opinion that batches, original Vasostrict that were the representative batches demonstrated the impurity profile, and we did not have the sales information for this batch for the 788435,

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the one that was on the slide. We didn't have the fact that it was sold.

We learned that there were additional sales records that were not produced to us in our case when we were consolidated with Amneal. We learned about this as we were getting ready for trial. It was a few days before we were supposed to start trial in January.

THE COURT: Had you asked for the sales records?

MS. WACKER: We did. We had some of them but

not all of them.

MR. BLACK: They had not asked for sales records for this time period because this is after the product had been launched and therefore everybody understood that if they launched. They missed this issue.

THE COURT: You've got to --

MS. WACKER: We asked --

THE COURT: Hold up. Sorry. The significance -- you said they had not asked for sales records because this is after the product had been launched.

MR. BLACK: They had all of the technical information, and the document that's on the screen during discovery, that has information about the stability data for that lot. So they had all of the information, all the pH information and everything they needed to run the case, but they didn't rely on it.

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They claim that they didn't rely on it because they didn't know it had been sold, but all of the data on that document is after the launch date and they never asked us. They didn't ask us for sales data specifically from 2015 forward because they knew it. But whether they knew it or not, they never gave us a report.

THE COURT: We'll get to that.

MS. WACKER: We did ask for all the sales records. I thought we resolved the issue. Dr. Kirsch's opinion, he's taking the position you have to establish impurities in pH are met at the time of sale and use, so at the time.

So when something is sold is important, and so we didn't know when these batches were sold or what was going on with them. We didn't even know if they had been sold because some batches are tested for stability.

So we got that information. We agreed, the parties agreed we could put it in a report, and he specifically in this paragraph 8 talks about the lot 788435 and says, I understand that these lots were on sale and in use before the patents-in-suit, and therefore are prior art reflective of the commercial original Vasostrict product and supported his earlier opinion.

And Dr. Kirsch after this submitted a response to the opinion analyzing the data for these batches. So

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there's no prejudice here. They've actually responded to it. MR. BLACK: There is severe prejudice because that --THE COURT: Just hold on a second. So let me just step back. You were just discussing homogeneity? MS. WACKER: Yes. We want to talk about how the impurity limitation in this batch was compared to the claims. And --THE COURT: Hold on. So you want to talk about the impurity limitation. So the stability data, and if it's MS. WACKER: helpful to go back to the slide, what's provided in the document, DTX-360, is the actual, like, stability information and pH information, and so we're showing --THE COURT: Right. I get the pH. I just want to make sure I understand it, because as I read the paragraph in the report, paragraph 8, and I think to read it in context, I have to read paragraph 7. Maybe not, but I thought I did and have to then refer to paragraph 53, which was the original reply report. It's all in the same context? MS. WACKER: In his opening report, he talks

about batches of Vasostrict. Although the impurity

1	limitations are not contested for infringement, we have to
2	establish it for invalidity.
3	So in his opening report he walked through how
4	the prior art would have established the impurity
5	limitations and what those levels are and how he calculated
6	them and not including any identified impurities, et cetera.
7	And so what he has done here is used that same
8	analysis with respect to a batch
9	THE COURT: Should we excuse the witness?
10	MR. BLACK: Yes, Your Honor.
11	THE COURT: Sir, can you please step outside of
12	the courtroom? I apologize.
13	(Witness excused.)
14	MS. WACKER: I can get
15	THE COURT: Hold on. He has not left yet.
16	MS. WACKER: Sorry.
17	THE COURT: Okay.
18	MS. WACKER: And then Dr. Kirsch responded
19	THE COURT: Again, let me just step back.
20	MS. WACKER: Okay.
21	THE COURT: You all know this very, very well.
22	I don't. I can't even read the chart.
23	So we've got impurity limitations and we've got
24	pH limitations?
25	MS. WACKER: Correct.

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THE COURT: All right. In paragraph 8 -- so if you read paragraph 8 in isolation, it references the data. MS. WACKER: Stability data includes pH Stability and impurity. information. THE COURT: I've got to go slow. I'm sorry. References the data, and he said that the sales records confirm that a number of lots of original Vasostrict for which I had cited stability testing in my reply report were, in fact, sold before the priority date. And then he has this very broad sentence, I understand that these lots were on sale and in use before the patent-in-suit, and therefore are prior art, reflective of the commercial original Vasostrict product. Now, I mean that's a very broad general statement. It doesn't say in what respect they're reflective. Now, in the paragraph before it he's talking about looking at the original Vasostrict product and he -essentially, I think what he's focusing on is that the data about the original Vasostrict, which would undermine the argument that the pH range was critical. Right? And so --MS. WACKER: I think that was the conduct. THE COURT: That's what I'm getting at. You

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think paragraph 8 is a standalone paragraph. I don't read it -- in other words, I'm not reading it as it's just further qualifying or elucidating what's in paragraph 7? MS. WACKER: I think that's right. He was using this to describe, when I talked about --THE COURT: My problem then is, okay. So if I read paragraph 8 standalone, it doesn't really tell me anything. It just basically says, hey, we've got the additional sales value, very, very broad, and basically, I consider it to be prior art. That's all it says. doesn't say anything about how it's prior art. There are only two aspects of the MS. WACKER: prior art of stability that are at issue in the case, impurities and pH. And I think the important aspect here is that Dr. Kirsch understood what he meant because Dr. Kirsch had a responsive report whereas he looked at just a handful of batches that were identified by Dr. Park, that we received the late sales information for and looked at the data himself and responded, so he understood what the opinion was. THE COURT: Show me where Dr. Kirsch said that. MS. WACKER: This is Dr. Kirsch's March 23rd, 2021 report, paragraph 7. Do we have copies of that? THE COURT: I've got Kirsch's report. It's the March 23rd, 2021. And so

MS. WACKER:

1 here he's talking about the specific lots, talking about the 2 stability tests and he says that --3 THE COURT: Okay. Please, let me just get the document first. So what paragraph? 4 5 MS. WACKER: Seven, paragraph 7. The March 23rd, 2021 report. 6 7 THE COURT: Okay. Hold on. Let me read it. 8 All right. Now, so this is kind of funny. 9 looks like Dr. Kirsch read paragraph 8 the way I read 10 paragraph 8, because what Dr. Kirsch says is in his 11 February 2021 report, Dr. Park newly identified original 12 Vasostrict lots, blah, blah, blah, blah. 13 Dr. Park alleges that he cited stability testing 14 for these lots in his reply report "in the context of challenging the criticality of the claimed pH range," and 15 16 that's how I read it. 17 MR. BLACK: Your Honor --18 MS. WACKER: It's explained in his reply report. 19 Now he's saying --20 THE COURT: What do you mean, now? 21 MS. WACKER: In his --22 THE COURT: Right now? 23 MS. WACKER: Sorry. 24 THE COURT: The only reason we're only looking 25 at Dr. Kirsch's report is because you said that Dr. Kirsch

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basically inferred from paragraph 8 what you're alleging paragraph 8 disclosed.

And I asked you, I said, I read paragraph 8.

All you're basically saying is that very, very broad, hey,
we got this new data and it's reflective of prior art,
period. It doesn't say anything else. Right?

And so you say, oh, no, no, no, no, no. This is a disclosure about stability testing and arguments you're making about stability testing. And I specifically said, okay. It looked to me maybe, maybe you've got an argument about it could be used to rebut a claim that the pH range was critical, the new pH range, maybe, because I read eight in the context of seven.

You say, no, no, no, it's broader, and you say, the proof of that, look at what Dr. Kirsch said. You point me to what Dr. Kirsch said. A least so far, maybe something else, he interprets it the way I do. What else?

MS. WACKER: He keeps going. He says that stability testing shows that every pH measurement for these original Vasostrict lots was 3.6 or lower.

THE COURT: Right. A pH range.

MS. WACKER: Yes. And then goes on to say, if
Dr. Park is correct that these lots are prior art reflective
of the commercial original Vasostrict product, this would
further support my opinion that he has not established that

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any original Vasostrict product that was on sale or in public use before the priority date of the asserted patents satisfied the pH limitations of the asserted patent. THE COURT: We're back to pH limitation. are we in homogeneity? I don't see it. MS. WACKER: When stability is mentioned by Dr. Park, he does limit it to stability. He said the stability data. THE COURT: I don't think he disclosed it. I'm going to sustain the objection. Now, to the extent, just so we're clear, Mr. Black, if there are critical limitations about pH in that document, then we've got a different issue, but they have not asked any questions about criticality. MR. BLACK: Yes. If they want to say the pH was 3.6 or lower during its shelf life, I'm fine with that. THE COURT: All right. We can bring the witness back in. BY MS. WACKER: All right. Dr. Park, on slide 62 we have DTX-1362, 8 Do you have an understanding as to whether Vasostrict lot 788435 was ever sold? That sales record shows that it was sold from Yes.

Q. Now, turning to pH limitation here 3.4 to 3.6 --

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sorry, pH limitation 3.7 to 3.9, how does that compare to the pH provided in the March 2015 Vasostrict label DTX-132? THE COURT: Ms. Wacker, I'm sorry. I got you up at the maximum volume. When you turn to the screen, we lose you and that coupled with the speed is really hard sometimes. BY MS. WACKER: So, Dr. Park, how does the pH limitation of 3.7 to 3.9 compare to the pH provided in DTX-132, March 2015 Vasostrict label? Original Vasostrict label indicates that pH was adjusted to a pH of 3.4 to 3.6. I'm not a lawyer, so I don't know the details about how it works, but as I understand, 3.6 is abutting arrange of 3.7 that make the claims obvious. So claim range requires some criticality and I understand that Dr. Chyall will address it later. And now you understand that Dr. Kirsch and Par are Q. reading 3.7 to 3.9 to go down to 3.65? According to -- yes, 3.65 can round up to 3.7. Α. Okay. 3.4 to 3.6 would go up to 3.64? Q. Α. Yes. So in your opinion, would a person of ordinary skill in the art expect a difference in the stability of a Vasopressin formulation that has a pH of 3.64 versus a

formulation of Vasopressin that has a pH of 3.65?

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Α. Well, I don't think any POSA will read that change. Again, as yesterday we went through, pH 3.64 and pH 3.65, there's only two percent difference in hydrogen ion concentration. And now thinking about how Dr. Kirsch and Par are reading the claims, a formulation that passes through 3.7 to 3.9 limitation for any period of time, would a person of ordinary skill in the art expect a difference in stability between a formulation that has a pH of 3.6 for its entire shelf life and a formulation that has a pH of 3.6 for its entire shelf life other than for five minutes where it's at 3.7? As I mentioned, even if you have 3.7 throughout shelf life, the difference is only two percent. I don't see how being at 3.7 for five minutes would change anything. Now, did you take a look at the pH specifications that were in place for original Vasostrict? Α. Yes. And DTX-26, at page 26 on slide 24, what were the pH specifications that were in place for original Vasostrict? Α. For original Vasostrict, the release specification is pH 3.3 to 4.0, and the way the shelf life, and during the shelf life, the specification 2.5 to 4.5. So when Par was manufacturing original Vasostrict, if

it manufactured a batch that had a pH of 3.7, could that be

1 released? 2 Of course, because it's within 3.3 and 4.0. 3 And if Par had released a batch of original Vasostrict that had a pH of 3.6 and drifted into the pH range of 3.7 to 4 3.9, would that be within its shelf life stability? 5 Again, yes, because it's well within 2.5 to 4.5. 6 7 Q. I'm only going to talk about, and I want you to focus only on the pH on this slide 65? 8 9 Α. Yes. 10 And so going back --Q. 11 MR. BLACK: I want to be clear, Your Honor. Ι 12 object to the slide, lot 788435. It has a lot of 13 information --14 MS. WACKER: This exhibit is in evidence. 15 Wait. Time out. It is in evidence? THE COURT: 16 The exhibit, we don't have an MR. BLACK: 17 objection to the document coming in, but we have an 18 objection to them talking about it, which was sustained, and 19 we have an objection to her showing a slide which is making 20 points which Your Honor has said are beyond the report. 21 know it's a bench trial, but --22 Okay. But just so I get it, the THE COURT: 23 document without the highlighting is in evidence? 24 MS. WACKER: Yes. I was going to ask him about 25 pH.

1 THE COURT: Can you put up a clean document? 2 MS. WACKER: Can I get DTX -- oh, perfect. 3 THE COURT: Now, just so I will know, see the highlighting there, is that in the original? 4 5 MS. WACKER: No, it's not. 6 THE COURT: Okay. All right. Well, now it's on 7 anyway. 8 MS. WACKER: This is DTX-360 at page 25. This 9 is again lot 788435. 10 BY MS. WACKER: 11 And what's was the pH of this product over stability, 12 over its shelf life? Over 24 months, it was 3.6. 13 14 And would you expect that there would have been a difference of impurities for this lot if it had drifted to 15 3.65? 16 17 MR. BLACK: Objection, Your Honor. Impurities 18 of lot 788435, that whole issue was stricken. He's allowed 19 to talk about pH but not impurities. He's trying to get it 20 in through the back door. 21 MS. WACKER: I was asking, trying to ask him if he expected a different formulation if they changed the pH. 22 23 THE COURT: All right. 24 MS. WACKER: Anyway, I already asked the 25 question and he gave an answer.

1 THE COURT: Well, wait, wait, wait, wait. 2 out. You already asked the question? 3 MS. WACKER: I asked an earlier question that I think covers it. 4 5 THE COURT: Don't you think it was in violation 6 of what my ruling was about five minutes ago? 7 MS. WACKER: Sorry. 8 THE COURT: Let's not try to back-door a document into evidence that I've already said can't come in. 9 10 MS. WACKER: I apologize, Your Honor. 11 THE COURT: I will strike the question. 12 Let's move on and let's try to avoid doing that in the 13 future. 14 BY MS. WACKER: 15 All right. So let's turn to original Vasostrict registration lot 310571. Is this one of the original NDA 16 17 submitted batches that Par relied on in its NDA? 18 Α. Yes, it is. 19 All right. And what were the impurities for this lot, 310571? 20 21 Initial indicated .8 percent, and after three months it's 1.8 percent. 22 23 And can we go back a slide? All right. And did you 24 look at the pH information for lot 310571? 25 Α. Yes. As you can see here, it was 3.5, but at

1 | 18 months time point, it has risen to 3.8.

- Q. Okay. And was that 3.8 measurement within the specification for this lot?
- A. Yes.

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- Q. And was Par relying on this lot in support of its NDA submission?
- 7 A. Yes, because it was a registration batch.
- 8 \ \Q. And was this information submitted to the FDA?
- 9 | A. Yes.
- Q. All right. Now we'll go to the next slide. Again,
 what was the stability information for this lot 310571?
- A. So it start at .8 percent and rose to 1.8 percent. So somewhere between time zero and three months, the level going through .9 to 1.7 percent.
- Q. And how does that compare to the impurity limitations of the asserted claims?
- A. The asserted claims again is .9 to 1.7 percent, and anytime between time zero to three months you should have the claim met.
 - Q. And this lot would have met the pH limitations of the asserted claims as Par is reading them?
 - A. Yes, because 3.8 was at the time 18-month point, so according to the interpretation Dr. Kirsch and Par had, any time you reach shelf life, the pH, the claimed range even for a few minutes, it infringes.

Q. And how did the impurities of the registration lot 310571 compare to the dependent claim?

A. So on the left side, there are individual impurities highlighted, and we have a name, like Gly9. And the limitation, as you can see on the right side, claim 2 of the '209 patent. For example, SEQ ID No. 2, but it also says Gly9.

at this. We can see that initial impurity .1 percent.

After three months, it was .5 percent. So if you go back to the right side, the limitation is .1 to .3 percent. So clearly, .1 percent to .3 percent is met by this lot.

So we can go back to the left side? Gly9, look

- Q. So in your opinion, would the impurities for lot 310571 have satisfied the dependent claims that are asserted of the '209 patent?
- A. Yes. If you go through, yes, it is.
- Q. And your opinion, impurity information for lot 310571 from DTX-47 at page 7?
- 19 A. Yes.

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- Q. And this is part of the NDA submission for the original Vasostrict product?
- 22 A. Yes. It was registration lot.
 - Q. Now, we saw earlier that the release specification for original Vasostrict was 3.3 to 4.0. Have you seen any evidence that commercial lots of original Vasostrict had a

- 1 pH within the 3.7 to 3.9 range at the time of release?
- 2 A. At the time of release? Yes. That's the one document
- I was thinking about. Thank you, yes.
- 4 | 0. And what is DTX-1314?
- 5 A. This is the lot 788436, so typical analysis.
- 6 Q. And what was the pH of this lot on release?
- 7 A. It was 3.7.
- 8 \ \Q. And how does that match up to the asserted claims?
- 9 A. The asserted claim is pH between 3.7 and 3.9.
- 10 Q. And do you know whether this lot 788436 was ever sold?
- 11 A. Yes.
- 12 Q. And turning to slide 72 at DTX-1362, pages 9 to 2,
- 13 when was this lot sold?
- 14 A. You can see the sales record indicates it was sold in
- 15 | November of 2015.
- 16 \ Q. Let's turn to the impurity limitations. Did you look
- at the impurity profile for 788436 and compare it to the
- 18 requirements of the asserted claims?
- 19 A. Yes, I did.
- 20 \ \Q. Okay. And what is your opinion with respect to the
- 21 | impurity level of lot 788436?
- 22 | A. Well, impurity level of lot 788436 is .7 percent:
- 23 **1.6** percent.
- 24 \parallel Q. Okay. And that the homologous impurity of 0.7, that's
- 25 **at release?**

A. Yes.

Q. Okay. And that's at -- that's below the range of .9 to 1.7?

A. Right. We narrowed it over time, Vasopressin degrades, so the impurity level increases. So from here, impurity level would only increase .7 percent to .9, et cetera. The difference in impurity level would depend on how you store it. If you store it at refrigeration temperature, it may increase slowly, more slowly than stored in the room temperature.

- Q. All right. So in conclusion, what is your opinion with respect to whether or not the independent claims of the '209 and '785 patents are met by the original Vasostrict product?
- A. Well, as we have gone through original Vasostrict label had all the elements of the claim, so the Vasostrict label makes the independent claims of the '209 and '785 patent.
- Q. Okay. And with respect to the pH limitation, when considering the original Vasostrict product as sold, do you believe that that limitation would be met by the prior art?
- A. Yes. Again, it was 3.6, and as I mentioned, it is abutting range to 3.7. I mentioned clearly I'm not a lawyer, so I don't know how the intricacies of how law

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works, but as I understand, 3.6 is the abutting range with 3.7, especially when we consider 3.64, 3.6. 3.65 to 7. difference is really minor, so I think it met the claims. And what is your opinion with respect to whether or not the original Vasostrict met the limitations of the dependent claims? Again, as we have gone through one by one, original Α. Vasostrict label indicates that all the impurity claims are So original Vasostrict renders dependent claims of the '209 and '785 patents obvious, too. All right. And if a POSA obtained the original Vasostrict product that was sold, would they be able to measure the impurity level of that product? If they have a sample, yes, they can measure, so impurity level will be there. You can measure pH impurity level from the product. So a person of ordinary skill in the art would be able to determine the properties of that product that was sold? Properties with inherent value, you can Right. Α. measure pH any time, so you can measure pH whenever there's a sample available. Do you understand that we're doing a little bit of prebuttal, what we think Dr. Kirsch is going to say here with respect to some of the teaching away evidence, but do

you understand that Dr. Kirsch has opined in this case that

1 there are certain references that teach away from the 2 claimed invention?

Yes, I do. Α.

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- Okay. Do you agree with him, the references teach 4 5 away from the claimed invention?
- I have not seen any, any evidence, any document 6 No. 7 that indicates that it teaches away from having a formulation that has a pH only for a few minutes during 8 9 storage.
- 10 And if looking at specifically PTX-145 at pages 302 to Q. 11 303, what do you understand this document to be?
- 12 This is the biopharmaceutics review. The FDA reviews Α. 13 and renders a decision like this.
- 14 And this is an FDA review with respect to Par's 0. original Vasostrict product? 15
- 16 Α. Yes.

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- 17 All right. And do you understand Dr. Kirsch is relying on that statement at the top of slide 76 in support 19 of his argument that they're teaching away from the claimed 20 invention?
 - That's what he understood, yes.
 - And would you understand this statement to be teaching away from a formulation that was manufactured at a pH of 3.4 to 3.6 that then drifted into a pH of 3.65 for five minutes?
- 25 I don't see any teaching like that from this

statement.

and that's what they rely on.

Q. Why do you disagree with that opinion of Dr. Kirsch?

A. Now, Dr. Kirsch may rely on this particular statement here. The pH of the formulation is critical because at pH's below 3.4 and above 3.6, degradation Vasopressin accelerates

Now, the issue always for scientists is without data, how do you know, how can you predict this, especially when interpretation by Dr. Kirsch and Par is that you can maintain pH 3.4 to 3.6 and go to 3.7 for five minutes, it still infringe.

Now, in that case, we want to know if pH goes to 3.7 for five minutes, how does it accelerate degradation?

This is why scientists need to see the data. Without data, anything that you say is speculation. That's why we always rely on data. So I don't see any data here.

So, first of all, I think the statement is rather vague and uncertain, especially in relation to Dr. Kirsch and Par's interpretation of the infringement.

- Q. Have you seen any information regarding the pH limitations for -- that were in the prior art before the priority date?
- A. I'm sorry. Your question again?
- Q. Have you seen any information with respect to the pH of, disclosed for Vasopressin injection formulation in the

- 1 prior art?
- 2 A. Oh, of the formulation? Yes.
- 3 Q. And on slide 77, at the top here, DTX-135, this is the
- 4 USP that we looked at earlier?
- 5 A. Yes. The monograph information.
- 6 Q. And what pH value does the USP provide?
- 7 A. Again, it's 2.5 to 4.5, which includes 3.7 and 3.9.
- 8 | Q. And on the top right, DTX-144 at page 3, what is
- 9 | DTX-144?
- 10 A. This is a Lithuanian patent. A patent clearly
- 11 | indicates that pH is tested and its value must be in the
- 12 range of 3.8 to 3.95. So clearly that range was already
- 13 known.
- 14 Q. And then the Pitressin product that we discussed,
- 15 DTX-188 at page 6, what was the pH of some of the -- of the
- 16 Pitressin that was available in the prior art?
- 17 A. Well, you can see the beginning, you start with
- 18 | 3.7 and measurements were made after three months, 3.8.
- 19 Six months, 3.7. Nine months, 3.7. So within the claim
- 20 range.
- 21 | Q. And we already discussed DTX-1314 at page 1 provides
- 22 | the pH specification for original Vasostrict. Why is that
- 23 relevant for your analysis here?
- 24 A. Why is it relevant to my analysis? This covers the
- 25 claimed range of 3.7 to 3.9.

1 Q. So in view of the prior art references to pH of 2 original -- sorry. 3 THE COURT: Is there an objection? Hold up. Go ahead. 4 5 MS. WACKER: Oh. I will let her finish the question. 6 MR. BLACK: 7 THE COURT: I thought you objected to something 8 else. 9 MR. BLACK: No. I'm objecting. She's calling 10 these prior art, but they are not prior art. The first two 11 documents were published, they are prior art. The bottom 12 two documents are internal Par documents that are not prior 13 art, so to call them prior art --14 BY MS. WACKER: Dr. Park, do you understand that Pitressin was being 15 sold before the priority date of this patent? 16 17 That is what sales records indicate. 18 All right. Q. 19 It was sold before the priority date. Α. 20 MS. WACKER: Can I get DTX-638 up on the screen, 21 page 1. BY MS. WACKER: 22 23 And -- I forget the number now. Oh, 78495. When was lot 78495 sold? April 18th? It was lot 78495 sold 24 25 April 18, 2008?

A. 2008, April 18th, yes.

Q. Can you go back to the slide. And do you understand that this lot of original Vasostrict 788436 that's referenced here on DTX-1314 was sold?

A. Yes.

Q. All right. And now back to my earlier question. A person of ordinary skill in the art having an understanding of these pH ranges and values of the prior art, Vasopressin injection product, how would that impact whether or not they are taught away from the 3.7 to 3.9 range?

MR. BLACK: Objection, Your Honor. Not prior art. The bottom two items are not prior art. They're internal documents.

MS. WACKER: The products, the lot themselves are prior art.

MR. BLACK: She hasn't established that the lots were sold at these various pH levels or that one of ordinary skill would have that information in front of them. She has to ask a precise question if she wants to call them prior art.

BY MS. WACKER:

- Q. Dr. Park, original Vasostrict lot 788436, we established that?
- 24 A. Yes.
- 25 Q. It was released with a pH of 3.7?

A. Yes.

Q. And a person of ordinary skill in the art, I think you testified to this earlier, would be able to measure the pH of that lot when it was sold, right?

- A. Again, pH is an inherent property that does not change, so anybody can measure it. So once product is sold, its property belongs to prior art. Again, I'm not a lawyer, but that's my understanding.
- Q. Okay. And would that be the same for the Pitressin lot, 78495?
- A. That's right.
 - Q. So now you have these two lots that are available with the -- that have pH's in the 3.7 to 3.9 range at some point in their shelf life and the prior art on the top, how would that impact whether a person of ordinary skill in the art would be taught away from 3.7 to 3.9?

MR. BLACK: Objection. Assumes facts not in evidence about 788436, that it was ever sold at a pH of 3.7.

The trouble is, if she wants to ask a question about whether USP 2009, for instance, would have taught something and what it would have taught to someone of skill in the art, that's a published document. If she wants to ask a question about an internal Par document for a lot that was sold, she has to establish that somebody had that

1 information. 2 POSAs don't test millions of vials sold. 3 There's one that came out at 3.7, what did that teach? We say nothing, but she has to ask a question. You can't call 4 5 it prior art unless she can establish it. BY MS. WACKER: 6 7 In your view --8 THE COURT: Hold up. What do you think? What he said sounds right to me, so what do you say? 9 10 MS. WACKER: I don't want to argue too much in 11 front of the witness, but this is the issue. They are 12 saying you have to establish it at the time that it was 13 sold, at the time it was administered, and what Dr. Park is 14 saying is that the information that we have, because we have only what Par has been able to produce to us, right, that's 15 representative of their original Vasostrict product, that 16 17 the information we have demonstrates that their products and 18 the earlier Pitressin products likely had a pH over its 19 shelf life in the same range. THE COURT: Let's do this. I'm sorry to ask 20 21 you, Doctor. 22 THE WITNESS: Yes, sir. 23 THE COURT: If you need to use the restroom, go 24 ahead.

25 THE WITNESS: Yes.

THE COURT: All right. It seems to me, I mean,

(Witness excused.)

pH.

I understand the objection, and so do you want to respond?

MS. WACKER: Yes. And so I think Dr. Park's opinion is that it's the original Vasostrict product that was sold as a whole, right, and so this is representative of the pH and he has shown a couple of examples of the pH that's representative of the original Vasostrict batches that were being sold and how they had multiple instances where it was released at a higher pH or rose up to a higher

THE COURT: Doesn't it have to be known to the artisan of ordinary skill that when it was released, it had the pH at whatever level you said?

MS. WACKER: Not for an on-sale bar. So for a product that's actually sold, and you can consider on-sale products --

THE COURT: Okay. You know patent doctrine.

I've heard of on-sale bar. I have no idea the relevance to this particular issue, so you are going to have to explain why.

MS. WACKER: Yes.

THE COURT: I thought that what is relevant here is we're on teaching away and we're on whether or not certain prior art references would teach something to an

Park - direct

artisan of ordinary skill. And Mr. Black has said, he has admitted that the Pitressin that was on the market, that, the fact that it's on the market and it's out there is prior art, but whether or not it had certain properties that you have to test for, if those test results are not public information, we can't assume that that information, the test result information, is prior art. That makes sense to me, and I think you agree with it?

MS. WACKER: I don't, I don't.

THE COURT: Okay.

MS. WACKER: So when you have a product that's sold, that you will be able to know the properties about, because if we were back in 2016 when this product was being sold, we could test it. Right? You could measure the pH, you could measure the impurities.

So if you have evidence as to what the properties of that product were, that is representative of the product that was being sold and you can use that in part of your obviousness defense.

So even though a POSA can't go back in time and test it, if those are the properties of the product, that's representative of what was being sold. So original

Vasostrict was being sold by Par. They had very broad pH specifications.

Now they are trying to say, well, that can't

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invalidate our patent even though you are copying it because you don't have specific evidence that we sold a batch on a day when it had those impurities and we're never going to have that evidence because you can't go back in time and take something from a hospital and measure it. So that's why the law is you can see what the properties are of that original product that was being sold. The same for Pitressin.

THE COURT: Okay. Mr. Black, what do you say to that?

MR. BLACK: I disagree that she is using -- let me put it this way. When a POSA is looking at the prior art, they can consider all the prior art that's available. They look to documentation, they look to standards, they look to the FDA that teaches away from 3.4 to 3.6.

They look at the documents that are out there.

They might also have other evidence. They might have -- but they have not shown that would be so in the real world.

And what they certainly can't do is say that because there was one lot of Pitressin that at one point in its life was 3.7 before it was sold, nine months before it was sold, that that is prior art, and they can't consider it alone.

The standard for Pitressin was 3.4 to 3.6 and all -- original Vasopressin. All the evidence we have,

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there were millions of vials actually sold at 3.6. They found a stray 3.7 value and are representing it as prior art.

THE COURT: Kind of like SVA1 is a stray value that has been --

MR. BLACK: No, Your Honor. SVA1 is reflective of the fact that their product has an upward statistically significant drift of the .024 pH at the time they registered the product, and when I cross-examine Dr. Park, I am going to show the same thing is true of their so-called optimization batch.

This is prior art. They're trying to invalidate our patent with non-public material. There are rules for that. It's wrong for her to say this is the prior art, this is what would have been taught. It's a misleading question.

There are plenty of ways to get the evidence in and then argue from it, but they're arguing that someone would teach away from the FDA if we ignore the FDA ruling because there's one document an internal document from Par. That doesn't make sense. It's also so far away to be relevant, it's inadmissible. We should move on.

MS. WACKER: And I think what he's saying is if a person of ordinary skill in the art understanding the properties and their original Vasostrict and Pitressin in prior art, knowing that they can go in the 3.7 to 3.9 range

1 would not be taught away from 3.7 to 3.9. 2 I have a case for you, Abbott versus Sunovion 3 Pharmaceuticals. 4 THE COURT: Well, here's the thing. 5 appreciate the case. I'm going to have to study the law and make an informed decision about this. 6 7 I am going to allow the question to be forward. You need to think about the risks of contaminating what 8 9 otherwise might be probative and what I would admit by 10 combining all these things together. I will leave that to 11 you. That's your discretion. 12 I will overrule the objection, but I'm going to 13 study the issue, and if Mr. Black turns out to be right, and 14 I have no way of knowing until I read the case law, you may have taken a great risk. But you'll have to decide how to 15 16 proceed. 17 All right. Let's bring the witness back in. 18 Thank you. 19 MS. WACKER: May I proceed? 20 THE COURT: Yes. 21 BY MS. WACKER: So, Dr. Park, just considering the top two 22 All right. 23 references, USP 2009 and the Lithuanian patent, having those 24 two prior art references in mind, would a person in your

view, would a person have been taught away from the 3.7 to

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3.9 range?

A. No. First of all, USP monograph that all formulation scientists follow clearly indicate pH between 2.5 and 4.5.

Also, there are patents claiming 3.8. The only document Dr. Kirsch referred to was a document with review. As I mentioned, the document statement is uncertain, vague without data, especially when we interpret the infringing claims that pH remains 3.4 to 3.46 for 24 months. At the end, only for five minutes goes to 3.7.

The question naturally occurs, how fast degradation accelerates. Does it make the whole product unusable? We don't know. It's new information. So overall, I don't think that teaches away. And I have not seen any document that indicates it teaches away.

- Q. And now a person of ordinary skill in the art understanding the properties of the Pitressin and original Vasostrict products that were on the market in the prior art, would that person of ordinary skill in the art have been taught away from the 3.7 to 3.9 range?
- A. No. Again, anyone can measure the pH and as we've seen here, some vials have 3.7. Yes, that's exactly what it's saying. Some vials, 3.7. So we have seen the data, product sold has a pH of 3.7. Anybody can measure.
- Q. All right. So let's turn to your opinions on materiality and here we have claim 1 of the '239 patent,

PTX-605, page 33.

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Can you describe generally what this claim is directed to?

- A. Yes. This claim is related to a method of treatment using Vasopressin formulation having a certain impurity profile drug concentration and with certain pH information.
- Q. Okay. And in the context of the '239 patent, what reference did you consider with respect to materiality?
- 9 A. Oh, again, 2014 data label. It has more information.
- 10 Q. Okay. That's DTX-36, pages 1 to 3?
- 11 A. Yes.
- 12 Q. Okay. And are you relying on Dr. Cross for any of these limitations?
 - A. Again, method of treatment is required, a medical doctor's statements.
 - Q. And so in your opinion, are you relying on Dr. Cross that the April 2014 Vasostrict label, DTX-36, satisfies the method of treatment limitation of claim 1 of the '239 patent on slide 8?
 - A. Again, any POSA can read the claims in the patent, but, again, I'm relying on Dr. Cross for method of treatment.
 - Q. All right. Now, when looking at the April 2014

 Vasostrict label, DTX-36, how does that compare to the formulation aspect of the '239, claim 1?

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Α. Now, this is the most interesting part. Now, label clearly indicates pH was adjusted to 3.4 to 3.6 and '239 patent claims pH 3.5 to 4.1, which includes 3.4 to 3.6. And how about the amounts, the unit amounts? How do 0. they compare of the April 2014 Vasostrict label? How does that compare to the limitations? Again, the label clearly indicates that one ml Α. solution contains Vasopressin with 20 units per ml, which we went through, within 0.07 mg/ml -- the claimed range. Also it has a chlorobutanol as a preservative and water and acetic acid. So label has every single element claimed in the '239 patent. Now, with respect to impurities, during prosecution, what did the Examiner state in the '239 patent? Well, the '239 patent, the claims indicate the level Α. of zero to two percent, individual impurity as claimed in the claims 2, 3 and 4. Now, the Patent Examiner stated that --MR. BLACK: Objection, Your Honor. Hearsay. He is going to testify about what the Patent Examiner said where the issue here is whether or not the degradation product limitation of the '239, zero to two percent. have not seen that before, but it's an impurity related measurement, zero to two percent degradation product.

Hold on one moment.

THE COURT:

1 MR. BLACK: Yes, I'm sorry. 2 THE COURT: All right. So the question is with 3 respect to impurities during prosecution, what did the Examiner state in the -- it's a hearsay objection. 4 5 MR. BLACK: And relevance. THE COURT: Okay. Well, let's deal with the 6 7 hearsay first. What is this being -- it's clearly, it's an out-of-court statement made by a declarant who is not on 8 9 the stand. Unless you come up with an exception, it's 10 hearsay. MS. WACKER: Public record is 8038. This is an 11 12 official prosecution history filed for a patent. 13 THE COURT: Okay. 14 MS. WACKER: And he's just reading a statement 15 from that public record. THE COURT: And what is it being offered for? 16 17 MS. WACKER: The fact that the Examiner said that the concentration --18 19 THE COURT: The fact -- why is it relevant? It's relevant to inequitable 20 MS. WACKER: 21 conduct. 22 THE COURT: Go ahead. 23 MS. WACKER: All Dr. Park is doing is comparing 24 the prior art label, which is something that as you heard 25 during Mr. Hales' opening, Par submitted declarations so

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Park - direct

that this label would not be considered as prior art for the '239 patent and then we're having Dr. Park just walk through the label and compare it to the claim. MR. BLACK: My complaint is that they're relying on a statement from the Examiner in the middle of prosecution. Those statements often are reversed. become moot. He's not qualified to talk about what the Examiner said and what the Examiner said during prosecution on the point, scientific point about the degradation product is irrelevant. If he has an opinion --Well, wait. THE COURT: Hold on. Hold on. MR. BLACK: I'm sorry. THE COURT: So because you started saying he doesn't, you're now talking about Dr. Park offering an opinion or interpreting. MR. BLACK: Yes. THE COURT: So my first reaction is that if the inequitable conduct is going to be based on assertions that are made in response to what the Patent Examiner asked, then the Patent Examiner's statements aren't being offered for the truth of the matter. They're not hearsay and they just

Now, if they're going to ask, well, what was the

give context and the fact that they were said.

Patent Examiner thinking or anything other than what the statement is and this is just a vehicle to get it read into the record, then do you object?

MR. BLACK: Yes.

THE COURT: Okay.

MR. BLACK: I object to that because that's what I think they are doing. Examiner, degradation product in the April label are inherent. He's going to give his opinion on whether they are inherent. That's okay. He's a scientist.

THE COURT: Right.

MR. BLACK: He can't back himself up with statements from the Examiner in the middle of prosecution. This doesn't really provide context for the inequitable conduct argument.

This part of the presentation is about whether the label matches up with the '239 patent and he's trying to use the Examiner's statement to try to give credibility to that. She's not here to be cross-examined. If he wants to give his opinion, that's fine.

MS. WACKER: And all he's doing here, he's not saying it's inherent. What's relevant to materiality is whether or not this reference would have been material to the issue of the patent. Right? And so this label was disqualified as prior art based on declarations, so he's

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Park - direct

comparing all of the limitations. The Examiner stated she believed these limitations were inherent. For the '239 patent then, which satisfies all of the pH, all of the other limitations, the Examiner has a view that the SEQ ID numbers, the impurities, were an inherent feature of the product. THE COURT: All right. I'm going to allow the question and I think it's appropriate to put on the record what the Patent Examiner said. It is a public record, first of all. More importantly, we've got an allegation of inequitable conduct before the Patent Examiner, and so -- I mean, let me ask you all. You're both patent lawyers. Have you ever tried a case with inequitable conduct before? MS. WACKER: I have not. THE COURT: Mr. Hales, have you? MR. HALES: I have, Your Honor, yes. THE COURT: Have you defended against inequitable conduct? HALES: Not in the one that went to trial. MR. THE COURT: But you've been in a trial where you argued inequitable conduct? MR. HALES: I take that back. Perhaps not. can't even remember one, but they're in some obviously and I've been on both sides.

THE COURT: I guess what I'm getting at is, the

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Park - direct

reason why I'm asking the question is I can't believe in patent cases we call to the stand patent examiners. MR. HALES: No, we're not. THE COURT: In fact, my guess is there's probably a bar on it, but I don't know. MR. BLACK: There was a cottage industry with former PTO commissioners doing expert reports, but that was put to bed some years ago. THE COURT: Right. And that makes sense to me as a non-patent person, and if we couldn't put the Patent Examiner's statement into the record, I don't know how you would almost ever prove or be allowed to go forward with an inequitable conduct case. So I'm going to allow the Patent Examiner's statements to be put in the record. I don't think they are hearsay, I think they're an exception, mainly because I don't think they're being asserted for the truth of the matter. Even if they were, I find a public record exception. Now, so the statements are coming in. Now, you've got another objection which is to what? Well, it depends on her question. MR. BLACK: THE COURT: Okay. MR. BLACK: Statements coming in, I don't really have a problem with that part of the part of the file

1 history being in. 2 THE COURT: That's what you objected to. 3 MR. BLACK: I was objecting to the No. question, and this witness, this witness' -- I'm objecting 4 5 to it being offered for the truth of the matter asserted. 6 Your Honor has dealt with that. It's hearsay, but it's a 7 public record exception. 8 THE COURT: It's not hearsay because it is not 9 being offered for the truth of the matter and I'm not going 10 to accept it as the truth of the matter. 11 MR. BLACK: Okay. 12 It goes to the Examiner's state of THE COURT: 13 mind to the extent the Examiner asked something. 14 example, if the Examiner said, I need this information 15 before I can sign off on this patent, that is admissible, it 16 seems to me, because, again --17 MR. BLACK: Yes. That's something we can -- the 18 use of the document we can deal with. I just didn't want to 19 let the moment pass and have their inherency case be put in 20 effectively by a statement from the Examiner that was being 21 offered for the truth. 22 THE COURT: Okay. I see. 23 Dr. Park can testify. MR. BLACK: 24 THE COURT: That's a fair thing. So just --25 MS. WACKER: And I wasn't going to ask him that

anyway.

THE COURT: I mean, in fairness to Mr. Black here, I would be doing what he did given that the title of the slide is degradation products in April 2014 Vasostrict label formulation are inherent, so I understand where he's coming from.

MS. WACKER: I understand.

THE COURT: All right. So, all right.

MS. WACKER: All right.

THE COURT: The objection is overruled. Let's

go.

BY MS. WACKER:

- Q. Dr. Park, and I don't recall if we have an answer to the question. Let me re-ask it. What did the Examiner say with respect to the impurity limitations in the '239 patent?
- A. Well, the record shows that the Examiner said a concentration of SEQ ID numbers 2 and 4, which are in brief here, in the formulations is an inherent feature of the prior art formulation, which is original Vasostrict.
- Q. Okay. And now the original Vasostrict product, do you know what the impurity limitations were for that product?

 And I think we went over this before, lot 310571. We have an excerpt.
- A. What was the question?
- Q. What were the impurities for that original Vasostrict

1 product? 2 Oh, the samples stored upright position at 25 degrees 3 Centigrade. Initially, we think the level was 1.1 percent. It rose to 2.4 percent at three months. So it clearly meets 4 5 the claim limitations of their two percent degradation 6 product. 7 So in your opinion would the April 2014 Vasostrict label have been material to claim 1 of the '239 patent? 8 9 MR. BLACK: Objection, Your Honor. Materiality 10 is a legal question. 11 BY MS. WACKER: 12 Was the April --13 THE COURT: Hold on. Are you withdrawing the 14 question? 15 I have withdrawn the question. MS. WACKER: 16 THE COURT: Okay. Sorry. Go ahead. 17 BY MR. WACKER: And in your opinion, would the April 2014 Vasostrict 18 19 label have invalidated the claims of the '239 patent? 20 Again, information described in 2014 Vasostrict label, 21 has all the elements claimed in the '239 patent, so I believe that it invalidates the '239 patent. 22 23 All right. Let's turn to the '209 patent. 24 Are you relying on Dr. Cross, this is on slide

85 and looking again at DTX-36, pages 1 through 3. Are you

1 relying on Dr. Cross with respect to the method of treatment 2 limitations here? 3 Yes. Again, I'm relying on Dr. Cross on method of treatment. 4 5 How does the April 2014 Vasostrict label compare to the formulation limitations of the '209 and '785 patents? 6 7 Again, as we went through, the amount of Vasopressin, Α. 20 units per ml, which is equivalent to .0377 milligrams per 8 9 ml, which is within the range claimed in the '209 and '785 patents. 10 11 And how does the pH limitation of the '209 and '785 12 patents compare to the pH described in the April 2014 13 Vasostrict label? 14 Again, I went through, so I'm not going to go through Α. all answers again, but 3.6 is abutting the range to 3.7. 15 And what were the impurity levels for, again, for lot 16 17 310571 and how did they compare to the claim limitations? And here on slide 87, we have an excerpt from DTX-45 at 18 19 page 7? 20 Again, the samples show that upright position at 25 degrees Centigrade, initially .08 percent at three 21

21 25 degrees Centigrade, initially .08 percent at three 22 months, 1.8 percent. So in between initial and three 23 months, the treatment level must have gone to .1 to 24 1.7 percent.

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Q. So in your opinion, would the April 2014 Vasostrict

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label invalidated the '209 -- let me repeat that. Was the April 2014 Vasostrict label have met all of the limitations of the '209 and '785 patent, claim 1? 2014 label meets all the elements except the pH Α. Yes. 3.6, which is the abutting range. So I think it makes the independent claim of the '209 and '785 patents obvious. MS. WACKER: No further questions. THE COURT: All right. Thank you. Mr. Black? MR. BLACK: Take a short break, Your Honor? THE COURT: Well, we're coming up on lunch, so I mean, you're going to be probably a while. Right? I'm going to be a while, yes. MR. BLACK: THE COURT: Why don't we go. Then we'll break for lunch. Is that all right? THE WITNESS: Yes. THE COURT: Okay. Thank you. THE WITNESS: For the record, I'd like to spell my name, K-i-n-a-m. Kinam is a good name, too. MS. WACKER: Your Honor, I forgot to move exhibits in. Do you want me to do that later? THE COURT: Just do it later. MS. WACKER: It's a long list. THE COURT: Then especially. Whenever you are ready, Mr. Black.

Thank you.

MR. BLACK:

1 CROSS-EXAMINATION 2 BY MR. BLACK: 3 Good morning, Dr. Park. 0. Good morning. 4 Α. 5 You started by discussing your qualifications and you are clearly an eminently qualified expert in pharmaceuticals 6 7 and with an expertise in peptides, and I think you would agree that Dr. Kirsch is a colleague who also shares those 8 qualifications. Would you agree? 9 10 Absolutely. Α. 11 Q. And Dr. Winter actually is part of the peptide club as 12 well. Right? Do you know him? 13 No, I don't know him. Α. 14 You don't know him. Okay. You do know Dr. Kirsch. 0. 15 Dr. Shawl is not a peptide expert; is that 16 correct? 17 Dr. Shawl? Α. 18 Q. Yes. 19 I don't think so. Α. 20 Q. You --21 Α. But I don't know. Just want to clarify. 22 He's not somebody who you run into at conferences, Q. 23 like Dr. Kirsch, see publications frequently. Correct? 24 I'm sorry. Dr. Kirsch? What about Dr. Kirsch again,

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please?

Park - cross

Q. You and Dr. Kirsch see each other at conferences, read each other's publications from time to time; correct?

A. Yes.

Q. That would not be true of Dr. Shawl; correct?

A. I have not met him, so, yes, that may be true.

Q. Let's discuss the ANDA process a little bit. Now, the ANDA holder files an application with the FDA, which is a voluminous set of documents and data where they try to establish sufficient similarity to an approved drug so that they can sell the generic drug. Right?

- A. Yes. Generic product needs to show safety and efficacy similar or same as the RLD.
- Q. And there are many variations between the generic drug and the RLD that are permitted by the FDA; correct?
- A. Could you be more specific, what variation.
 - Q. You put up a slide that implied that the only thing that mattered were in this case the pH and a few other things, but the FDA will actually permit significant variation in characteristics between the generic drug and the RLD; right?
 - A. Well, I cannot answer in general. I don't know the FDA regulations.
 - Q. The impurity profile of the generic drug can be significantly different from the impurity profile of the RLD; right?

A. That I don't know.

Q. You don't know?

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- A. I don't know the FDA regulation.
- Q. Okay. In a case like this, we only have a limited amount of data available to determine what the commercial product will actually be like; correct?
- 7 A. I don't think that's true.
- Q. In a typical ANDA case, you have a few batches, some data on those batches and the FDA has to make a decision about whether to approve the product based on those limited batches, but, of course, in the real world, there will be many, many, many commercial batches; is that correct?
 - A. In real world, there are many more commercial batches, but FDA requires testing certain batches and the product, can the product be represented by the stability study.

 That's what Dr. Kirsch understood. Do you present whole product using stability study, and that's what FDA requires.
 - Q. But the actual data that's available in a patent trial like this is much more limited than the data that would be available if we had pH and impurity readings from all of the commercial batches; correct?
 - A. What do you mean all of the commercial batches.
 - Q. Dr. Park, if we had a thousand commercial batches that

 Eagle has, is planning to make, we'd have a lot more

 information about the commercial product and how it works in

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Park - cross

the real world than we can get from just looking at the handful of batches that have been made to date; correct? Once again, ANDA, all NDA process such that product is represented by the stability study. Nobody measured every single vial of every batch. If you measure every vial of the whole batch, there's nothing to sell. That's not the point of a production FDA. FDA required testing certain number of samples to represent the batch. That's what FDA requires. And how many batches were tested using a pH of 3.64 at release using the optimized process that Eagle has claimed in this case? Well, I can go back to the table. I cannot recall that, but let's put the table -- I gave you the table before. Right? Let's look at the table and go through it and we can find out. Okay. Let's do that. Go to the Elmo. All right. Thwarted again. It's not my forte here. I broke the screen? THE COURT: I don't think you broke it, but check it out at the next break. Certainly, everybody is welcome to try out the equipment. BY MR. BLACK: All right. Let's try again. Okay.

THE COURT: I think you may have it fixed.

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Yes.

Park - cross

MR. BLACK: Do you have it? All right. Well, riveting cross, I know. Let's go to DTX-993, which is the summary of the data. BY MR. BLACK: All right. This is a summary of the data that you referred to a moment ago. Do you have it in front of you? Can you read it? Α. Yes. Okay. Now, I understand it's your position that batches SVA001 through 6 are no longer representative of the commercial product; correct? I said no longer represent pH of the commercial product. No longer represent the pH of the commercial product. Q. All right. So let's take a look at this chart and we'll start with SVA7, and what it shows in the first column, there's an initial pH at release; is that correct? Yes. Α. And then the second column we have one-month data, then three-month data, et cetera; correct? Α. Yes. All right. So if we start with SVA7, the initial pH was 3.50; is that correct? Α.

Q. And if you look at that, you'll see that there are a large number of values which are above 3.50, including 3.54 right at the one month date.

Do you see that?

- A. Yes. Several numbers, yes.
- 6 Q. Okay. So 3.54, 3.51, 3.55, 3.51, 3.51, 3.52,
- 3.51, 3.51 are all values which are above the release value for SVA007.

9 Do you see that?

- 10 A. What do you mean above a release value?
- 11 Q. The initial value at 3.50 is less than nine values
- that were recorded during the shelf life of SVA7; is that
- 13 correct?

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- 14 A. As I understand release value, 3.4 to 3.6.
- 15 \ Q. No, no. I'm asking a question. This is actual data?
- 16 A. Right.
- 17 Q. You said we have to decide this case based on actual
- 18 data; right?
- 19 A. Absolutely.
- 20 Q. And you say this data is representative of the actual
- 21 | commercial product; correct?
- 22 | A. Yes.
- 23 Q. And you asked me to put this table in front of you to
- 24 help answer the question; right?
- 25 A. **Yes**.

- 1 \ \Q. And SVA7 released at 3.50; is that correct?
- 2 | A. Yes.
- 3 Q. And then Eagle and AMRI got nine values that were
- 4 above 3.50 during the shelf life; right?
- 5 A. I would say around 3.50.
- 6 Q. Sir, the values recorded in this table, which is
- 7 evidence that was provided by Eagle, indicates that there
- 8 were nine values above 3.50 that were recorded in their
- 9 stability data; is that correct?
- 10 A. Above 3.50, yes.
- 11 Q. Yes.
- 12 A. But, again --
- 13 Q. And the highest value -- the highest value above was
- 3.55, which was an increase of .05 over release; is that
- 15 correct?
- 16 A. Right. And you can see at 18 months it goes down to
- 17 **3.46.** Below 3.5.
- 18 Q. Let's try to do it this way. I will ask the questions
- and you can answer the questions I've asked and if your
- 20 counsel wants to follow up with additional information, they
- 21 | have an opportunity to do so. Would that be okay?
- 22 A. Yes. Thank you.
- 23 Q. Thank you.
- Now, SVA08 started at a release of 3.52; is that
- 25 correct?

- 1 A. Yes.
- $2 \parallel Q$. And it had release values of 3.53, 3.53, 3.53
- 3 and 3.53, five different release values -- excuse me, five
- 4 different values during stability that were above the
- 5 release value; is that correct?
- 6 A. You mean compared to 3.52?
- 7 \mathbb{Q} . Yes. I'm comparing the release value against the
- 8 shelf life.
- 9 A. Yes. Compared to 3.52, 3.53 is above, yes.
- 10 \parallel Q. Right. That was a difference of .01 maximum; is that
- 11 correct?
- 12 A. Yes.
- 13 Q. Okay. Then SVA9 started at 3.48 and then immediately
- within a month got a value of 3.52; right?
- 15 A. Yes.
- 16 \ \Q. And then at three months they got values of 3.52 and
- 17 | 3.53; right?
- 18 A. Yes.
- 19 Q. And then 3.50 and lots of other values, including 3.54
- 20 at 12 months; is that correct?
- 21 A. Yes.
- 22 Q. In fact, every single value for SVA9 was above the
- 23 release value of 3.48; isn't that right?
- 24 A. Compared to 3.48, yes.
- 25 Q. All right. And the maximum increase recorded here is

1 .06; right? Difference between 3.48 and 3.54?

A. Yes.

- Q. Now, SVA11, we have a double asterisk there, because we had six starting values. I'm going to come back to that
- 5 and address that later.
- SVA12 and 13, those are recent additions to the case. SVA12 started at 3.48 and immediately jumped up to
- 8 | 3.51, 3.51, 3.52; right?
- 9 A. I agree.
- 10 Q. Thank you.
- 11 A. But I don't like the description immediately jumped.
- 12 Q. Okay. Release value was 3.48, the recorded release,
- 13 the recorded value of stability at one month by Eagle and
- 14 \parallel AMRI was 3.51. The recorded value by Eagle and AMRI at
- 15 three months was 3.51 and the recorded value by Eagle and
- 16 AMRI at six months was 3.52?
- 17 A. Yes.
- 18 Q. Agreed?
- 19 A. Yes.
- 20 Q. Thank you.
- SVA13 -- oh, I forgot something. The difference between 3.48 and the maximum value, we only have six months
- of data, is already .04; is that correct?
- 24 A. Yes.
- 25 Q. All right. And on SVA13, the release value was 3.49

and the first recorded value at one month during stability
was 3.53; is that correct?

- A. Yes.
- O. And then 3.54 and then 3.53; is that correct?
- 5 A. Yes.

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- 6 Q. And the increase from release to the maximum number
- 7 3.54 is .05; is that correct?
 - A. Yes.
 - Q. And so we have a bunch of values based on the so-called optimized batches which show significant drift between the time of release and the stability testing.
- 12 Agree or disagree?
- A. There is a drift, but I would not describe it as a significant drift, because values are all around 3.51, 51, 50, 52.
 - Q. There is a drift, and therefore if Eagle releases a product at the upper end of the -- upper end of the pH range of their approval 3.64, this product will go into the commercial world at a pH above 3.64; is that correct?
 - A. No, no, no, that's not true. We heard yesterday Dr.

 Aungst clearly indicate that pH 3.64 will not be released

 because it doesn't meet the in-process specification. He

 spent hours to explain that.
 - Q. Dr. Aungst testified, I believe, that there would be nothing to prevent the release of a batch that met the

Park - cross 1 in-process spec and the release spec; isn't that right? 2 That is not how I understand. Could you explain to Α. 3 me? 4 0. Sure. 5 THE COURT: Well, wait. I don't want him to testify and I don't want you to ask him questions. 6 7 THE WITNESS: No, no. To clarify. 8 I get you, and it's fair to ask for THE COURT: 9 a clarification if you don't understand something. 10 keep it with questions. 11 MR. BLACK: Sure.

- 12 BY MR. BLACK:
- Q. What is your understanding of the value of the in-process specification top end?
- 15 A. The top end specification, 3.42 to 3.54.
- Q. And therefore the top end would be 3.54; is that correct?
- 18 A. Yes.
- Q. And the release specification goes up to 3.64; is that correct?
- 21 A. Yes.
- Q. So a product that meets the in-process specification
 when it's made of 3.54 and then is released at 3.64 would be
 permissible under the ANDA; is that correct?
- 25 A. That's the point I like to make. Thanks for pointing

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Park - cross

it out. If 3.54 is met as an in-process specification, I don't see how all of a sudden it jumped to 3.64. All the data you describe here, whether they show .17, I don't see it that way, so my answer is no.

Q. Dr. Park, step by step. We'll get there. Right now I just want to know your understanding that if there is a product which has a value of 3.54 in-process spec and if that product drifts to 3.64 and is released at 3.64, the ANDA permits Eagle to sell that product; is that correct?

Well, that I cannot agree with you.

Q. Okay. I'm not asking you whether you think as a matter of science that's going to happen. We'll get there. I'm just asking your understanding of how the specifications work.

The way the specifications works is Eagle is allowed to make a product that just meets the in-process spec at 3.54 and they can still sell it and release it at 3.64. Right? That's what the FDA will permit them to do. That's the evidence in this case, isn't it?

- A. Well, again, I don't know what FDA will do, but if that's the case, yes, but, again I don't see how --
- Q. Thank you for your answer. I would appreciate it very much so, very much if you would answer my question yes, no, or if you have a question about my question, feel free to ask for a clarification, but not add extraneous material.

Your counsel will have an opportunity.

All right. So we've established two things -that the FDA would allow the sale of that 3.64 product and
that there's upward drift between release and stability in
the ANDA product, which we will see in commercial batches;
is that correct?

- A. No, I don't think so. First of all, as I said, I don't know what FDA will do. Also, second, I don't see the .1 upward drift.
- Q. All right. Well, you agree that we've seen upward drift in SVA7 to 13 in the range of .05 to .06. Those are real world values; right?
- 13 A. .06, we were talking about 3.48 to 3.54. Right?
- 14 0. Yes.

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- 15 A. Yes.
- 16 Q. Okay.
- A. .06 went from 3.48, so, again, I don't see how it relates to .1 of which.
- Q. Okay. I'm not sure what that meant, but we can move on to the next point.
- All right. I've got another chart here. Okay.

 If you will take a look -- if we can pull up DDX-2-39 from

 Dr. Park's slide, slide 39.
 - Okay. We've got our marker on release to stability .05 to .06. Let's talk about increase in pH from

any in-process spec to the release.

So in-process and release happen -- the release testing happened very close together in time; is that correct?

- A. Yes. Samples are collected the same place, same time, yes.
- Q. Right. And we have on the chart here, and this comes from DTX-993 and we're using slide DDX-2-39, we have the pre-filtration, post filtration, and then the initial values which, for each of the batches from SVA7 to 13, which you said are representative of the commercial batch; correct?
- 12 A. Yes.

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- Q. Okay. So if we look at the post filtration column and we look at the initial data for SVA11, we see something interesting that you commented on, which is that between the post filtration value of 3.50 and release, we see pH values that are in a range of one-tenth of a pH unit; is that correct?
 - A. I'm sorry. The question was one-tenth of a pH unit?
- Q. You testified before that the initial values for SVA11
 were in a range of one-tenth of a pH unit to the variation;
 is that correct?
 - A. One-tenth means .1 variation.
- 24 | Q. Yes.
- 25 A. **Yes**.

Q. By the way, do you think that variation is caused by the product, by operator error or just random variation in

3 | the vial?

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- A. I think it's accurate. I think it's due to the measurement itself.
- Q. Yes. So just to be clear -- I'm sorry. Did I cut you off?
- 8 A. Vials.
- 9 | Q. Oh?
- 10 A. Because of the vial pH itself.
- 12 Okay. Just to be clear, initial are the release
 values and post filtration is the process, in-process value;
- 13 | is that correct?
- 14 A. Yes, in-process.
- 2. All right. All right. So at least between the 3.50 and SVA11 in-process and the highest initial values, we see a difference of .06 and .07; is that correct?
- 18 A. In a 3.50, 3.56 and 3.57.
- 19 Q. Yes?
- 20 A. **Yes**.
- 21 Q. And those are possible values that have actually been experienced in the real world by Eagle; is that correct?
- 23 A. Yes.
- Q. Okay. And then for the next batch, SVA12, we have a 3.44 in-process reading and then three other readings that

are higher, including a 3.50 reading that shows an increased value of .06; is that correct?

A. Correct.

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- Q. And then SVA13, we've got 3.49 and then one value that's above at 3.50; is that correct?
 - A. Correct.
- Q. Okay. So it's representative of the commercial
 batches to expect a difference in-process versus the release
 reading could be in the order of .07 or .06, because that's
 the data that we have now; is that correct?
- 11 A. Yes.
 - Q. And, of course, we only have limited data. If we had a hundred different sets of this data, we'd see -- undoubtedly, we would see values that are higher than these values?
 - A. No, I don't think so.
 - Q. You don't think so. You think if we did this experiment 100 times, experiment where the variation is .1 of a pH unit, that you would never see a value higher than 3.57 in those 100 experiments?
 - A. It may be higher than 3.57 or 3.58, but if you increase the number of measurements, so the value will be narrower and narrower close to the real value, which is 3.50.
- MR. BLACK: Your Honor, now would be a good time

1	to break so I can fix the Elmo for one thing.
2	THE COURT: Oh, okay. That's a good idea.
3	Let's do that.
4	Can you step down for a minute, Dr. Park?
5	THE WITNESS: Yes.
6	THE COURT: You can head off early. Because you
7	started cross-examination, you can only speak with the
8	lawyers about what is for lunch. Don't talk about your
9	testimony.
10	THE WITNESS: Absolutely.
11	THE COURT: Okay. Thank you very much.
12	THE WITNESS: Thank you.
13	(Witness excused.)
14	THE COURT: I just want to quickly see the
15	lawyers. Then we'll break for lunch.
16	How long do you all want? Is 30 minutes good or
17	do you need more?
18	MR. BLACK: Half-an-hour would be great.
19	THE COURT: Are you sure? Long enough? That
20	will be enough?
21	MR. BLACK: I will take more if you give it to
22	me.
23	THE COURT: I prefer to go 30, but if you want
24	to go
25	MR. BLACK: I would like to go a little longer.

1	This is obviously important cross.
2	THE COURT: Okay. So do you want to, like, ten
3	of?
4	MR. BLACK: That would be great. Thank you,
5	Your Honor.
6	THE COURT: Are you all good with that?
7	MR. HALES: We're fine with that.
8	THE COURT: All right. Just a quick point. I
9	just want to make sure. I see people were fighting over it
10	yesterday. I think you are all on the same page. You all
11	agree initial is the same as release. Fair?
12	MR. BLACK: Yes.
13	THE COURT: And that the post-filtration it
14	sounds like is really the end of the on-process. Right? Do
15	you all agree on that?
16	MR. BLACK: Yes.
17	THE COURT: You don't have to fight over that
18	anymore?
19	MR. BLACK: Yes.
20	THE COURT: You were fighting over it yesterday.
21	MR. BLACK: I didn't think we were fighting over
22	it.
23	THE COURT: I think the testimony will bear it
24	out.
25	MR. HALES: It might be it was the end of the

1 in-process, the start of the release. 2 So then that brings up -- so that's THE COURT: 3 what remains then. I thought there has to be, and I think the last line of questioning showed there's some distinction 4 5 between in-processing and release because they don't match up there. Right? And so I thought post-filtration would 6 7 represent the end of the processing for purposes of 8 measuring pHs. 9 MR. BLACK: My understanding is that -- yes. 10 The issue -- I think the confusion is they do an in-process 11 test. At the end of the process, they're going to have to 12 meet 3.54. 13 THE COURT: All right. 14 MR. BLACK: They do a release test, which is 15 based on vials taken not exactly the same time, same day, and then they release the product. And the product -- the 16 17 release is actually a -- they sign a piece of paper that has 18 a release date on it. 19 Right. And it has to be no more THE COURT: than 3.6 at that point. 20 21 MR. BLACK: Right. And the in-processing --22 THE COURT: 23 in-processing, is that the right one? 24 MR. BLACK: In-process.

THE COURT:

In-process there's really two

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Park - cross measurements, but the one that counts is the post filtration? Is that right? MS. WACKER: They all count. There's one that is measured that's still the same. You take out the It's in all of the vials and some of the vials are used for the post-filtration test, but it's from the vials. Some of them are used for --THE COURT: But what's represented to the FDA --MR. BLACK: Post filtration. THE COURT: I thought there were times when somebody was just saying the in-processing rate or spec and they don't say whether it's post filtration. I think we've mostly been talking MR. BLACK: about the post filtration except in Dr. Park's testimony when he went into a little more granular level of what they are doing with the tank. And they actually take multiple readings that aren't reported to the FDA while they're fiddling and trying to get the pH right. Then they take a pre-filtration test. They lock it down. Then they filter it to get out any microbes. They do a post-filtration test before they fill. They put that value, that has to go to the forms, that goes

23 24 to the FDA. Then they take vials that are filled and do a 25 release test.

1 THE COURT: I think everybody agrees --2 MS. WACKER: That was not accurate. 3 Post-filtration is taken from vials. I think you misspoke. Right. 4 5 THE COURT: But maybe I'm just misremembering 6 things, which happens a lot. Is there ever a time where 7 somebody said to the FDA, here's the in-processing pH spec? 8 MS. WACKER: Yes. It is part of the proposed 9 batch records. 10 THE COURT: And when they do that, are they 11 reporting the post-filtration spec? What are they reporting 12 at that point? 13 MS. WACKER: All of it, the pH adjustment. 14 THE COURT: That sounds like three numbers as 15 opposed to one number. They report all of the different --16 MS. WACKER: 17 THE COURT: So they never just report an 18 in-processing pH level. You're saying they always report 19 the three components -- I'm sorry, three separate readings. 20 What are you telling me? Get some clarity. 21 MS. WACKER: I just want to make sure we're on 22 The proposed commercial batch record is the same page. 23 something that went into the FDA that said here are the 24 specifications we're using for each of those steps. Does 25 that make sense? But for the factors that are going out,

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Park - cross

like if we were proving going out in the future, I don't believe that those in-process, like the adjustment, all of that would be reported to the FDA. I got that from the video testimony. MS. WACKER: Yes. THE COURT: I got what would ultimately. I'm talking about though what's being reported now and what is being represented will be met. MS. WACKER: What is being represented will be met because what's in that proposed commercial batch record that has an adjustment pH, stabilization pH, in-process of --It has both and they separate them THE COURT: out. MS. WACKER: Correct. THE COURT: They don't conflate -pre-filtration, post-filtration. MS. WACKER: Correct. All right. But we can at the very THE COURT: least equate initial with release. Nobody is going to dispute that. MS. WACKER: Yes. We'll come back THE COURT: All right. Good. at 12:50. Thank you. (Luncheon recess taken.)

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2 (Afternoon session, 12:56 p.m.

THE COURT: All right. Please be seated.

Just because of scheduling, I'm five minutes

late. I have a call with a Third Circuit committee at

8:00 o'clock tomorrow morning. I will not be able to break

it off if it doesn't end at 8:30. It hopefully will. If

you all could be ready to go at 8:30. Thank you.

Ready, Mr. Black?

MR. BLACK: Okay. Thank you, Your Honor. We got the Elmo working.

THE COURT: Oh, good.

MR. BLACK: So I've been warned by my team not to use a red Sharpie on the Elmo like I did in Federal Court in Texas and I almost ended up in jail. I will be more careful here now.

BY MR. BLACK:

Q. And I have this slide here now which is what we would have done had the Elmo been working, which summarizes your testimony from before about batches SVA7 through 13.

What the slide shows, I've highlighted in green the values for stability, which were above the release values. I've circled in each case the highest value for instability and then I put on the right-hand side the difference between the initial and stability values at the

highest number on the right-hand column. Does that look accurate to you?

A. I think so.

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Q. Thank you. Okay.

So you had a slide with a ruler in it and we've made up a similar slide.

All right. Now, just before the break, Judge Connolly asked a couple of questions to clarify what is in-process release and the different numbers that have been discussed because the process of making the product is a little complicated and there are a lot of places where pH readings are made. And I just want you to clarify that the in-process specification, the final in-process specification number is taken at post-filtration; correct?

- A. Correct.
- Q. And in the process optimization confirmation batches, and this is your slide 30, the specification that they used was 3.42 to 3.50; correct?
- 19 A. Correct. **50**.
- Q. 50 for the process optimization batches. The top end number is 3.50; correct?
- 22 A. 3.50?
- 23 Q. Yes. Process optimization.
- 24 | A. I'm sorry, yes. I was looking -- yes. 3.50, yes.
- 25 Q. And those are the batches regarding which we were

1 using to calculate the data; correct? Let me rephrase.

- A. I'm sorry.
- 3 | Q. Some of the data that you've been showing the Court
- 4 comes from process optimization and confirmation batches; is
- 5 | that correct?
- 6 | A. Yes.

- 7 Q. And when those batches were made, the in-process
- 8 specification that was used was 3.42 to 3.50?
- 9 A. Yes.
- 10 \parallel Q. On the right we have yellow. This is your slide, the
- 11 intended commercial production batches. And there, the
- 12 | specification has been broadened slightly to be 3.42 to
- 13 3.54; is that correct?
- 14 A. Yes.
- when they make a whole bunch of commercial batches to make
- sure that if something goes wrong, they've got a little bit
- 18 of extra room on the pH; is that correct?
- 19 A. I don't know what their thought was. The fact that
- 20 **3.54** is way less than 3.64.
- 21 Q. Okay.
- 22 A. And as Eagle's documents show, their goal was to meet
- 23 | the pH target around 3.50. So this is well within the --
- 25 think we can agree that the actual commercial batches, 3.54

is the number that they are telling the FDA they want approval for; right?

A. Yes.

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- Q. Okay. And they are also still telling the FDA they
 want approval for release of 3.64; is that correct?
 - A. I'm not sure what approval you're looking at.
- Q. Have you ever seen Eagle propose a release number that was less than 3.6 rounded, i.e., 3.64?
- A. Eagle's release specification is 3.4 to 3.6, so 3.6 may include 3.64, but, again, going back to Dr. Aungst --
 - Q. Thank you. Excuse me, sir. I really hate to cut you off. It's very rude, but I would appreciate it if you would answer the question that I asked. The release specification in the current ANDA that the Court has to evaluate is 3.6 rounded, i.e., 3.64; is that correct?
 - A. 3.64, yes.
 - Q. So Eagle is asking the FDA to approve a product that when they are making it in the vats and when they pull the vial out to do the in-process spec can be as high as 3.54, but by the time they release, it can be as high as 3.64; is that correct?
- 22 A. I think so.
- Q. Right. Now, Eagle had the option, right, of saying to the FDA, you know, we don't need all of that extra room and changing their release spec to the 3.54; right?

Park - cross 1 Α. I don't know whether they had an option or not. 2 But if they did that, we'd have another case here; 3 Because there would be a very significant difference right? between 3.54 and the infringing range; correct? 4 5 Well, I don't know -- frankly, I don't know what you are talking about. But 3.54 is well within the 6 7 specification, so I don't know what is the issue here. 8 And if they reduced the pH to 3.54 release, they dramatically reduce the chance that any batch would ever 9 10 reach 3.65; is that correct? 11 Α. Again, I don't know whether it's reduce the chance or 12 Specification right now is well within the spec, so I 13 don't know what your concern is. 14 Isn't Eagle taking a bit of a gamble here by asking

- Q. Isn't Eagle taking a bit of a gamble here by asking the FDA for permission to release products that could be as high as 3.64 when the patents at issue in this case, which they've known about for three years, have a pH of 3.7 in the range?
- A. Again, I don't know whether Eagle is taking a gamble or not. Clearly, 3.64 is a range, but not 3.7.
- Q. Okay. Thank you.

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All right. So let's talk about the commercial batches. And I've got a ruler here a little bit like the one that you put up before. We've got three items on this, on this chart.

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Yes.

Park - cross

First we've got the in-process spec. Now, for the commercial batch, so the top end in-process spec is 3.54; is that correct? Α. Yes. All right. I will color that in. Now, we've seen the data just before lunch from SVA11 that showed that the Eagle product can go from the in-process spec to a release spec and we saw an actual batch with a different, an increase of .07 in those values; is that correct? Can you remind me again exactly what you're referring Α. to? The SVA11 data. All right. The SVA11 data Sure. showed that the -- we don't have the -- sorry. We need to get the -- SVA11 had a pH of 3.50 at the in-process spec and then it had six different readings at release, the highest of which was 3.57. Do you recall that? Α. Yes. Okay. Thank you. Q. All right. So if we go from in-process with the data from that batch up seven tenths, 3.54 plus .07 is 3.61; is that correct? If you add --Α. Is that correct? Q.

I'm going to answer. If you added to 3.54,

1 0.07, yes.

- Q. And we also saw upward drift during the shelf life on the order of .05 and .06 in multiple batches of the optimized product; is that correct?
- 5 A. Yes. Around 3.50.
- Q. Yes. No, that's not my question. My question was
 here we showed it from release through product life. We saw
 increases of .05, .04, .04, .06, .04, .05 in the data that
 you say is representative of the batch between release and
 shelf life; correct?
- 11 | A. Yes.
- 12 Q. So if we were to add to the 3.61, .05 or .06, we'd be
 13 in the infringing range, wouldn't we?
- 14 A. The highest --
- Q. Yes or no, sir? Yes or no, sir? Would that value be within the infringing range?
- 17 A. The value of the infringing range --
- 18 Q. Thank you.
- 19 A. -- but your calculation is wrong.
- Q. Eagle is maintaining the right to sell a product that releases while in commercial manufacture at a pH of 3.64; is that correct?
- 23 A. Again, I) don't know what -- what right they have.
- Q. Okay. Let's talk a little bit about the refrigerated data and, again, the question of optimization. This is your

slide 19, a little blurry, but for my purposes, it won't

matter that much. And this shows, I believe you said, the

pH behavior of the registration and characterization batches

at the top; is that correct?

A. Yes.

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- Q. And the pH behavior of the optimization and PPQ batches at the bottom; is that correct?
- 8 A. Yes.
- 9 Q. And you kind of crunched the scale a little bit, so
 10 it's hard to see, you know, exactly what's going on here,
 11 but you testified that this is -- these pH profiles are
 12 different, these are different behavior; is that correct?
- 13 A. Yes, I said that.
 - Q. And so you have two batches with the identical component made by the same people at the same plant under the same ANDA and you have very different pH profiles, don't you?
 - A. That's not true.
 - Q. Well, are the pH profiles different?
- 20 A. Usually not changing in manufacturing process.
- 21 Q. Okay. Go ahead. So you have two different pH
 22 profiles that have the same formulation, the same
 23 manufacturer, a different manufacturing process; right?
- 24 A. Correct.
- 25 Q. A little bit like giving the ingredients to a cook and

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seeing what you get. You might give the same ingredients to me and my wife, but the omelet would come out a lot differently. Fair analogy? I don't understand your point. Okay. I will move on. The point is that the pH properties of the formulation, an identical formulation can be very different based on the characteristics of the manufacturing process; correct? Again, as I explained, stirring is much longer for optimization process compared to registration batches. longer stirring make mixing more homogeneous. That's why Eagle has such a narrow pH reading. But all I'm saying is the manufacturing process in your view has significantly contributed to changing the pH profile of the Eagle product; correct? Α. Yes. Okay. And have you examined the manufacturing process for Vasostrict? Which Vasostrict? Α. Either one? Q. Α. You mean API? You just testified that the manufacturing process can make a big difference in the pH profile of identical products by the same manufacturer in the same plant and I'm

just wondering if you have looked at Par's manufacturing

Park - cross

process to compare it to the manufacturing process used by

Eagle, because I didn't see any of that in your report.

- A. I'm so sorry. I don't understand your question.
- Q. Have you reviewed Par's manufacturing process? I

 didn't see anything in your report about it. I didn't hear

 any testimony about it.
- A. No, no. I explained that during my direct, I went through the process for the ANDA.
 - Q. Have you made a detailed analysis and provided a report on Par's manufacturing process --
- 11 A. Oh, Par's?

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12 Q. Par's, yes.

correct?

- 13 A. I'm sorry. No. Par. No, I don't.
- Q. So therefore since the manufacturing process can
 affect the pH profile, even by the same manufacturer, i.e.,
 Eagle, it follows that the manufacturing process of Par
 could affect the pH profile of Par's product; is that
- 19 A. If they change, yes, it could.
 - Q. And therefore it would be wrong to say that the products are identical simply because they're generic because the pH profile, which is what is important to this case, is dependent on the manufacturing process, not just the ingredient in the product; is that right?
- 25 A. I'm not sure what you are comparing, but generic and

RLD compare the active ingredient, excipient and others and whether it's safe or not based on ingredients and excipients. So other than that, I don't know what you are talking about.

- Q. The FDA doesn't go through and compare the manufacturing processes of Eagle and Par. In fact, those are closely held secrets; is that correct?
- 8 A. That I don't know.
- 9 Q. Okay. There are other differences between original
 10 Vasostrict and the Eagle product, including the impurity
 11 profile; correct?
- 12 A. Impurity profile depends on the API.
- 13 Q. Right?

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- A. So API -- unless getting the API from the same source, impurity profile may be different.
 - Q. And you're aware that the source for Eagle's API is different from the source for Par's API?
- 18 A. I don't know, but maybe.
- Q. And if they were different, then you would see different impurities in the formulation and different amounts of impurities; right?
- 22 A. Right.
- 23 Q. And the FDA goes through and they decide whether or
 24 not the differences are relevant to safety and
 25 effectiveness; is that correct?

1 A. Again, what FDA does, I don't know.

2 Q. Okay. Are you aware that Eagle is sending data to the

3 | FDA relating to impurities and other matters relating to

- 4 original Vasostrict?
- 5 A. That I don't know. I don't remember.
- 6 Q. I'd like you to take a look at one of the documents
- 7 that's in your binder, DTX-133. I'm not going to switch
- 8 because I don't want to lose my Elmo.
- 9 A. DX-133?
- 10 Q. Yes.
- 11 A. Yes.
- 13 communications between Eagle and the FDA and consistent with
- 14 | the Court's instructions, we are not going to try to admit
- 15 | the entire document, but there are two pieces of it I would
- 16 like to refer you to.
- 17 A. Yes.
- 18 Q. So if you take a look at page 36, this is an e-mail or
- 19 letter, it's not clear, from Adrian Hefner, Chief Medical
- 20 | Officer at Eagle, to Charmaine Lintildus, Regulatory
- 21 Business Process Manager at FDA.
- 22 Do you see that?
- 23 **A.** Yes.
- 24 THE COURT: I'm sorry. I'm lost. What page is
- 25 | it?

Park - cross 1 MR. BLACK: Oh, it's DTX-133.0036. 2 THE COURT: Okay. Thank you. 3 MR. BLACK: I will put it up on the Elmo. BY MR. BLACK: 4 And if you look to the fifth paragraph, it says, Eagle 5 would like to know if it is acceptable to the FDA to perform 6 7 the comparative analysis of the generic product versus the current product RLD in the market which differs slightly in 8 9 formulation provided in the table below and in the table, 10 the image of which was cut off here, your exhibit. 11 Do you see that? 12 Α. Yes. 13 That was a request by Eagle to perform their testing 14 for FDA approval against the reformulated Vasostrict because they didn't have original Vasostrict vials to work with 15 16 anymore; is that correct? 17 That I don't know. 18 Okay. And then if you go back two pages to page 34, 19 DTX-133-0034, we have the FDA's response at the bottom of 20 the page in the block quote, and it says, "It is recommended 21 that you conduct the comparative studies requested in the CRL Section B, questions 1, 2 and 3, using samples of both 22 23 presenting of the current RLD formulation, i.e., both the 24 one milliliter single dose and ten-milliliter multiple dose

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presentation."

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Park - cross

And this letter is dated September 26, 2018. the current RLD would be reformulated Vasostrict; is that correct? What is the current -- current RLD formulation, both ml and ten ml presentation. The current RLD as of the date of this communication, September 26, 2018, was reformulated Vasostrict; is that correct? As I read again, it is recommended that you conduct a comparative study because original Vasostrict was not available anymore, so it used a current Vasostrict. Yes. 0. Yes, that's what it says. And you have not seen any evidence that Eagle refused Ο. to cooperate with the FDA and do the testing based on the current RLD, reformulated Vasostrict? I have no information, but I guess not. Okay. Let's talk about your validity opinion. You agree, do you not, that the person of ordinary skill is someone who has some formulation background in peptides; is that correct? Α. Yes. So let's imagine such a person sitting in their office in the bowels of a pharmaceutical company and gets a phone call from the bosses upstairs and the researcher is told,

Park - cross

we're going to try to make a version of Vasopressin product.

2 Can you get to work gathering some information? Imagine

- 3 this person is a POSA. Okay?
 - A. Yes.

- 5 Q. The first place the POSA would probably look, one of
- 6 the first places, would be the USP to see what the basic
- 7 | information is available about the drug; is that right?
- 8 A. Before that, a POSA would look into RLD if they want
- 9 to make a copy.
- 10 | Q. I was thinking that would be second, but first.
- 11 Sounds good. First you start with the RLD and what's
- 12 available is the label; is that correct?
- 13 A. Yes.
- 14 \ \Q. On the priority date, that label would have said in
- 15 | the label for original Vasostrict, it would have said pH 3.4
- 16 to 3.6; is that correct?
- 17 A. Label, yes.
- 18 Q. Yes. And the USP monograph would have said 2.5 to
- 19 **4.5**; right?
- 20 A. **Yes**.
- 21 Q. And the researcher probably would have looked further
- 22 | to see what other information the FDA had published on the
- 23 drug; right?
- 24 A. Yes, I would.
- 25 Q. And they would have looked for the biopharmaceutics

Park - cross

review and chemistry review which said, make the product at 3.4 to 3.6, because if you go outside that range, you're going to have poor stability; right?

- A. They would, but again as we discussed, it doesn't really mean that much without data.
- Q. That is all that the POSA would have. Right? The POSA wouldn't have the underlying data, but they would have the FDA's conclusion about the data. Is that fair?
- A. That's not the only data. We know looking at the patent, other product, they all show the pH values around 3.7 and 3.8, so that's not the only data they see.
- Q. Your reaction was the POSA would first go to the approved FDA product. Do you really think that the first thing a POSA would look at would be a Lithuanian patent for Vasopressin derived from the pituitary gland of pigs?
- A. I didn't say first. It is part of the nature of the search. It's all about the literature search, prior art search in this case.
- Q. Right. And the literature search would show a 2.5 to 4.5 range and various values within that range you would find Bi and Singh that talks about 3.4, 3.5 and other references; correct?
- A. Yes.

Q. Now, you provided some testimony and we had some back and forth with the lawyers about the products that were on

the market at the time and the only Vasopressin product on
the market on the priority date would have been reformulated
Vasostrict; right?

A. I'm sorry. Product on the market?

Q. Yes. The priority date here is February of 2017; is

A. Yes.

that correct?

- Q. And you have to evaluate a POSA's knowledge and inclination as of February 2017 in doing your obviousness analysis; is that correct?
- 11 A. Correct.
 - Q. And at that point, the only vials that someone would have been able to get of Vasopressin product would have been the reformulated Vasostrict; is that correct?
- 15 A. That I don't know.
 - Q. Okay. Certainly, the vials that were manufactured in 2015 that you mentioned before, original Vasostrict, they would have been unavailable by then; is that correct?
 - A. That I don't know.
 - Q. Okay. If a POSA did manage to get ahold of some original Vasostrict and test it, you would have to, if you are going to consider that in your analysis, it would be fair to consider all of the vials that were manufactured, the hundreds of thousands of vials, not just the single point value that you hand picked for your obviousness

Park - cross

1 analysis. Is that fair?

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- A. No, I don't think so. No.
- Q. You think it's fair for you to pick one, one record from an internal document at Par that a POSA would have had and include that in your obviousness analysis?
- A. If I have a problem, so I need a Vasopressin product,

 I will take the one approved by FDA. One vial representing

 the whole batch. So that particular batch is a

 representation, a representation of the batch that pH was

 3.7.
 - Q. But if you went to the market and picked one vial to test, you would have no basis to assert that that vial, that specific vial that a POSA would test would have been outside the label claim of 3.4 to 3.6, do you?
 - A. But that's what happened, so it would happen.
 - Q. You are misunderstanding my question. There are millions of vials of Vasopressin sold; is that correct?
- 18 A. I don't know how many of them.
- 20 Q. You put the sales records into the record, hundreds of thousands or millions of vials are sold; is that correct?
- A. Okay. A hundred-thousand I understand, but I don't know millions.
 - Q. Okay. Hundreds of thousands of vials are sold. Are you assuming that the POSA would know the pH of all of those hundreds of thousands of vials or only the 3.7 test

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result of the product that you saw in that one document from 2 Par?

- A. Of course, I wouldn't know because they did not test the whole thousands of bottles, but one vial, 3.7.
- Q. I'm asking for your obviousness analysis, are you assuming that that POSA is starting with the knowledge of that vial of 3.7 or they are starting with the knowledge of all of the vials of Vasopressin, the hundreds of thousands that were sold and all of their pH values?
- A. Again, we wouldn't know all the vials or not because nobody tested, but one vial tested was 3.7.
- 2. But for the obviousness analysis, the POSA wouldn't know whether -- POSA wouldn't have that information, would they, if it came from an internal Par document?
- A. No. Again, the POSA could measure that particular vial.
 - Q. By the way, that 3.7 value was taken nine months before the product was sold; correct?
- 19 A. I look at the record. Maybe, yes.
- Q. Not maybe yes. The answer is yes, isn't it? Do you know that for a fact?
- A. Again, I don't recall the exact number, but that may be the case.
- Q. And that could have been a value of 3.65 rounded; is that right? Single vial?

Park - cross

1 Α. That I don't know whether the value was rounded or 2 not. 3 That value came from a much longer document that had 0. 15 or so batches in it; correct? 4 5 Α. I don't recall how many vials. 6 Q. Not the vials. There was a batch that you relied 7 on --8 Α. Yes. 9 -- and there were about 15 batches that were in the 10 same document that all had pH values listed. 11 Do you remember that? 12 I don't remember exactly which document I'm sorry. Α. 13 you're talking about. 14 Okay. Do you agree with the statement that Dr. Chyall made in his report in his deposition that Bi and Singh 15 16 determine the stability of Vasopressin solutions under 17 various pH environments and found favorable stability at a 18 pH of around 3.5? Do you agree? 19 I'm sorry. Could you show the document or can I go to Α. 20 the --21 It's a quick -- we can do it if you need to, but I'm just asking whether you agree with the statement. I'm just 22 23 telling you to be fair, it came from Dr. Chyall's report. 24 Bi and Singh determined the stability of Vasopressin 25 solution under various pH environments and found favorable

Park - cross 1 stability at a pH of around 3.5. 2 Do you agree with that? 3 That's his statement. Α. 4 I'm saying do you agree with it as someone that 5 reviewed that reference? I don't have a particular reason not to agree, but the 6 7 vasopressin is stable. Again, without the data, I cannot Stable for what? 8 say. 9 Do you agree with his statement that pH for 10 Vasopressin formulations had already been optimized at the 11 priority date? 12 No, I don't know what optimization he's talking about. 13 He testified that -- I'm going to ask you whether you 14 agree with this. In your opinion, the pH of Vasopressin 15 formulations had already been optimized as of the priority 16 date of the patents-in-suit; correct? His answer was, at 17 least by that time, yes. 18 Do you agree with that or disagree? 19 No, I don't agree. Α. 20 Q. Okay. 21 MR. BLACK: Pass the witness. 22 All right. THE COURT: Thank you. 23 redirect? 24 MS. WACKER: No. I just have to move exhibits.

THE COURT:

Okay. Go ahead. See what Mr. Black

Park - cross

says.

MS. WACKER: I have a really long list. We also were going to admit all of the citations on that summary document as we discussed at the pretrial conference.

I was wondering if maybe it makes sense to confer with Mr. Black during the next break and we could give a written list up. That might be easier than reading 100 numbers. I don't think it's a hundred, but --

THE COURT: Sounds like a good idea.

MR. BLACK: I concur. I have a couple that I want to move in, including my horrible handwriting on this, which he did verify was accurate during his examination, and part of the FDA correspondence, because I know you don't want everything.

THE COURT: On that, again, I'm struggling. I told you, and I really appreciate the input of you all, especially that have practiced in Texas and other places.

Part of me is like let it in, but then what do we do about when it gets to the Federal Circuit and somebody decides to highlight, oh, we're making a big argument that I never even heard of. And it does happen. I am not saying you all would do it.

MR. BLACK: I'm not sure -- I don't understand how it would be proper for them to consider an argument that hadn't been made below about a critical document. If they

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Park - cross

do, if that is happening in cases, I have not seen it. sure it has happened, but I don't see there's any way to really stop it. THE COURT: Well, I mean, I had a 101 case where a new appellate lawyer, you know, sounded like from a different universe. MR. BLACK: Yes. THE COURT: And the person who argued it in front of me said -- and I'm just going by what my colleagues say, and I have to rely on what my colleagues say. Judge Robinson was the first to institute the rule, but everybody else has followed. But maybe I will just let it go. I mean, there is something to be said for having the entire document admitted. Right? MR. BLACK: Yes. THE COURT: It's in context if a question arises. Why don't we do this. We'll work on MR. BLACK: this outside of the courtroom, but we'll give you a list. If there are issues where either party thinks it's likely to be overbroad to have a big document in, we can talk about it. If there's an issue, we can show Your Honor. THE COURT: Okay. That sounds great. All

right. And you're excused. Thank you very much.

1 (Witness excused.) 2 THE COURT: Is Dr. Park going to be asked to 3 come back? 4 MS. WACKER: He is not. 5 THE COURT: Okay. So this concludes our evidence on 6 MS. WACKER: 7 noninfringement, so I renew my Rule 52 motion. 8 THE COURT: I will reserve judgment on that. 9 And then what's next? 10 MS. WACKER: We have Dr. Carmen Cross testifying 11 next. 12 All right. THE COURT: MR. KWON: Your Honor, my name is Sam Kwon 13 14 and I am counsel for Eagle. I will be cross-examining Dr. Cross. 15 THE COURT: All right. 16 Thank you. 17 Do you all want to move whatever books? 18 Somebody probably wants to clear them off. 19 ... CARMEN CROSS, having been duly 20 sworn/affirmed as a witness, was examined and testified as 21 follows... THE WITNESS: Good afternoon. 22 23 THE COURT: All right. Mr. Kwon? 24 MR. KWON: Thank you, Your Honor. 25 DIRECT EXAMINATION

Cross - direct

1 BY MR. KWON:

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Q. Good afternoon, Dr. Cross.

medicine and trauma critical care.

- 3 A. Good afternoon.
- 4 Q. Could you please introduce yourself to the Court.
- A. My name is Carmen Anthony Cross. I'm a medical doctor
 with 42 years of experience in the specialty of emergency
- 8 | Q. Dr. Cross, what will you be testifying about today?
- 9 A. Are we using slides?
- Q. Oh, yes, we are. Let's move to slide 2. Dr. Cross, what will you be testifying about today?
 - A. I will be testifying about the clinical and historical use of Vasopressin in clinical medicine as well as my review of the patents-in-suit and other clinically pertinent information I was given.
 - Q. Let's turn to slide 3. Can you please provide an overview of your educational background?
 - A. Certainly. I received my bachelor's degree in biology from Syracuse University in 1975 and then went on to New York Medical College, where I obtained my doctor of medicine degree.
 - I left in '79 to attend Brown University

 Affiliated Hospital for the study of internal medicine and critical care, and then left in 1982 to pursue board certification in emergency medicine.

Cross - direct

Q. Slide 4, please. Can you please provide an overview of your educational background?

A. Certainly. After I left post-graduate training, I started working at an affiliated Albany Medical Center called Columbia Medical Hospital, and there for most of my tenure I served as the chairman of the department of trauma and emergency services as well as a practicing clinical and emergency medical physician.

I left in 2000 and started working for the

Mohawk Valley Hospital System at St. Elizabeth's Medical and

Trauma Center, and I work there to the present day. There

I'm a senior attending in emergency in trauma medicine as

well as having other administrative functions.

- Q. Dr. Cross, do you use Vasopressin in your course of work?
- 16 A. Yes, I do.

- Q. How long have you used it for?
- A. I've used it for all of the 42 years I've been practicing medicine.
 - Q. What do you use Vasopressin for?
 - A. Vasopressin has been used for a number of indications, but mostly as an agent that can increase blood pressure in people who are hypotensive and therefore in shock.
 - MR. KWON: Your Honor, at this time defendants proffer Dr. Cross as an expert clinician, including in the

Cross - direct 1 specialties of emergency medicine and critical care. 2 MR. RHOAD: No objection, Your Honor. 3 THE COURT: All right. Thank you. MR. KWON: Now we're on slide 5. 4 5 BY MR. KWON: Doctor, here you have excerpts from DTX-178, DTX-246, 6 7 DTX-249 and DTX-36. What are these documents? These documents are product labels for Vasopressin. 8 9 You can see on the left-hand side of the screen, there are 10 three product labels of unapproved generic versions of the 11 drug and on the right, Par's original Vasostrict product 12 label. 13 Now, which of these products were available for use 14 during your career? All of them. 15 Α. 16 Do you recall using any of these products 17 specifically? Yes. All of them. 18 Α. 19 And what were these products used for? 20 Well, since the reference drug in these products was Vasopressin, these drugs were used as a method of increasing 21 blood pressure in shock states as well as other indications. 22 23 Now, at the present time, of the four products on the 24 screen, which product is available?

A. The present time, original Vasostrict.

Cross - direct 1 Q. Do you mean original Vasostrict or reformulated 2 Vasostrict? 3 Well, the reformulated Vasostrict. Α. Now, has the manner in which Vasopressin is used since 4 5 Par introduced Vasostrict to the market changed during your 6 practice? 7 Α. Not at all. 8 Now, I want you to take a look at Pitressin, which is 9 on the center of the left side of the slide and original 10 Vasostrict, which is on the right. How are these products 11 relate? 12 When Par submitted its NDA, it expressed a desire to 13 manufacture the generic Pitressin of JHP, which is the one 14 you see in the middle on the left. After it received its NDA, it then marketed the original Vasostrict as an FDA 15 16 approved drug. 17 Okay. Let's look at slide 6. Let's now look at Par's 18 submission to the FDA on Pitressin. 19 Here we have excerpts from pages 9 and 30 of 20 DTX-25. What is this document? 21 This is the pre-NDA submission letter authored by Par's predecessor, JHP, that was filed to the FDA in 2011, 22 23 expressing their desire to manufacture Pitressin. 24 And what did Par tell to the FDA about when Pitressin

25

became available?

A. They expressed what clinicians had long known, that
this was a 100-year-old drug that had been very well
established in the medical profession.

- Q. And what did Par tell the FDA about what Pitressin had been used for?
- A. They had noted that it had several indications, but most importantly, it's used as a vasodilatory shock pressure agent. In other words, in vasodilatory shock, it can raise blood pressure.
- Q. What did Par tell the FDA about the safety and efficacy of Pitressin?
- A. They told them that it was a safe and very well accepted medication.
 - Q. Now, are Par's representations that we just talked about to the FDA regarding use of Vasopressin consistent with how you have used Vasopressin in your practice?
- 17 A. Certainly.

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- 18 Q. Now, following this pre-NDA submission, did Par submit
 19 an NDA on Vasopressin?
- 20 A. They did.
- Q. Let's turn to slide 7. Doctor, here you have an excerpt from page 8 of DTX-42. What is this document?
- A. This refers to a clinical reference from Par's NDA for Pitressin.
- 25 Q. And has Par performed any clinical trials to support

Cross - direct

its NDA?

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- A. No, not at all.
- 3 Q. And what sort of evidence did Par rely on instead?
- A. They based all of their evidence solely on pre-existing literature or what you people call prior art.
 - Q. Let's now talk about the literature Par relied on to support its NDA. Let's look at slide 8.

Now, this slide shows excerpts from page 9 and page ten of DTX-42. What literature was in the clinical part of Par's NDA based on?

- A. Par cited a number of literature references, and particularly you could see on the screen a number of major investigators, and all of these investigators described the dosing that was effective for various states of shock, particularly post cardiotomy shock and septic shock.
- Q. Now, during your 40-year practice, were you familiar with any of these references in your course of work?
- A. Practically all of them.
 - Q. And were clinicians like yourself relying on these references regarding use of Vasopressin?
- 21 A. Certainly.
 - Q. Let's now look at the approved product label for original Vasostrict, slide 9.
 - Here we have excerpts from page four of DTX-132.

 What is this document?

Cross - direct

A. This is the original Vasostrict approved label.

- Q. And what was the indication that was approved for original Vasostrict?
- A. The original Vasostrict was indicated as a method of increasing blood pressure in patients with distributive or vasodilatory shock, specifically referring here to post-cardiotomy shock and septic shock. They also gave very specific dosage ranges that were drawn directly from pre-existing literature.
- 10 Q. And what are those dosages?

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- A. Post cardiotomy shock, it was recommended and still is recommended to start with a dose of 00.3 units per minute up to a high of 0.1 unit per minute, and for septic shock, a start dose of 0.01 unit per minute to a high of 0.07 units per minute, and all of these would be given by continuous IV infusion.
- Q. Now, is a manner in which original Vasostrict is used, including the dosage and indication, different from the unapproved Vasopressin product we just talked about earlier?
- A. Not at all.
- Q. Okay. Let's take a look at the '209 patent. We're on slide 10 now. What priority date did you apply in forming your opinion?
- 24 A. February 7th of 2017.
- 25 Q. And what aspects of the '209 patent did you consider

1 in connection with your opinion?

- Only the clinical elements of the patents. Α.
- 3 And to be clear for the record, what do you mean by the clinical element?
- The clinical elements are highlighted in this slide. 5
- That is a method of increasing blood pressure in a patient 6
- 7 who needed it. That would be a hypotensive patient. And
- always within a specific standard dosage range of anywhere 8
- 9 between 0.01 units per minute to a high of 0.1 unit per
- 10 minute by continuous IV infusion.
- 11 Q. Okay. Move on to slide 11. Doctor, did you hear Dr.
- 12 Park provide a definition of POSA?
- 13 I did. Α.

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- And what is that definition?
- The definition is defined adequately on the screen, 15 Α.
- 16 and this is the definition that I used throughout my
- 17 analysis of all of the data.
- 18 And how does your analysis in this case relate to this
- 19 definition?
- 20 Well, I would be a clinician with whom a POSA would
- 21 collaborate and work with.
- Let's now jump back to the original Vasostrict label. 22
- 23 We're moving to slide 12.
- 24 How does a description of use of the -- let me
- 25 start again. How does the description of use of original

Vasostrict as provided in the label compare to the clinical elements of the '209 patent?

- A. We could see that the original Vasostrict label teaches the '209 patent's clinical elements.
- Q. And more specifically, how did the indications and the dosages provided in the Vasostrict label compare to the clinical elements?
- A. On the left-hand side of the screen, you could see the indication for Vasostrict being used as a method to increase blood pressure for distributive or vasodilatory shock states and that means in patients who are hypotensive. They also give on the left-hand side specific dosage ranges as we had just mentioned that all fall within the standard widely accepted dosage range of anywhere between 0.01 units per minute to a ceiling of 0.1 units per minute if by continuous IV infusion.
- Q. Now, would a clinician that would work with a POSA rely on this label for guidance on how to use original Vasostrict?
- A. Definitely.

Q. Now take a look at slide 13. I would like to switch gears a little bit and now talk about the 2014 label for Vasostrict in connection with Eagle and Amneal's inequitable conduct case.

Now, let's start with the '239 patent. Do you

Cross - direct

understand how the '239 patent is related to the patents-in-suit?

- A. Yes. I was informed that the '239 patent is a parent patent of the patents-in-suit.
- Q. And this slide identifies excerpts from pages 13 of PTX-36. What is PTX-36?
- 7 A. PTX-36 is the April 2014 Vasostrict label.
- Q. And how does the description of use provided in the
 April 2014 Vasostrict label compare to the clinical elements
 of the '239 patent?
 - A. The April 2014 Vasostrict label again teaches each of the clinical elements of the '239 patent.
 - Q. More specifically, how do the indications, the dilution step, and the dosages as provided in the April 2014 Vasostrict label compare to the clinical elements of the '239 patent?
 - A. They are the same. On the left-hand side, once again, we see Vasostrict being used as a method to increase blood pressure in patients who were hypotensive. Also, the dilution steps that are noted in 2.1 are the same as the dilution steps of the '239 patent. We can see the range of 0.1 unit per ml as the low dilution and one unit per ml as the high dilution, the same as the '239 patent.

And on the bottom of the left side of the screen, once again, we see the specific dosage ranges

IV infusion.

Cross - direct

recommended for post-cardiotomy shock and septic shock all falling within the standardized dosage range noted in the '239 patent, and that is, once again, 0.01 unit per minute to the ceiling of 0.1 unit per minute by IV infusion.

Q. Okay. Now we're on slide 14. Now let's now turn back to the '209 patent.

How does the description of use as provided in the April 2014 Vasostrict label compare to the clinical elements of the '209 patent?

- A. The April 2014 Vasostrict label teaches each of the elements of the '209 patent.
- Q. And, more specifically, how do the indications and dosages as provided in the April 24th Vasostrict label compare to the clinical elements of the '209 patent?
- A. They are the same. On the left I repeat Vasostrict being used to increase blood pressure in patients with distributive shock who, of course, would be hypotensive.

 And, again, the recommended doses for both cardiotomy shock and for septic shock are listed with their low and high ranges, again all falling within the standardized dosage range widely accepted for the previous 45 years of 0.01 units per minute to the high of 0.1 unit per minute by
- Q. Thank you, Doctor. My colleague just pointed out that I may have asked a question without actually referring to

1 the claims of the '209 patent, so let me re-ask that.

A. Certainly.

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- Q. How does the description of use as provided in the
 2014 Vasostrict label in terms of indications and dosages
 compare to the clinical elements, clinical claim elements of
 the '209 patent that are provided on the right side of the
 screen?
 - A. Yes. The April 2014 Vasostrict label teaches each of the clinical elements of the '209 patent.

10 MR. KWON: No further questions.

THE COURT: Mr. Rhoad?

- 12 BY MR. RHOAD:
- 13 Q. Good afternoon, Dr. Cross. My name is Bob Rhoad.
- 14 Nice to meet you.
- 15 A. Good afternoon, Bob.
- 2. Jumping ahead of the binders. Let me get you some binders. We got your name wrong, too, apparently on the slide. We brought up the wrong person. There we go.
 - Now, Dr. Cross, based on your experience, if the FDA were to approve Eagle's ANDA, you would expect that clinicians like yourself would administer Eagle's product in accordance with the instructions provided on the package insert; right?
- 24 A. Yes, I would.
- 25 Q. And in particular, you would expect that clinicians

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Cross - direct

would follow the instructions with respect to the dosage and administration portions of the package insert; right? Α. Yes. And you agree that drug products can be used at any time during their approved shelf life so long as they've been stored properly; right? Α. I do. And do you understand that Vasostrict is typically stored in refrigerated conditions? I understand so. Α. And you would expect that Eagle's product, if it were to be approved, would also be stored in refrigerated conditions? MR. KWON: Objection, Your Honor. This is outside of his report. I'm not allowed to ask questions MR. RHOAD: outside of his report? This is outside the scope of the MR. KWON: The direct was related to obviousness of the direct. clinical claim elements of the patent. Now, Par is asking questions directed to infringement. All right. THE COURT: Hold on. My computer I was actually trying to figure out what's going on. froze. I need to pull up the live transcript. MR. RHOAD: I will save us time, Your Honor. I

Cross - direct

1 will withdraw the question.

infringement report; right?

2 THE COURT: I still want to pull this up, but go

- 3 ahead.
- BY MR. RHOAD: 4
- 5 Now, you actually prepared and served two expert reports in this case; right? 6
- 7 Α. Yes.
- 8 And in one of them was a rebuttal to Dr. Coralic's
- 10 Yes. Α.

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- And although in that report you provided rebuttal to some of Dr. Coralic's opinions, you didn't rebut any of his opinions in the report about the infringement of the '209 patent?
- 15 That's correct. Α.
- 16 Now, and you're not expressing any opinions here today 17 about the ultimate validity of any patent claims; right?
- 18 Α. With respect to clinical patent claims?
- 19 The ultimate obviousness of the patent claim as a Q. 20 whole, you're not expressing those opinions, are you?
- You're only talking about clinical limitations; right?
- 22 Yes, that's correct. Α.
- 23 And you understand that the claims also include 24 formulation related limitations like pH and impurities; 25 right?

Cross - direct

A. I do, sir.

- Q. And you are not expressing any opinions about those?
- 3 A. No, not at all.
- 4 | Q. So you are not expressing opinions about the ultimate
- 5 | validity of the patent claims. You're just explaining what
- 6 you believe was disclosed with respect to some of the
- 7 | limitations?
- 8 A. That's correct.
- 9 | Q. And, in fact, you would lack the experience and
- 10 expertise to talk about the formulation-related elements
- 11 | like pH and impurities; right?
- 12 A. Yes.
- 13 Q. And now you testified about that JHP and Par didn't
- 14 conduct any clinical studies relating to the NDA, in support
- of the NDA that they filed for Vasostrict.
- Do you recall that testimony?
- 17 A. Yes, with respect to the clinical elements.
- 18 | Q. Right. And so you're certainly not suggesting that
- 19 Par or JHP didn't do a lot of work characterizing the
- 20 | impurities and studying impurities and doing stability
- 21 studies on a product and things like that; right?
- 22 A. No.
- 23 \ Q. Now, I just want to make sure we're all on the same
- 24 | page as to exactly what the clinical limitations you're
- 25 talking about are, so I've pulled up PDX-5-2.

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Cross - direct

So this is claim 1 of the '209 patent and we've highlighted what you are referring to as the clinical limitations; correct? Correct. Α. And the ones that aren't highlighted, those are the formulation related limitations you are not talking about? Α. That's true. The '239 patent, here again, this is claim 1 of the '239 patent, and we've highlighted in yellow what you're referring to as the clinical limitations that you are testifying about and they're highlighted in yellow; right? That's true. And the ones that aren't -- the limitations that are not highlighted here are the formulation related limitations of the '239 patent that you are not testifying about; right? Also true. All right. I would like to now pull up paragraph 39 of your report, which is shown here on PDX-5-4, and by the way, the '239 slide was PDX-5-3. So this is paragraph 39 of your report. this is where you give a list of all of the clinical limitations that you identified in your report and that you expressed opinions about; right?

A. Yes. The common clinical elements, yes.

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prior art; right?

0. And so when you in your report are talking about, quote unquote, the clinical limitations, you're talking about this list of limitations that you provided in paragraph 39 of your report; is that correct? Yes. Α. And your testimony is that clinicians used Vasopressin products in accordance with all of these clinical limitations before Par filed for its patents. Is that your testimony? Yes. Α. And your opinion is that all of these clinical limitations were taught in the scientific literature that was published before Par's patent. Is that your testimony? That's also true, sir. Α. And your belief is that each of these clinical limitations was taught by the April 2014 Vasostrict -strike that. Your testimony -- your opinion is that each of the clinical instructions that are provided on the April 2014 label were taken directly from earlier prior art; is that right? Α. That's true. And your testimony is that each of the clinical limitations that are taught by the April 2014 label merely recite what you say was long known and practiced in the

A. True.

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- Q. Now, during the course of your work in this case, you reviewed the prosecution histories from the '239 and '209 patents; is that correct?
- A. I did.
- Q. And your review revealed that the Examiner agreed with you about the testimony you've provided here, that all of the clinical limitations of these patents were taught by the prior art; right?
- 10 A. Yes.
- 11 Q. And those findings, Par never attempted to rebut those 12 findings by the Examiner; is that correct?
- 13 A. That's correct.
 - Q. And the Examiner allowed the patents to issue not because of any differences between the claims and the prior art relating to the clinical limitations, but based instead on other arguments; right?
- 18 A. That is what I understand.
- Q. And you agree with the Examiner about what the prior art disclosed about those clinical limitations; right?
- 21 A. I do.
- Q. So with that in mind, let's go back to claim 1 of the '209 patent.
 - So the Patent Examiner found that each of the clinical limitations highlighted here on PDX-5-2 from

Cross - direct

claim 1 of the '209 patent was disclosed in the prior art;

- 2 right?
- 3 **A. Yes.**
- 4 Q. And, in particular, she found that they were disclosed
- 5 by Tanja Treschan and Russell references; right?
- 6 A. Yes. Tanja Treschan and James Russell for sure.
- 7 \ Q. So please take a look at PTX-238 in your binder.
- 8 A. PTX-238, yes.
- 9 | Q. That's a copy of the Treschan reference; right?
- 10 A. This is one of Tanja Treschan's articles, yes.
- 11 \ \Q. This is the one that you relied upon in your report
- as, this is the one that was cited by the Examiner during
- 13 | the prosecution of the patents?
- 14 A. Yes.
- 15 Q. The '209 and '239 patents?
- 16 A. Yes. This is the article I believe that the Examiner
- 17 cited.
- 18 \ \Q. And the Treschan and Russell references were prominent
- 19 references in the field as it comes to the -- whether it
- 20 comes to the use of Vasopressin; right?
- 21 A. Certainly.
- 22 \ Q. All right. And I think I may have misspoke. Am I
- 23 correct that the Treschan reference that we just looked at
- 24 was DTX-238, not PTX-238?
- 25 A. It's "D," David.

Cross - direct

Q. Thank you. And I apologize for that misspeaking.

2 Now, and so the Examiner allowed the '209 patent 3 to issue based on arguments relating to the pH of the claimed formulation, not because of any of the clinical 4

As I understand it, yes.

limitations; right?

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7 All right. Let's take a look at the '239 patent that's shown on PDX-5-3 with highlighting.

So the Patent Examiner found that each of the clinical limitations highlighted here that you've been testifying about was disclosed in the prior art; is that correct?

- That is correct.
- And, in particular, she found that these limitations, again, were taught by the Treschan and Russell references; right?
- Treschan and Russell and Young. Α.
- So a number of references? 18
- 19 Certainly. Α.
 - And the Examiner allowed the '239 patent to issue for reasons other than the differences between the clinical elements of the claim and the prior art; right?
 - That is true. Α.
- 24 All right. Let's go back to your list of all of the 25 clinical limitations and that's again on PDX-5-4.

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Cross - direct

And the Examiner found that all of these limitations were known and practiced before Par's patent; right? I certainly found all of these limitations as having existed in prior art. I'm not sure she enumerated each one of these. Let's talk about a couple of them. So the Examiner understood and believed that Par did not invent intravenous administration of Vasopressin; right? Α. True. And the Examiner understood and believed that Par did not invent --Objection, Your Honor. MR. KWON: questions going to the state of mind of the Examiner. think these questions should be rephrased. MR. RHOAD: Your Honor, he reviewed -- he has reviewed the prosecution histories. He has testified about what the Examiner knew and understood at the time, and, in fact, in his report, he had a whole section entitled Examiner confirmed my opinions about the prior art, or something along those lines. THE COURT: Let me see counsel at sidebar. (Sidebar conference held as follows.) MR. RHOAD: So it goes to the issue of inequitable conduct and materiality and intent to deceive.

Cross - direct

So they've alleged that, for example -- they've elected, for example, that Mr. Kannan made a false declaration when he said, I invented the subject matter of this claim.

And, you know, Dr. Kannan was outside. I invented storage, refrigerated storage and I certainly didn't intend to suggest I invented, you know, the use of Vasopressin for intravenous. And we just want to establish there's no materiality here. The Examiner understood all of that. In fact, she found it all disclosed.

THE COURT: Right.

MR. RHOAD: There's no materiality here.

MR. KWON: A couple rebuttal. First off, we don't agree with the description of false declarations at all, but beside that, we have no objections to this line of questions.

Our objection was directed to how the question was phrased. The question was the Examiner understood, the Examiner understood this was already in the art. Right.

This was going to the state of mind of the Examiner, so we thought we could rephrase the question. We have no issue with this topic.

MR. RHOAD: Based on his review of the prosecution history, where the Examiner said what she was finding and what she believed the prior art to disclose.

Cross - direct

1 MR. KWON: I just --2 THE COURT: I will tell you what. What's good 3 for the good for the goose is good for the gander. I mean, why don't you just --4 fine line. 5 MR. RHOAD: Ask what the Patent Examiner found? 6 THE COURT: Yes. Found or recognized or stated. 7 MR. RHOAD: Okay. 8 THE COURT: I just take out the word understand. 9 Especially given that Mr. Black had objected to the prior 10 statement, it would be consistent. 11 (End of sidebar conference.) 12 BY MR. RHOAD: 13 Okay. So, Dr. Cross, you would agree that the 14 Examiner recognized that Par did not invent the use of 15 Vasopressin to treat intravenous administration, to treat 16 patients using Vasopressin by intravenous administration; 17 right? 18 Α. That's correct. 19 And you understand that the Examiner recognized that 20 Par did not invent the use of Vasopressin to treat patients 21 who are hypotensive; right? That is also correct. 22 Α. 23 And you understand that the Examiner recognized that 24 Par did not invent the use of Vasopressin to treat patients 25 suffering from vasodilatory shock?

- 1 A. Also true.
- 2 \ Q. And you understand that the Examiner recognized that
- 3 Par did not invent treating post-cardiotomy shock with a
- 4 starting dose of 0.03 units per minute; right?
- 5 A. Also true.
- 6 Q. And you understand that the Examiner recognized that
- 7 Par did not invent treating septic shock with a starting
- 8 dose of 0.01 units per minute?
- 9 A. Yes.
- 10 | Q. All right. Let's take a look, please, if you would,
- 11 at PTX-1440.
- 12 A. "P" like Peter; right?
- 13 Q. Correct. Are you there?
- 14 A. Yes.
- 15 \ Q. Do you recognize this as an office action that was --
- 16 that issued in the course of the prosecution of Application
- 17 Number 14717877?
- 18 **A. Yes.**
- 19 Q. Do you see the notification date on the first page of
- 20 | January 11, 2016?
- 21 A. Yes, I do.
- 22 \ Q. Okay. Now, do you see on page 4 of Exhibit PTX-1440,
- 23 | that's where the Examiner is making certain findings, for
- 24 | example, with regard to Treschan?
- 25 A. **Yes**.

1 Q. And this is an example of the type of office action 2 that you reviewed and relied upon for your testimony that 3 you have given about what the Examiner recognized? 4 Α. Yes. 5 Now, you agree that the prior art references that were before the Examiner during prosecution of the '209 and '239 6 7 patents contain the same teachings with respect to the clinical limitations of those patents as the April 14, 2014, 8 9 Vasostrict label; right? Yes. 10 Α. 11 And the Examiner was also aware during prosecution of 12 these patents about the clinical uses of Pitressin; right? 13 Can you please repeat that? 14 The Examiner was aware of the clinical uses of 15 Pitressin during the course of the prosecution of these 16 patents; right? 17 Yes. 18 And from your perspective, any information with 19 respect to Pitressin that was material to the pending patent 20 claims was before the Examiner during prosecution; right? 21 Α. You are referring to clinical? With respect to clinical limitations? 22 Q. 23 Α. Yes. 24 MR. RHOAD: No further questions. I pass the

witness, but I would move to introduce into the record

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Cross - direct

PTX-1440 and DTX-238, the Treschan reference. 1 2 THE COURT: Okay. 3 MR. KWON: Your Honor, no redirect from defendants, but we would like to move to admit exhibits. 4 5 THE COURT: Well, hold on. Do you object to the admission? 6 7 MR. KWON: No objection. No objection, Your 8 Honor. 9 THE COURT: All right. They are admitted. 10 I've got to tell you, Mr. Rhoad, this is an 11 example where I feel like I'm going out on thin ice here, 12 that we're going to introduce an article where there has 13 been no discussion of the contents of the article. So just 14 keep in mind under my rule, neither side is going to get to talk about this Treschan article unless there was testimony 15 16 elicited about the subject matter. This is exactly what 17 I'm --18 MR. RHOAD: Your Honor --19 THE COURT: I am not saying you intend to do 20 that at all. 21 MR. RHOAD: The reason I introduced it for the record, the Examiner relied on it and in 1440 cited 22 23 provisions for it. 24 THE COURT: Right. 25 MR. RHOAD: I just want to have in the record

Cross - direct

1 the reference that the Examiner was talking about and 2 referencing during the course --3 That's fair. What I don't want is THE COURT: if you were going to go in post-trial briefing and say, so 4 5 the Examiner relied on it and here's what it said, this is all the important stuff, I just think there has to be 6 7 some --8 MR. RHOAD: I don't think we'll be pointing to 9 anything besides what the Examiner herself pointed to in the 10 exhibit that was just admitted into evidence and I'm not 11 admitting it for any other purpose. 12 THE COURT: Okay. All right. Then it's admitted. 13 14 (Exhibits admitted into evidence.) 15 THE COURT: All right. Mr. Kwon, you had some 16 exhibits you wanted admitted? 17 MR. KWON: Yes, Your Honor. Defendants would 18 like move to admit Exhibits DTX-36, DTX-246, DTX-178, 19 DTX-249, DTX-25, DTX-42 and DTX-132. 20 THE COURT: All right. Any objection? 21 MR. RHOAD: No objection, Your Honor. THE COURT: All right. They are all admitted. 22 23 MR. KWON: Thank you. (DTX-36, DTX-246, DTX-178, DTX-249, DTX-25, 24 25 DTX-42 and DTX-132 were admitted into evidence.)

1 THE COURT: What's next? You may step down. 2 Thank you very much. 3 THE WITNESS: Thank you. (Witness excused.) 4 5 MR. HALES: Your Honor, we now have a series of 6 video depositions to play. 7 THE COURT: Okay. 8 MR. HALES: And this is a series of, as I 9 mentioned in opening, I think of unavailable witnesses. 10 THE COURT: Oh, this is the inequitable conduct 11 case? 12 MR. HALES: Well, there's a series of them. 13 It's not all inequitable conduct, but those are coming. 14 The first one we're going to play is Inventor 15 Kannan. THE COURT: 16 All right. 17 MR. HALES: That will be one. We have a few 18 more. 19 THE COURT: Okay. 20 MR. HALES: Your Honor, this has per protocol, 21 the integration of the clips designated by both parties. 22 The time for defendants is 22 minutes, 51 seconds. Time for 23 plaintiff, 18 minutes, 03 seconds. You'd better restart. No audio. 24 25 (The videotaped deposition of Vinayagam Kannan

1 was played as follows.) 2 "Question: Could you please state your full 3 name for the record? 4 "Answer: My full name is Vinayagam Kannan. "Question: And it would have been December 2012 5 6 that you started working with JHP; is that correct? 7 "Answer: That is correct. 8 "Question: And what was the purpose for which 9 you were comparing the attributes of JHP's Vasopressin 10 product and other unapproved Vasopressin products? 11 "Answer: The product was, at that time, I was 12 asked to look at because there was an issue when the product 13 was diluted with sodium chlorate and dextrose. 14 having an issue that the potency was declining when it was diluted with dextrose. So I was asked to make a comparison 15 16 and try to help troubleshoot. I was working with Mike 17 Bergren and helping him with my analysis. 18 "Question: So after the project to investigate 19 dilution of the formulation of Vasopressin in dextrose, the 20 next project you worked on was under the direction of Suketu 21 Sanghvi; is that right? 22 "Answer: That's my recollection. There may be 23 some activity happen in between. 24 "Question: So you said the project was related 25 to filing of something for refrigerated product?

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Kannan - designations

1 "Answer: Okay. PAS. That is the prior 2 approval supplement. 3 "Question: Okay. Got it. "And why was Par looking to file a PAS for a 4 5 refrigerated product? "Answer: My recollection is at that time, the 6 7 original approval was for 12-month shelf life with room 8 temperature, so Par was looking to file refrigerated 9 product. 10 "Question: And later you were involved in a 11 project to reformulate that original Vasostrict formulation, 12 correct? 13 In between I was also involved in "Answer: 14 converting that original Vasostrict product into 15 refrigerated product. 16 "Question: When you say you were involved in 17 converting the original Vasostrict product into a 18 refrigerated product, is that something different from your 19 work in reformulating the product? 20 "Answer: That is correct, that is different. 21 "Question: What was involved in the project to 22 convert the original Vasostrict formulation into a 23 refrigerated product? 24 "Answer: I think I answered to this question I worked with Matt Kenney. We evaluated the data. 25 before.

Kannan - designations

We did statistical analysis of what -- whatever need to be done for the submission.

"Question: My question is whether you had any role in determining the indications and usage for Vasostrict as reflected in the label.

"Answer: No, I did not.

"Question: Okay. And then if we look at the dosage and administration section of the approved -- or the label attached to the April 24th approval letter, did you have any role in determining the dosage and administration for the Vasostrict as reflected in the label?

"Answer: As I described earlier to another question, I was involved in evaluating compatibility of Vasopressin injection with sodium chloride and five percent dextrose. The label was updated after that.

"Question: Okay. So April 2014, the approval is given with a 12-month room temperature shelf life, and then later that year in September 2014, a 24 months at two to eight degrees Celsius shelf life is given; is that your understanding?

"Answer: My understanding is around September time frame had 24-month shelf life at 2 to 8 degrees C.

"Question: Do you know what was done to support the change between the April 2014 label with a 12-month room temperature shelf life and the later label with a 24-month

Kannan - designations 1 refrigerated shelf life? 2 "Answer: My recollection is that it was a prior 3 approval supplement filed with FDA. 4 "Question: Okay. Were you involved in the work 5 that led to that prior approval supplement? "Answer: Yes. I was involved with my 6 7 co-worker, Matt Kenney, in evaluating all the data and making recommendations on what the shelf life would be if we 8 9 store at 2 to 8 degrees C. 10 "Question: Okay. And that is different from 11 the later supplement to seek an out-of-refrigerated time; is 12 that right? 13 I do not recollect if that is a "Answer: 14 later -- that later submission was a supplement or some 15 other type of filing. My recollection is that we amended further to add out-of-refrigeration time of 12 months. 16 17 "Question: Between two Vasopressin formulations with identical shelf lives, are you aware of any clinical or 18 19 other advantage for a product that has less impurities 20 within the specifications? 21 "Answer: I am not aware. Okay. And between two Vasopressin 22 "Ouestion: 23 formulations with identical shelf lives, are you aware of

"Question: Okay. And between two Vasopressin formulations with identical shelf lives, are you aware of any clinical or other advantage for a product that has a higher assay value within the specifications?

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1	"Answer: I am not.
2	"Question: And ultimately, as far as you know,
3	the shelf life for the new formulation was not, in fact,
4	improved over the shelf life of the original Vasostrict
5	formulation, correct?
6	"Answer: As far as I know, we have data that is
7	supporting at least 18 months stability when I was at Par.
8	The part that I don't know is that whether that was actually
9	filed with the FDA seeking approval.
10	"Question: I'm handing you a document marked
11	Exhibit 30. And it's a response to the office action in the
12	prosecution of Application No. 14/717,877.
13	"Do you see that?
14	"Answer: Yes.
15	"Question: And that is a patent on which, the
16	patent application on which you were a named inventor; is
17	that correct?
18	"Answer: Yes.
19	"Question: Okay. And if you look at the claims
20	that are being proposed at the time, that's on page 3 of the
21	document.
22	"Do you see that?
23	"Answer: Yes.
24	"Question: Okay. And so there's a claim 16
25	which describes a method of increasing blood pressure in a

Case 1:18-cv-00823-CFC-JLH Document 333 Filed 01/18/22 Page 197 of 359 PageID #: Kannan - designations 1 human in need thereof the method comprising. 2 "Do you see that? 3 "Answer: Yes. "Question: And then it talks about providing a 4 5 particular pharmaceutical composition with certain 6 attributes, right? 7 "Answer: Yes. 8 "Question: Then it talks about storing the unit 9 dosage form from 2 to 8 degrees Celsius; right? 10 "Answer: Yes. "Question: And then administering the unit 11 12 dosage form to a human, given certain amounts of Vasopressin; correct? 13 14 "Answer: Correct. 15 "Question: And the human is hypotensive, right,

"Question: And the human is hypotensive, right, in that claim?

"Answer: Correct.

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"Question: I handed you a document marked

Exhibit 31. It is an office action, and the notification
date is October 21, 2015.

"Do you see that?

"Answer: I see that.

"Question: And this is, again, in the prosecution of Application Number 14/717,877.

"Do you see that?

	3.0029
1	"Answer: I see that.
2	"Question: And the Examiner is Christina
3	Bradley?
4	"Do you see that?
5	"Answer: I see that.
6	"Question: Okay. And then if you look at page
7	3 of the document, you'll see that certain claims of the
8	application are rejected as anticipated by or in the
9	alternative as obvious over the FDA label for Vasostrict
10	published April 2014.
11	"Do you see that?
12	"Answer: I see that.
13	"Question: And then the Examiner goes on to say
14	what the FDA label teaches.
15	"Do you see that?
16	"Answer: I see that.
17	"Question: And then the Examiner goes on, on
18	page four to explain how the label describes limitations of
19	other claims.
20	Do you see that?
21	"Answer: I see.
22	"Question: I've handed you what is being marked
23	as Exhibit 32. Exhibit 32 is a copy of the April 2014
24	Vasostrict label taken from the prosecution history of
25	application number 14/717,877.

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Kannan - designations

1 "Do you recognize this document? 2 "Answer: Yes. 3 "Question: And this is the same Vasostrict label that was approved by the FDA when they approved the 4 original formulation of Vasostrict in April 2014, correct? 5 "Answer: Correct. 6 7 "Question: If you can keep Exhibit 32 handy, but I'm handing you what's being marked as Exhibit 33. 8 9 "And Exhibit 33 is a response to final office 10 action and request for continued examination, also in 11 prosecution of application number 14/717,877, electronically 12 filed on November 24, 2015. 13 "Do you see that? 14 "Answer: Yes. 15 "Question: Okay. And then if we look on page 6 16 of Exhibit 33, you'll see towards the top the statement is 17 made that "Applicant submits that the label is not prior art 18 because the label was disclosed less than one year prior to 19 the effective filing date by another who obtained the 20 disclosed subject matter directly or indirectly from the 21 joint inventors." 22 "Do you see that? 23 "Answer: I see that. 24 "Question: If you go -- stay on the same page 6 25 of Exhibit 33. You'll see there's an explanation as to why

Kannan - designations

1	Par was claiming that the disclosure was made by another who
2	obtained the subject matter disclosed directly or indirectly
3	from the inventor or a joint inventor.
4	"Do you see that?
5	"Answer: I see that.
6	"Question: And one of the bases for that
7	statement is a declaration from you, correct?
8	"Answer: Correct.
9	"Question: And on page 7, the second to last
10	paragraph, you see there it states, "As described by
11	Vinayagam Kannan, the label discloses part of the subject
12	matter of the claims, including a method of increasing blood
13	pressure in a hypertensive human.
14	"Do you see that?
15	"Answer: I see that.
16	"Question: And then it goes on to describe some
17	of the information in the label.
18	"Do you see that?
19	"Answer: I see that.
20	"Question: And then it says, Kannan states that
21	the FDA obtained this information from v. Kannan and Matthew
22	Kenney as they invented this subject matter.
23	"Do you see that?
24	"Answer: I see that.
25	"Question: And for that statement, it's relying

1 on paragraph 7 of your declaration, correct? 2 "Answer: Correct. 3 "Question: I'm handing you what's been marked as Exhibit 34. Is this the declaration that you submitted 4 5 during prosecution of the application number 14/717,877 that's being referred to in the office action Exhibit 33? 6 7 "Answer: Yes. 8 "Question: Okay. And in paragraph 4, you state 9 that you've reviewed the final office action, correct? 10 "Answer: Correct. 11 "Question: And then you include a section in 12 your declaration titled, 'The label recites subject matter 13 that the FDA obtained either directly or indirectly from 14 joint inventors of the '877 application.' 15 "Do you see that? "Answer: I see that. 16 17 "Question: Okay. And if we look down, you 18 describe what the label discloses in paragraph 7. 19 "Do you see that? 20 "Answer: I see that. 21 "Question: Okay. And you say -- and at the end of that paragraph, you say, 'The FDA obtained this 22 23 information from me and Matt Kenney as we invented this 24 subject matter.' 25 "Do you see that?

1 "Answer: I see that. 2 "Question: Right. And you provided the 3 declaration in response to the Examiner's rejection in order to convince the Examiner to withdraw the rejection of your 4 5 claims over the April 2014 Vasostrict label and grant you a 6 patent, correct? 7 "Answer: Correct. 8 "Question: Did you work with anyone to put together this declaration? 9 10 "Answer: The declaration was drafted by the 11 legal department. 12 The internal Par legal department? "Ouestion: 13 "Answer: Internal Par legal department. 14 "Question: Okay. If we go to paragraph 7 of 15 your declaration, Exhibit 34. You state, 'The label 16 discloses part of the subject matter of the claims including 17 a method of increasing blood pressure in a hypotensive The label recites that Vasostrict is indicated to 18 19 increase blood pressure in adults when vasodilatory shock 20 who remain hypertensive.' 21 "Stopping there. You did not invent the method 22 to increase blood pressure in adults with vasodilatory shock 23 who remain hypotensive as described in the label, correct? "Answer: That is correct, I did not invent. 24 25 "Question: Now, the label describes a

1 pharmaceutical composition for intravenous administration 2 having 20 units of Vasopressin per ml. 3 "Do you see that? "Answer: I see that. 4 5 "Question: And if you need to refer to the label in Exhibit 32, you can. The formulation is described 6 7 on page 6 in Section 11. 8 "Do you see that? 9 "Answer: Yes, I see that. 10 "Question: And, again, you did not invent that 11 formulation described in section 11 on page 6 of the 12 April 2014 Vasostrict label, right? 13 "Answer: Correct. 14 "Question: And your declaration goes on to state, 'The label further recites that the Vasopressin 15 16 formulation comprises chlorobutanol and water for injection, 17 USP adjusted with acetic acid to pH 3.4 to 3.6.' 18 "Do you see that? 19 "Answer: I see that. 20 "Question: And that is part of the formulation described in the label, Exhibit 32, that you did not invent, 21 22 correct? 23 "Answer: Correct. 24 "Question: The paragraph 7 of your declaration, 25 Exhibit 34, goes on to state, 'The label recites the

Kannan - designations

infusion rate of the claim by stating that for post
cardiotomy shock, start with a dose of 0.03 units per ml
per minute, 'excuse me, for septic shots, start with a dose
of 0.01 units per minute.
"Do you see that?
"Answer: I see that.
"Question: And you didn't invent that subject
matter either, correct?
"Answer: Correct.
"Question: Okay. Then it states, the label
recites refrigeration of the diluted Vasopressin for up to
24 hours.
"Did you invent that subject matter?
"Answer: I may have contributed to that
subject.
"Question: The FDA, we saw earlier, suggested
refrigerating the Vasostrict formulation, correct?
"Answer: Based on the document we reviewed
earlier.
"Question: Right. And so you did not invent
the idea of refrigerating diluted Vasopressin, correct?
"Answer: What I am trying to say, I contributed
in evaluating that refrigeration of diluted Vasopressin.
"Question: How did you contribute to that?
"Answer: Early on when I started at JHP

1	pharmaceuticals, if you recall, we were trying to address
2	the issue of loss in potency with dextrose injection.
3	"Question: Right.
4	"Answer: And diluted, so
5	"Question: Right. And that you didn't invent
6	the idea of refrigerating a Vasopressin formulation for
7	24 hours after dilution, correct?
8	"Answer: That 24 hours, it is described here,
9	it was based on the data generated at that time.
10	"Question: In what way was it based on that
11	data?
12	"Answer: Could you please clarify your
13	question?
14	"Question: Yes. How was the statement in the
15	label for refrigeration of the diluted Vasopressin for up to
16	24 hours based on data that you prepared?
17	"Answer: I don't know how it is based on the
18	data that I prepared. I am saying I contributed to that
19	evaluation.
20	"Question: That evaluation was related to
21	dextrose in a diluted formulation, correct?
22	"Answer: Correct.
23	"Question: It wasn't related to testing amounts
24	of refrigeration of a diluted formulation, right?
25	"Answer: It was also related to study and

1 testing of diluted formulation. 2 "Question: In what way? 3 "Answer: The label claim is based on the data generated at that time. 4 5 "Question: So it's your testimony today, sir, that of all of the information that is provided in paragraph 6 7 7 of your declaration, you contributed in some way only to the claim for refrigeration of diluted Vasopressin for up to 8 9 24 hours, right? 10 That is correct. I would also like to "Answer: 11 clarify that for the claims related to this patent 12 application, my major contribution is towards refrigerated storage at 2 to 8 degrees C for the final product. 13 14 "Question: Under penalty of perjury, you told the Patent Office that you and Matthew Kenney contributed 15 all of the subject matter in paragraph 7, correct? 16 17 "Answer: As it states here, that the 18 information was taken from me and Matt Kenney. 19 "Question: Right. As you invented this subject 20 matter, right, that's what you told the Patent Office. 21 Correct? I see that it states here. 22 "Answer: 23 "Question: Right. And that this subject matter 24 is referring to the subject matter earlier in the paragraph 25 7; correct?

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Kannan - designations

"Answer: I would like to clarify again for the record my contribution to the claims in this particular patents were related to refrigerated storage at 2 to 8 degrees C. I did not invent other claims related to -related to composition or how it is administered. "Question: Right. And so your statement that you and Matthew Kenney invented all the subject matter in paragraph 7 of your declaration, Exhibit 34, was false; right? "Answer: I am not sure if this is meant to say that we invented all of the subject matter in the claims. "Question: That's what it says, though. That's how it reads; right? "Answer: As I mentioned before, my contribution was related to refrigerated storage. And as it states in the document, we invented the subject matter. "Question: I'm handing you what has been marked as Exhibit 35. It's a copy of an office action dated January 11, 2016. "Do you see that? "Answer: I see it. "Question: And it's responsive to a communication on November 24, 2015. Do you see that? On the second page, you can see that.

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Kannan - designations

1 "Question: Do you see on page 2 of the 2 document, there's discussion of withdrawn rejections. Do 3 you see that? 4 Yes. "Answer: 5 "Question: And the Examiner withdraws the rejection over the April 24th Vasostrict label? 6 7 "Do you see that? 8 "Answer: Yes. 9 "Question: And the Examiner states that 'The 10 declaration by inventor Vinayagam Kannan,' that's you, 11 correct? 12 "Answer: That's correct. 13 "Question: And that's referring to the 14 declaration we just looked at, Exhibit 34, correct? 15 "Answer: Correct. "Question: So the declaration by inventor 16 17 Vinayagam Kannan includes an unequivocal statement that he and Matthew Kenney invented the subject matter disclosed in 18 19 the FDA label and relied upon in the rejection. 20 "Do you see that? "Answer: I see that. 21 22 Okay. And did you ever correct your "Question: 23 declaration to specify that you, as you say now, only meant that you invented the subject matter of the refrigeration 24 25 conditions?

1 "Answer: I don't remember any further 2 communication happened after that. 3 Regardless of what you meant in "Question: paragraph 7 of your declaration, Exhibit 34, it is not true 4 that you invented all the subject matter that's described in 5 paragraph 7 of your declaration, right? 6 7 "Answer: I would like to clarify again my contribution related to refrigerated storage of the product. 8 9 "Question: Right. And none of the other 10 subject matter in paragraph 7 of your declaration, 11 Exhibit 34, right? 12 That is correct. "Answer: 13 "Question: Dr. Kannan, I'm handing you what has 14 been marked as Exhibit 36 to your deposition. 15 document entitled 'Declaration under 37 C.F.R. Section 1.132 16 by Vinayagam Kannan, and it is filed in the Application Number 14/717,882. 17 18 "Do you see that? 19 "Answer: I see that. 20 "Question: And in this declaration, you are 21 analyzing data that was previously provided to the Patent 22 Office by Sunil Vandse, correct? 23 "Answer: Yes. I see it's stated here. 24 "Question: Okay. In paragraph 15 of your 25 declaration, Exhibit 36, you're describing the technique of

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Kannan - designations

1 normalization. 2 "Do you see that? 3 "Answer: Yes. "Question: And is normalization an important 4 5 technique in comparative stability studies? "Answer: Normalization in this case would be 6 7 applied because we had two different starting amounts of Vasopressin in two different studies. We wanted to compare 8 9 the data, so we normalized as percent change in assay values 10 so we can get a direct comparison between two studies. 11 "Question: Okay. And then if you look in 12 paragraph 16, you state, 'As described above, because the 13 procedures for each of the experiments were the same, and 14 because pH was the only variable that was not normalized, I conclude that the difference in the assay percent label 15 claim Vasopressin remaining and percent total impurities 16 17 results for each formulation were attributable to changes in 18 pH. ' 19 "Do you see that? 20 "Answer: I see it. Okay. So you told the Patent Office 21 "Question: that the only variable that was not normalized in the data 22 23 that was provided to them was pH, correct? 24 "Answer: This paragraph is referring to the 25 paragraph above. Can I read that paragraph --

1 "Question: Yes. 2 "Answer: -- before I answer your question? 3 "Counsel, this is referring to the paragraph above which talks about normalization for percent decrease 4 5 in Vasopressin assay. In that context when we're looking at assay data percent pH, assay data was normalized. 6 7 "Question: This sentence talks about three 8 variables, pH, assay and impurities; correct? 9 This is in paragraph 16 of your declaration, Exhibit 36. Do you see that? 10 11 "Answer: Yes. I would like to clarify that the 12 total impurities data was not normalized, because it is not 13 impacted by the starting amounts of Vasopressin as a 14 percent. 15 "Question: Are you telling me that the starting point of a test of impurity increase is not important? 16 17 "Answer: No. I would like to clarify again. 18 In this strategy related -- the studies discussed here, 19 there were two different studies. 20 "The only difference between those study was the difference in the starting amount of assay. I don't recall 21 the numbers. One was started close to 100 percent and the 22 23 other was lower. 24 "When we were evaluating the data of absolute 25 amounts of Vasopressin or total impurities, total impurities

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Kannan - designations

show a clear trend between pH values. Assay was not. this -- this response was related to a cushion why there is a discrepancy between two data sets. Then we normalized the data so that we are able to clearly see the trend for the assay values. "Question: You didn't normalize the data for the impurities, right? "Answer: We did not. "Question: Well, rather, actually, you didn't normalize them in the information provided to the Patent Office, right? Yes. My understanding is that we did "Answer: not normalize impurities. "Question: Right. So for this test, the test of Vasopressin formulations between 2.5 and 4.5, by the time you submitted your declaration to the Patent Office, you had normalized data for impurities as well. Correct? "Answer: I don't remember that. "Question: You've been handed documents marked Exhibits 37 through 39. Exhibit 37 is a document Bates labeled Par-VASO-0034737. And it is an e-mail from yourself to Matt Kenney. "Do you see that? "Answer: I see that.

"Question: And then you state, 'Please see if

Case 1:18-cv-00823-CFC-JLH Document 333 Filed 01/18/22 Page 213 of 359 PageID #: Kannan - designations 1 these make sense. If yes, we can clean up and send to 2 Gina.' 3 "Do you see that? "Answer: I see that. 4 5 "Question: Who was Gina that you were referring to in Exhibit 37? 6 7 "Answer: Gina is from Legal. 8 "Question: Can you take a look, please, at 9 Exhibit 38. Exhibit 38 is a document bearing 10 Par-VASO-0034748. 11 "Do you see that? 12 "Answer: Yes. 13 "Question: And I will represent that this is 14 one of the attachments to the e-mail that is Exhibit 37, 15 okay? "Does this plot, Exhibit 38, provide normalized 16 17 data for rate of change of impurities for the pH study you conducted? 18 19 "Answer: It shows the rate of change of

impurities, so...

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"Question: Okay. And that's using normalized data for pH and impurities, correct?

"Answer: You normalize only the impurities data and plot across the pH range.

"Question: I'm handing you what has been marked

1 as Exhibit 40 to your deposition. This is the declaration 2 of Sunil Vandse, and it's dated January 22, 2016. 3 "Do you see that? "Answer: Yes. 4 5 "Question: If we look at Figure 1 of the declaration on page 5. This is Sunil Vandse's declaration, 6 7 Exhibit 40. This is presenting percent total impurities versus pH at the 25-degree temperature, correct? 8 9 "Answer: Correct. 10 "Question: And if we look between 3.3 and 3.4 11 and then down to 3.5 to 3.6, the data is presented as 12 showing a steep drop in impurities; correct? 13 "Answer: Between 3.4 and 3.5? 14 "Question. Correct. 15 "Answer: Is that your question, counsel? 16 "Question: Yes. 17 Answer: That's where the data points are, yes. 18 "Question: So for 25 degrees that we see in 19 Figure 1 of Sunil Vandse's declaration, the break in total 20 impurities between 3.4 to 3.5 is due to the fact that the starting material had more than twice the amount of 21 impurities than for the two point -- for the 3.5 to 4.5 22 23 study, correct? 24 "Answer: I'm not able to conclude one way or 25 the other based on 25-degree C data alone. I would like to

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Kannan - designations

see the 40 degrees C data also. "Question: So my question is only about the 25-degree data and the break between the values for the total impurities for the 3.4 and the break in the impurities for the 3.5. Can we focus on that? "Answer: Yes, we can focus on that. What I'm trying to say is the extent of change at 25 degrees C will not be significant to see difference across the pH range. That's the reason why I'm saying we would have to look at the 40 degrees C data also to make that conclusion. "Question: That's fine. I'm just trying to understand the reason for the break between the data point for total impurities at 3.4 at 25 degrees C and the data point for total impurities at 3.5 at 25-degree C. So I want to focus on that. "Answer: Yes. "Question: So if we look at the starting total impurities at 3.4, there were 1.75 percent at initial? "Answer: Correct. "Question: And at 3.5, 0.815 percent, correct? "Answer: Correct? "Question: And so the 3.4.25 degree sample started with more than twice the amount of impurities as the 3.5 sample, correct?

Mathematically, yes.

"Answer:

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Kannan - designations

"Question: And at four weeks, the 3.5 sample hadn't even reached the starting impurities for the 3.4 sample, correct? "Answer: At 3.4 weeks at 25 C, it is 2.03 percent. And at pH 3.5, it is 1.23 percent. "Question: Right. And so it's the difference in the starting point of impurities that accounts for the break between the data in the total impurities at 25 degree C between 3.4 and 3.5 pH, correct? "Answer: It may be possible that that could be the difference. "Question: At one point I tried to write this I think you said you know we have data supporting at least 18 months shelf life at room temperature and you don't know whether the data was actually filed with the FDA. "Is that -- are my notes accurate as to what you testified? "Answer: That is correct. "Question. What are you referring to when you say you know there's data supporting at least 18 months shelf life at room temperature? "Answer: While I was at Par Pharmaceutical, we were continuing the room temperature stability studies and established data that supports a shelf life of at least 18 months.

1 "Question: And was that an improvement over the 2 original Vasopressin, originally approved Vasopressin 3 formulation? 4 That is, in my opinion, is an "Answer: improvement over the original formulation that was initially 5 approved with 12 months -- 12 months of shelf life. This is 6 7 a significant improvement." 8 (End of videotaped deposition.) MR. KWON: Your Honor, pursuant to the 9 10 deposition designations, defendants would like to move to 11 admit Exhibits DTX-36, DTX-666, DTX-67, DTX-328, PTX-329, 12 PTX-330, PTX-373, PTX-376 and PTX-377. 13 MR. RHOAD: No objection. 14 THE COURT: No objection. All right. They're admitted. 15 (DTX-36, DTX-666, DTX-67, DTX-328, PTX-329, 16 17 PTX-330, PTX-373, PTX-376 and PTX-377 were admitted into 18 evidence.) 19 MR. KWON: Your Honor, the next witness is 20 another deposition clip by Matthew Kenney, who is another 21 named inventor of the three -- another named inventor of the 22 patents-in-suit and the '239 patent. 23 Eagle's time for this clip is five minutes and 54 seconds. Par's time is four minutes and four seconds for 24 25 the total time of nine minutes and 58 seconds.

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Kenney - designations

Par has also designated counter-designations from Matthew Kenney's June 2020 deposition in the Amneal Matthew Kenney was also deposed in his personal capacity, and in that clip Par's time is four minutes and 58 seconds and Eagle has no time, so for the total time of four minutes and 58 seconds. MR. RHOAD: Your Honor, just one thing. told that the exhibit we wanted introduced from Dr. Kannan's testimony was not stated on the record, so it's PTX-372 that we would ask be admitted. MR. KWON: No objection, Your Honor. THE COURT: All right. It's admitted. (PTX-372 was admitted into evidence.) (The videotaped deposition of Matt Kenney was played as follows.) "Question: Could you please state your full name for the record. "Answer: Matthew Walter Kenney. "Question: Where do currently work? "Answer: Par Pharmaceutical. The court reporter has handed you "Question: what's been marked as Exhibit 4. The beginning Bates number for this document is Par-VASO 001573. It ends with Par-VASO 0015586.

"Okay. Now, this is the original Vasostrict

1	label approved with the approval of the application; is that
2	correct?
3	"Answer: This document says it was from the
4	approval from the from 2014 with revised April 2014.
5	"Question: Now that you have reviewed the
6	label, can you please let me know if you worked on or
7	contributed to any portions of this label, including any
8	content thereof?
9	"Answer: I don't recall doing any work that
10	contributed to the information on this label.
11	"Question: Do you recall submitting any
12	information to the regulatory department whatsoever
13	regarding the April 2014 label?
14	"Answer: I don't recall contributing any
15	information for this label.
16	"Question: Now, when you say that pH is most
17	stable at 3.8
18	"Answer: Mm-hmm, yes.
19	"Question: Is that for all types of Vasopressin
20	formulations?
21	"Answer: I can't say that.
22	"Question: Okay. Is the pH of 3.8 the most
23	stable pH regardless of the type of buffers or components
24	used in Vasopressin formulations?
25	"Answer: Yeah, we can't say that unless we

1 studied all buffers. 2 "Question: Okay. Let's talk about reformulated 3 Vasostrict. 4 "Answer: Okav. 5 "Question: Over that product's shelf life, does 6 its pH change? 7 "Answer: There's no trend in pH. There -- it's 8 just some noise. 9 "Question: What do you mean by noise? 10 "Answer: It means just the error in 11 measurement. It might be 3.8 one time, 3.9, 3.7, but it 12 never trends outside that very close buffered range. 13 "Ouestion: Is there a critical difference in 14 stability to Vasopressin formulations between the pH of 3.8 15 and pH of 3.6? "Answer: Critical is subjective. And either 16 17 way, we would have to do the study with only those two 18 variables changed. 19 "Question: Have you performed such study with 20 regard to the original Vasostrict? 21 "Answer: Original Vasostrict, no, not as far as I recall. 22 23 "Ouestion: What is different about shelf life between the original Vasostrict and the reformulated 24 25 Vasostrict?

1 "Answer: The reformulated Vasostrict has the 2 ability to have a commercially viable room temperature shelf 3 life of 19 months. "Question: Didn't you just testify that the 4 5 room temperature shelf life of the reformulated Vasostrict is 12 months? 6 7 "Answer: That's what's stated in the package 8 It is capable of a 19-month shelf life at room insert. 9 temperature. 10 "Question: Based on Par's internal studies, is 11 that what you are saying? 12 "Answer: Yes. 13 "Question: So in terms of practical 14 consequence, in terms of customer use of the reformulated 15 Vasostrict and the original Vasostrict, both the room temperature and the refrigerated shelf lives are the same; 16 17 is that right? "Answer: Our reformulated Vasostrict would have 18 19 less impurities than the original Vasostrict. 20 "Question: At what time point? 21 "Answer: Throughout the shelf life. 22 "Question: Now, you testified earlier that the 23 formulation itself doesn't control the level of impurities 24 at time zero; is that right? 25 "Answer: Correct.

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Kenney - designations

"Ouestion: So provided that the measurement of total impurities was undertaken as soon as the drug product gets made, then the level of impurities would not be attributable to the degradation from the drug product; is that right? "Answer: I'm not aware of any impurities that form as soon as you make the drug product. "Question: If the starting impurities in different samples that were tested were different, why would you ever not normalize the data before comparing them? "Answer: And I don't -- I don't know what I would do in that situation. It would depend on how the data looks, if it would make sense to do that, do it that way. "Question: Now, earlier you testified that you believe the pH of 3.8 is the most stable or optimal pH for Vasopressin formulations; is that right? "Answer: Yes. "Question: What does -- in terms of the timing, at what time point are you referring to that pH? be -- so are you saying that the pH of 3.8 is the most optimal pH at the time the product is formulated or at some time after? According to our studies, pH 3.8 "Answer: affords the most stable drug product over shelf life.

"Question: And do you recall reviewing one of

Kenney - designations

the stability tables for the original Vasostrict which started at pH of 3.5 and then 'spiked' to pH 3.8 at 18 months?

"Answer: Yes. I don't know if 'spiked' is the correct term, but yes, I remember that data table.

"Question: So at that pH at 18 months, would that Vasopressin formulation be more stable than when it started out with by the virtue of the pH?

"Answer: I don't know. We would have to run that study.

"Question: Doesn't studies that Par performed up to this date that show that pH of 3.8 is the best pH, also show that the formulation that increased its pH of -- its pH to 3.8 at 18 months would also be most stable?

"Answer: No, that cannot be concluded because you are bringing up the original Vasostrict that was a different formulation, it was not buffered and it contained chlorobutanol. So the data we have were on buffered solutions without chlorobutanol, so that batch in particular that you are referencing, that study would have to be carried out to determine if it was more stable than in the other Vasopressin batch.

"Question: Okay. Why does the fact that the original Vasostrict has chlorobutanol have any implication on what the most stable pH would be?

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Kenney - designations

"Answer: Because when we compare the studies, we have to control the variables, and the studies that I looked at didn't have -- or not all of them had chlorobutanol in them. "Question: So would it be correct to say that the pH of 3.8 is the most stable pH for a Vasopressin formulation that does not contain chlorobutanol and that is buffered? "Answer: According to the studies we've ran, that was our conclusion. "Question: And you cannot extrapolate that conclusion to another Vasopressin formulation at a pH of 3.8 that contains chlorobutanol and that is non-buffered; is that right? "Answer: I wouldn't be comfortable making that conclusion based on the non-controlled variables. "Question: But the testing that was done in November 2015 for the pH range of 2.5 to 3.4 used the same lot of Vasopressin as the test that was performed on March 17 th, 2015 from the pH range of 3.5 to 4.5; is that correct? "Answer: Yes. "Question: All right. And it's possible that the lot of API would have had more impurities through

degradation during the time span starting from March 2015

1 to the second testing date of November 2015; is that right? 2 "Answer: We would have to compare the stability 3 data on this lot. 4 "Question: So would it be correct to say that 5 every lot of API supplied to Par could have a different level of total impurities? 6 7 "Answer: If they met -- yes, as long as they 8 were within specification. "Question: Can you quantify how much better 3.8 9 pH is compared to a 3.6 pH? 10 11 "Answer: So one thing that we can quantify 12 is our current formulation versus the previous Vasostrict, which the previous Vasostrict had an estimated shelf 13 14 life of 15 months. And then with our formulation work, we were able to improve the estimated shelf life to 15 19 months. 16 17 "Question: And is that out of refrigeration, in 18 refrigeration, or what? 19 "Answer: That's room temperature stability. 20 "Question: And what is Exhibit 11? 21 "Answer: This is the product development technical report I wrote for Vasostrict 20 units ml pH 3.8 22 23 acetate buffer, single dose. 24 "Question: In the course of your reformulation 25 work, how many experiments do you think that you ran?

	kenney - designations
1	"Answer: Probably close to 100.
2	"Question: And in the course of your
3	reformulation work, how many formulations did you and the
4	other inventors of the '785 patent study?
5	"Answer: So it was at least 70 formulations.
6	"Question: Can you please turn to the page
7	ending in 625.
8	"Answer: Okay.
9	"Question: Did you and your the other
10	co-inventors of the '785 patent present any conclusion about
11	the most stable region of pH for total impurities within
12	Exhibit 11?
13	"Answer: Yes.
14	"Question. What was that conclusion?
15	"Answer: So the conclusion is based on the data
16	pH 3.8 is the desired target for pH for Vasopressin
17	solutions.
17 18	solutions. "Question: Did you make any conclusions about
18	"Question: Did you make any conclusions about
18 19	"Question: Did you make any conclusions about the most stable region of pH for total impurities within
18 19 20	"Question: Did you make any conclusions about the most stable region of pH for total impurities within Exhibit 11?
18 19 20 21	"Question: Did you make any conclusions about the most stable region of pH for total impurities within Exhibit 11? "Answer: The most stable region of pH for total
18 19 20 21 22	"Question: Did you make any conclusions about the most stable region of pH for total impurities within Exhibit 11? "Answer: The most stable region of pH for total impurities is 3.7 to 3.9.
18 19 20 21 22 23	"Question: Did you make any conclusions about the most stable region of pH for total impurities within Exhibit 11? "Answer: The most stable region of pH for total impurities is 3.7 to 3.9. "Question: Did you and the other co-inventors

1 original Vasostrict product? 2 "Answer: Yes. 3 "Question: And what did the data show with respect to the relative assay over shelf life? 4 "Answer: So the estimate shelf life of the 5 original Vasostrict was around 15 months. The reformulated 6 7 Vasostrict with the 3.8 pH acetate buffer, the shelf life was increased to around 19 months. 8 9 "Question: And similarly, what did the data 10 show with respect to the relative amount of impurities over the shelf life? 11 12 "Answer: So I don't remember those numbers, but 13 the amount of impurities was decreased dramatically over the 14 estimated shelf life. "Ouestion: And has that been shown -- is the 15 16 data you're relying on been shown to be statistically 17 significant? 18 "Answer: Yes, so the shelf life calculations 19 are done with FDA compliance software to estimate shelf life 20 with a 95 percent confidence interval. 21 "Question: Just one more question, Mr. Kenney. Even though the labels are the same with respect to storage 22 23 conditions for the original formulation of Vasostrict and 24 the reformulated product, does Par have data to support a

longer shelf life out of refrigeration for the reformulated

1 product? 2 "Answer: Yeah, so our data indicates an 3 estimated shelf life with 95 percent confidence interval of 19 months, which means that it is possible to have this 4 5 product on the market with an 18-month room temperature shelf life on the label." 6 7 (End of videotaped deposition.) 8 MR. KWON: Your Honor, pursuant to the 9 deposition designations, defendants move to admit DTX-30. 10 MR. GREENE: Your Honor, Blake Greene of Plaintiffs move to admit DTX-1115. 11 Dechert. 12 THE COURT: Will either of you object to the 13 others? 14 MR. GREENE: No objection. 15 MR. KWON: No objection. They're all admitted. 16 THE COURT: 17 (DTX-30 and DTX-1115 were into evidence.) 18 THE COURT: All right. I guess we probably need 19 to give the court reporter a break, so why don't we come 20 back at 3:25. All right? Thank you. 21 (Short recess taken.) 22 23 (Proceedings resumed after the short recess.) 24 THE COURT: All right. Please be seated. 25 right. Next?

1 MR. LASKY: Your Honor, defendants call Dr. 2 Leonard Chyall. 3 THE COURT: All right. ... LEONARD JESSE CHYALL, having been duly 4 5 sworn/affirmed as a witness, was examined and testified as follows... 6 7 MR. LASKY: Your Honor, may I approach? 8 THE COURT: Please. 9 MR. LASKY: I believe the binders have already 10 been delivered. 11 THE COURT: All right. 12 DIRECT EXAMINATION 13 BY MR. LASKY: 14 Good afternoon, Dr. Chyall. Can you please introduce yourself to the Court? 15 Good afternoon. I'm Dr. Leonard Chyall. 16 17 And did you prepare demonstratives to help with your 18 testimony today? 19 Α. Yes, I did. 20 Can we look at your slide number 2. Can you please 21 describe your educational background? I received a Bachelor's degree from Oberlin College 22 23 with a major in chemistry. I then attended graduate school at the University of Minnesota and received a Ph.D. from the 24 25 chemistry department. I was a post-doctoral fellow at

Chyall - direct 1 Purdue University, 1992 to 1996, also in the chemistry 2 department. 3 MR. LASKY: Your Honor, the microphones are very 4 quiet. 5 THE COURT: If the streaming feed is off, does something happen? I think it might be the cooling system. 6 7 MR. LASKY: Okay. I wasn't myself able to hear 8 Dr. Chyall. 9 THE COURT: You cannot hear him? 10 MR. LASKY: I couldn't hear him very well. 11 THE COURT: I thought you were complaining --12 not complaining in fairness to you, that there was noise in 13 the air. Try that. 14 THE WITNESS: Can you hear me okay? BY MR. LASKY: 15 16 Yes. That's better. Did you write a thesis for your 17 Ph.D.? 18 Α. Yes, I did. 19 What did the thesis relate to? Q. 20 My thesis concerned the stability of a class of 21 organic compounds called cyclopropanes, so I synthesized those compounds and then conducted a study to measure the 22 23 study by looking at their degradation products. Can we move to slide 3, please. Can you describe your 24

profession background for the Court, please.

Chyall - direct

A. After my postdoctoral fellowship, I worked at Great

Lakes Chemical Corporation. That company was in the

business of manufacturing and selling organic compounds into

a variety of markets and I worked in the R&D division with

the focus of identifying new products for the company.

In the year 2000, I joined a pharmaceutical services company called SSCI and that company later became Aptuit.

SSCI is in the business of helping other drug companies with the chemistry aspect of the drug development concept. So while I was there, I conducted a lot of research that was done on behalf of other drug companies with a focus on the chemistry aspects of their products.

And then in the year 2011, I started my own consulting business called Chyall Pharma, where I consult in the same technical areas that I did when I worked at SSCI.

- Q. In part of your work experience, have you conducted analytical stability testing of pharmaceutical compositions?
- A. Yes. Stability testing is something I've done on numerous occasions, really starting in the year 2000 when I worked at SSCI.
- Q. Do you have any teaching experience?
- A. Yes, I do. Most recently, I taught introduction to organic chemistry at Purdue University and this was last summer and I taught undergraduates that are majoring in some

1 aspects of the life sciences.

- Q. Can you please open your binder to DTX-203-A.
- 3 A. Okay. I'm there.
 - O. What is DTX-203-A?
- 5 A. It's a recent copy of my CV.
 - Q. Does it accurately represent your education and work experience in the field?
- 8 A. Yes.

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- 9 Q. What types of analyses were you asked to do in this 10 case?
- 11 A. I was asked to look at some stability studies that Par 12 scientists conducted that relate to stability of Vasopressin 13 as a function of pH.
- 14 Q. And do you have experience relevant to that analysis?
- 15 A. Yes. PH dependent stability studies are things that
 16 I've done in the past.
 - MR. RHOAD: Your Honor, we tender Dr. Chyall as an expert in the analytical testing and pharmaceutical formulations and evaluation of data for such testing.

MR. BLACK: No objection to that extent, but I know that he was not offered as an expert in stability of peptide formulations as the other two experts. His expertise is more limited. Some of the questions may be beyond the scope, but if they are, I will check at that time.

1 THE COURT: All right. Thank you. 2 BY MR. RHOAD: 3 Dr. Chyall, we move to slide 4. At a high level, what opinions will you be giving the Court today? 4 5 Today I plan to testify that Par has not shown that the pH limitations of the asserted claims are critical to 6 7 the Vasopressin stability. Also, I plan to testify that Par's pH study could not have addressed criticality when 8 9 compared to the April 2014 original Vasostrict label 10 formulation. 11 What materials did you consider in forming your 12 opinions? 13 I considered the patents-in-suit, the specifications 14 as well as the prosecution histories. I considered earlier 15 Vasopressin patents that are in the same family along with 16 their prosecution history. 17 I considered documents that were produced as 18 part of this litigation from Par's laboratories that 19 involved studies of Vasopressin formulations and I also 20 considered the expert opinions offered by Par's expert, Dr. 21 Kirsch. 22 Ο. Okay. Slide 5. 23 What legal framework did you apply to your 24 criticality analysis in this case?

It's my understanding from Eagle's counsel that when a

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Chyall - direct

claimed range overlaps or abuts a range or value in the prior art, then there is a presumption of a prima facie case of obviousness. The patentee could overcome that obviousness by showing that the claimed range is critical relative to the prior art range or value.

But that demonstration of criticality has to be shown across the full scope of the claim, and a difference has to be in difference in kind, not simply a difference in degree.

- Q. Okay. Turning to slide 6, why is criticality important in this case to your understanding?
- A. Well, it's my understanding that the original Vasostrict is prior art and that original Vasostrict product has a pH of 3.4 to 3.6 and the pH limitations of the asserted claims concern values that are pH 3.7 to 3.9.
- Q. Now, on the left of slide 6, you have DX-30. What is that document?
- A. This is a description from the prescribing information of the original Vasostrict.
- Q. Now, you mentioned the abutting range of original Vasostrict and the claims. What is your understanding of the consequence of that?
- A. It's my understanding from the legal framework that, and this is an example of an abutting range, and as a result there's a presumption of obviousness when you have the prior

Chyall - direct 1 art that abuts those claims. 2 Okay. Slide 7. 3 Has Par elicited any opinions from its experts in an attempt to show criticality? 4 5 I'm aware of opinions offered by Dr. Kirsch in 6 that regard. 7 And in asserting criticality of the claimed range, what evidence does Dr. Kirsch rely on? 8 9 Dr. Kirsch relied on experiments done by inventors 10 that could be found in declarations that were put in front 11 of the Patent Office for the prosecution of earlier 12 Vasopressin patents. And those experiments can also be 13 found in examples 9 and 10 of the asserted patents here. 14 Dr. Kirsch is also relying on comparisons of the reformulated Vasostrict product to other Vasopressin 15 16 products to show criticality. 17 Okay. Slide 8. Q. 18 Now, let's start with the pH study. Do you have 19 an understanding as to why par submitted the data from the 20 pH study during prosecution of the patent? 21 MR. BLACK: Objection. Lack of foundation. Do you have an understanding? If he 22 THE COURT: 23 doesn't have one, it's not going to go. 24 MR. BLACK: I think he does have one.

THE WITNESS: Yes, I have an understanding.

1 BY MR. RHOAD:

- Q. And where did you get that understanding?
- 3 A. From my review of the prosecution history of the
- 4 earlier Vasopressin patents. In particular, the '478, '239,
- 5 | '**526 patents**.
- 6 Q. And based upon your review of the file histories, what
- 7 was Par relying on the pH study to show?
- 8 A. The pH studies were done to show criticality over
- 9 another prior art formulation. It's called PPC,
- 10 | Pharmaceutical Partners of Canada, which has a disclosed pH
- 11 range between 2.5 and 4.5.
- 12 \ \Q. And the file histories you were referring to that you
- 13 | had reviewed, what are the exhibit numbers of those?
- 14 A. So the 478 patent is DTX-7. The '239 patent is
- 15 DTX-ten and the '526 patent is JTX-7.
- 16 \ \Q. And to be clear, those are the file histories, not the
- 17 patents themselves; right?
- 18 A. That's my understanding.
- 19 Q. Now, in your opinion, do the pH studies that Par
- submitted to the Patent Office in these prosecutions show
- 21 criticality of the claimed pH range?
- 22 A. It's my opinion they do not.
- 23 | Q. Okay. Moving to slide 9, at a high level, why in your
- 24 | view do the pH studies not show criticality of the claimed
- 25 range?

Chyall - direct

A. There's two reasons that I have. The first is that the pH study that Par scientists conducted don't address the full scope of the asserted claims and it's also my opinion that those studies themselves, just looking at the data, don't show criticality.

- Q. Your first opinion, why is it important that the pH study did not address the full scope of the claims?
- A. Because it's my understanding that is a requirement to show criticality in the face of a rejection, an obviousness rejection.
- Q. And for purposes of your analysis, what is your understanding of Par's interpretation of the full scope of the pH limitations of the claim?
- A. My understanding is Par considers the full scope of the asserted claims to consider Vasopressin formulations that would be manufactured outside of the claimed pH values of 3.7 to 3.9, but then drift into that range at some point during the shelf life.

THE COURT: So hold up. The question was actually -- the questioner, Mr. Lasky, referred to the full scope of the pH limitations of the claim. I just want to make sure I'm not missing something as a legal matter because this is new to me.

Isn't it the full scope of the claim? In other words, so I'm going to be looking at the pH limitation

relative to the rest of the claim? No?

MR. LASKY: That's true, Your Honor, but again the dispute in this case is over the claims, is over the pH limitations in the claim and what those mean and what the -- what the -- a difference in pH means with respect to stability and so the point here is that Par is asserting that the claims cover a formulation that is manufactured at 3.4 to 3.6 and then drifts into the range for as little, I think we heard, of one minute.

THE COURT: I get that, but I just want to make sure, because your question struck me as it was unusual.

You've got bullet points in earlier slides and even in the witness' answer to your question, which referred to the full scope of the claim, but your question was about you wanted him to address -- so here's what it says:

"Question: Your first opinion, why is it important that the pH study did not address the full scope of the claims?"

There was an answer.

"Question: And for purposes of your analysis, what is your understanding of Par's interpretations of full scope of the pH limitations of the claim?"

MR. LASKY: Correct.

THE COURT: That was intentional?

MR. LASKY: That was intentional because, of

Chyall - direct 1 course, the full scope of the claims is made up of the full 2 scope of each of the individual limitations. 3 THE COURT: Fair enough. I can see how it will be relevant. I want to make sure I understand it. 4 5 MR. LASKY: Yes. We're going to see -- we're 6 going to talk about a few of those limitations. 7 THE COURT: Thank you. BY MR. LASKY: 8 9 So just to reiterate, what is your understanding of 10 Par's interpretation of the full scope of the pH limitations 11 of the claims? 12 My understanding is that Par considers the full scope of the pH limitations to cover products that would be 13 14 manufactured outside the pH range of 3.7 to 3.9, but then drift into that pH range at some point during its shelf 15 16 life. 17 And the pH studies submitted by Par to the -- during 18 prosecution addressed that scenario where the formulation is 19 manufactured outside of the claimed range and then drifts 20 into the claimed range? 21 Α. It's my opinion they do not. Now, moving to slide 10, here we have some excerpts 22

from the patent-in-suit, JTX-2, at columns 97 to 99 and

Now, what are these excerpts?

JTX-3 at column 96 to 98.

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A. So these are two examples that are from the patents in suit that describe the pH experiments that were performed by Par's scientists and included in the declaration for the earlier patents. So these experiments describe how samples were prepared and then analyzed.

- Q. Now, we see a reference to pH in each of these examples. What do you understand that reference to pH to mean for purposes of these examples?
- A. So pH here, the focus is on the initial pH of the various experiments. So what Par scientists did was they prepared a solution of Vasopressin and then adjusted it to a particular pH and then the results obtained were always related back to the initial pH of the experiment.
- Q. And is that consistent with your review of the other data that was submitted in the declaration in other prosecution?
- A. Yes, it is.

- Q. And how does that treatment of pH in examples 9 and 10 compare to the full scope of the claim as Par is asserting it in this case?
- A. This is much narrower because these experiments don't cover a formulation that would say start out at a lower pH and, for example, drift into a higher pH region, such as the claimed region.
- Q. Now, are you aware of any evidence that a Vasopressin

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Chyall - direct

formulation that is manufactured outside the claimed range but drifts into the claimed range after manufacture confers any stability benefit? I'm not aware of any of that evidence. Α. Okay. Turning to slide 11, please. What was the composition of the formulation used in the pH --THE COURT: Can I have a sidebar, please? (Sidebar conference held as follows.) THE COURT: I just want to kind of make sure. This is an issue in Markman, was it, when to measure the pH? It wasn't an issue with Markman and MR. LASKY: we're not raising a claim construction dispute, to be clear. We're saying this is how they're interpreting it. accepting it for purposes of infringement, accepting it for validity, but their decision to interpret the claims to cover just rising into the claimed claim range for a minute, that has consequences on validity. If they want to show criticality, they have to show it at the full scope of the claims. They have to show --THE COURT: I'm just asking because I have other cases with ranges. So, for instance, in this Markman hearing, nobody asked me to say, to construe -- for instance, you guys didn't ask me to construe the claimed

range to be measured at the time of release. That was not

1 an issue in claim construction? 2 That wasn't, and we didn't have the MR. LASKY: 3 inventor testimony about the, you know, what they actually did at that time, but we didn't, we didn't ask for that. 4 5 THE COURT: Okay. So you both agree, there was 6 no dispute, nobody asked me to limit the time frame or --7 MR. BLACK: No. 8 THE COURT: When the range was measured. 9 MR. BLACK: They decided in order to raise 10 that -- you are probably thinking, Your Honor, what they 11 have in the pretrial order by the time they got there, but 12 they had maintained the position without going to Markman 13 the that ordinary meaning was it had to be measured at 14 manufacture. They were going to try to run the case at 15 trial, an ordinary meaning defense on that, which they then 16 dropped. 17 THE COURT: Okay. But then did they raise it? 18 Did it come up in a pretrial order that somebody was 19 suggesting I needed to construe this? 20 MR. LASKY: No. 21 MR. BLACK: I think in the pretrial order you took the position that measurement had to be at manufacture, 22 23 but they dropped that as we got towards the case. 24 THE COURT: When you say you, you are not

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referring to me, the Court?

1 MR. BLACK: I am sorry, Your Honor. 2 THE COURT: I want to make sure. 3 MR. BLACK: My recollection is that Eagle Yes. was maintaining throughout its expert reports and up to the 4 5 pretrial order just before trial that the measurement had to be at manufacture. 6 7 THE COURT: Okay. And, well, it doesn't matter. 8 Now they are not. I just wanted to make sure. I apologize. 9 I think I have another case where that argument is being 10 It's when do I measure the pH. made. 11 MR. BLACK: This argument they are raising now 12 though, I don't know how well developed it is in the 13 pretrial order. We don't think they have the law right. 14 think they're asking the wrong questions. We'll have to deal with that later. 15 16 THE COURT: I assume that has to do with, in 17 other words, what does it matter what Par thinks? 18 MR. BLACK: Yes. What's the scope of the claim? 19 THE COURT: That's what you are going to argue? 20 MR. BLACK: Yes. Also, we just saw evidence with Dr. Park that they make their product and then within 21 the first month, the pH rises, so sometimes in the first 22 23 couple days or weeks. What does any of that mean. 24 THE COURT: Okay. 25 MR. LASKY: Your Honor, we heard from Dr.

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Chyall - direct

Kirsch, it can go up one minute until the very end of the shelf life. He already testified that's enough to meet the limitation. THE COURT: I agree, he testified to that. MR. BLACK: It may be a claim construction point that has to be resolved where we can decide the criticality That's for post-trial briefing. point. MS. WU: Your Honor, if I may? THE COURT: Yes. MS. WU: Amneal didn't specifically push this issue because we understood that it was previously vetted in the Eagle case and we didn't want to re-raise things that might have been dealt with, but I do note that claim construction is an issue that would be considered de novo, so a Court can at any time consider that. In one case I had with Judge Robinson, she construed a claim early on in the case and at trial had a different construction because it was new evidence. THE COURT: Okay. But here's the thing. Amneal is suggesting that -- I'm not saying you are, but if you were going to suggest that post-trial Amneal is going to ask for a new claim construction on validity, it's going to be denied. I'm just saying if it's helpful. No. MS. WU:

I'm not asking --

THE COURT:

No.

1 MS. WU: Amneal would be happy for you to 2 construe it now. 3 THE COURT: I didn't think it would be helpful. MR. BLACK: You're confused because their 4 5 pretrial order is going in one direction. They never 6 contested --7 THE COURT: I just wanted to make sure. The line of questioning suggested to me that I might have 8 9 addressed this in claim construction. We're good. 10 (End of sidebar conference.) 11 THE COURT: All right. Thank you. 12 BY MR. LASKY: So, Dr. Chyall, before the sidebar we were just 13 14 discussing again the pH study described in Examples 9 and 10 of the asserted patents. What was the composition of the 15 formulations used in those pH studies? 16 17 The formulations here, the experiments were prepared 18 by dissolving Vasopressin in water along with a ten millimolar acetate buffer and then adjusted the pH. 19 20 Now, do the claims of the patents-in-suit require an 21 acetate buffer? The claims are broader than that. 22 23 cover an acetate buffer but along the other buffers at other 24 concentrations and then also other excipients and even the 25 absence of buffers and excipients.

1 Q. Are you aware of any data regarding the impact of 2 buffer choice on the stability of a Vasopressin formulation? 3 I have seen in data from Par scientists. Α. Yes. What data are you aware of? 4 I'm aware of data that was submitted in a declaration 5 6 that Par provided -- scientists submitted to the Patent 7 Office in support of patentability of another Vasopressin 8 patent. 9 Okay. Moving to slide 12. Here you have an excerpt 10 from DTX-7, pages 2165 to 66. What is this document? So this is a declaration that Dr. Kannan wrote in 11 Α. 12 support of the patentability of the '478 patent and the data 13 here that I've shown concerns his experiments involving the 14 differences in stability of Vasopressin with an acetate buffer and with a citrate buffer. 15 You mentioned the '478 patent. How does that relate 16 17 to the patent-in-suit here? 18 It's an earlier patent in the same family as the Α. 19 asserted patents. 20 And what did the data that Dr. Kannan submitted in his 21 declaration show? Here, the data showed that the citrate buffer is 22 23 actually inferior in stabilizing the Vasopressin. The state 24 of the impurities are nearly three times higher, 9.3 percent

compared to 2.9 percent for the acetate buffer.

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Q. Would the asserted claims in this case cover a formulation with a citrate buffer? Yes, it would. Α. We turn back to examples 9 and 10. And in your opinion, do those pH studies address the full scope of the claims with respect to the composition used? The experiment 9 and 10, the studies concern only Α. No. one buffer and only at one concentration, so it's much narrower. Okay. Moving to slide 13. So we've discussed your opinion regarding the full scope of the claims. What is the next basis for your opinion on criticality? It's also my opinion looking at the data collected for these pH stability studies, the data itself doesn't show criticality in any event. Okay. Let's move to slide 14. Here you have some excerpts from DTX-10.2362, pages 2362 to 65. What are those excerpts? These are the charts that Par scientists prepared that Α. show Vasopressin stability as a function of pH and stability is shown in two different ways and at two different That's why we have four charts. temperatures. So the top two are with the impurities that are

present after four weeks of storage of these Vasopressin solutions at either 25 or 40 degrees C. And the bottom two

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Chyall - direct

charts, those concern the amount of Vasopressin that degraded upon storage at either 25 or 40 degrees C for the four-week storage period.

And the amount of degradation is expressed as a change in assay.

- Q. Let's move to slide 15. Now, on the left of this slide, you have again a page from DTX-10-2362. What is that?
- A. So this is the chart that plots the impurities that are in each of these Vasopressin vials at four weeks, so the dots here are the individual data points for the individual pH's and the trend line is just a trend line that goes through the data.
- Q. Now, does this graph here show the amount of impurities that were formed over the four-week period of the study?
- A. No, no. This is the impurities that are present at week four. And if you look on the chart on the left, excuse me, on the upper right that I prepared, the percent impurities at week four, these values are like 1.23 percent for pH 3.5. It's that week four impurity value,

 1.23 percent, we would see it there as a 3.5 value.
- Q. I was going to ask you about that table next. It's labeled DTX-7, pages 1893 to 96. What is that document?

So this is from the Vandse declaration that was

Chyall - direct 1 submitted in support of the '478 patent. 2 And what are you showing in the table? 3 The table are the impurities that were measured by Par's scientists at the week four period and then also in 4 5 the center column, those are the initial purities, the week 6 zero impurities that are in the center column. 7 Now, in your opinion, does this graph show that a pH of 3.7 to 3.9 is critical stability? 8 9 No, it doesn't. And the blue box, the annotation that 10 I have at the bottom of the chart, this shows a very broad 11 range of stability that starts at around pH 3.5 and extends all the way to pH 4.5. 12 13 The particular value for the particular 14 impurities present at week four, what I've done in dark blue 15 in my table, I've highlighted values that provide numerical numbers that are just as good, if not better, than the 16 17 values that were measured for pH 3.7 to 3.9. 18 So moving to slide 16, this is another one of Okay. those charts from the Kannan declaration. 19 20 THE COURT: May I just ask a quick question? 21 MR. LASKY: Sure. So the data that is in the table, I 22 THE COURT: 23

just want to make sure, because it looks like it comes from a different document than the graph. But is that data what's represented in the graph?

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1 MR. LASKY: Yes. So what they did here --2 THE COURT: First of all, is it a yes? 3 MR. LASKY: Yes. THE COURT: Okay. Is that going to be an issue? 4 5 Do you all disagree with that? 6 MR. BLACK: No, Your Honor. The data at DTX-7, 7 which is the file history, this is all data presented to the Examiner. 8 9 THE COURT: Right. I just want to know because 10 I wanted to see if it's stipulated, but I you don't agree. 11 guess it's not, that this data in the box is reflected in 12 the graph and that it is stipulated to -- let's cut to the chase and just acknowledge it. 13 14 MR. LASKY: Just to clarify, the graph is only the week four data. We've also included the week 0 data in 15 16 That's not reflected in the graph. So we have a 17 target pH, then there was initial impurities. 18 THE COURT: All right. Zero. But the right 19 column? 20 MR. LASKY: The right column. 21 THE COURT: You're saying the right column, but it's not like he happens to notice there's this similarity. 22 23 I just want to know. Is that in dispute? It's the data. 24 MR. LASKY: As I understand it, it's not. 25 MR. BLACK: Some of the data --

1 THE COURT: Okay. 2 MR. BLACK: -- that's in the patent, the numbers 3 are correct. 4 THE COURT: Okav. 5 MR. BLACK: He read them in. They're a 6 demonstrative, not in evidence. 7 THE COURT: So if I'm looking at it after you guys are gone, I can go, yeah, these numbers are the same 8 9 numbers that are here. Okay. Thank you. 10 MR. LASKY: Okay. BY MR. LASKY: 11 12 Dr. Chyall, to Mr. Black's point, you point to one of the data points that you said is the same as or better than 13 14 the claimed range in terms of impurities at week four? So if we look at the data point for pH 4.1, the amount 15 Α. of impurities that are present in the vial at week four is 16 17 .094 percent and that numerical value is actually lower than any of the three values, pH 3.7, 3.8 and 3.9. 18 19 Okay. So we were on the second graph, I think. 20 does the second graph from the Kannan declaration, DTX-10, page 2363, show? 21 This is the same type of impurity data, but now it's 22 23 for the 40-degree storage condition. So, again, this is a 24 plot of the impurities that are present in each of the vials 25 at week four plotted as a function of pH.

1 0. And have you done the same thing with the table on the 2 right that you did with the previous impurities chart that 3 we looked at? 4 Yes, I did. I highlighted the value at pH 4.0 and the reported value for impurities present at week four is 5 1.64 percent and my point is that value at pH 4.0 is 6 7 actually lower than the value of pH 3.7 that had a reported impurity of 1.75 percent. 8 9 So in your view, does this graph show that a pH range 10 of 3.73 to 3.9 is critical to stability? 11 Α. No, it doesn't. It shows that there's a broad range 12 of stability and adjacent values that are comparable 13 compared to the values that are the same as the claimed 14 range. 15 Okay. So let's move to slide 17, please. And this is 16 the third graph from the Kannan declaration, DTX-10, page 17 2364. What is that document? So DTX-10 is the declaration that Dr. Kannan submitted 18 19 to the Patent Office in support of the '239 patent. 20 Okay. I apologize. So what does the graph show? 21 Sorry. The graph here is a plot of the stability of 22 Vasopressin but now with a focus on the amount of 23 Vasopressin that has degraded. 24 So this is a change in assay as it's called. So

the way this is done is the assay for the value at the week

zero, the initial assay, that's subtracted from the week
four, so you could have an understanding of the change in
the assay.

- Q. Okay. And so can you explain how the data in your table on the right relates to the data in the chart?
- A. So in the appendix of DTX-7, that's the Vandse declaration, it contains the numerical value. What they have in the declaration are the week zero assay and the week four assay, but the chart on the, the column on the right, this is what I did to show what the actual numerical values are that are plotted in the graph that is shown on the left of the slide here.
- Q. Now, so just to be clear, what are the dots seen on this graph?
- A. The dots are related to the change in assay, so it's the amount of the Vasopressin that's degraded. The higher the data point, that means the more degradation of the Vasopressin that has occurred.
- Q. Now, are there dots on this graph for every pH value represented on the graph?
- A. No. There's no data point plotted at pH 3.5, 3.7, or 3.8.
 - Q. And have you calculated what the data point should have been at those points?
- 25 A. Yes. From the data in the appendix of DTX-7, I

plotted, well, I calculated that those would actually be negative decreases in assay.

- Q. What does it mean to have a negative decrease in assay?
- A. Well, that would -- that somehow the Vasopressin is actually improving in purity over time and that is just not scientifically possible, so this just speaks to scatter in the data using this analysis method.
 - Q. We also see in addition to the dots, there's a wavy line that goes down and then comes up. What is that line?
 - A. So the straight line is a trend line basically just meant to represent the stability of Vasopressin and the function of pH.
 - Q. Now, if we focus on the missing point at 3.5, what does the trend line suggest the assay degradation is at that point?
 - A. So if you look at the pH 3.5 value for the following trend line and then taking it over to the Y axis, the vertical axis, that would suggest around a .9 percent change in assay.
- Q. And how does that compare to the actual change in assay for the pH 3.5 study?
 - A. Well, the pH 3.5 study has effectively no change in assay, so the trend line would suggest greater instability of 3.5 than the actual data shows.

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there was a missing 3.5.

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0. Now, how does that compare with where the trend line is for the other two missing data points? So at pH 3.7 and 3.8, those are also values that have a negative decrease, which means that they are stable, an area of stability. Here the trend line actually demonstrates that. Now, in your opinion, is there any rational scientific Q. basis for leaving out the data points at pH 3.5, 3.7 and 3.8? It's my opinion that all data points should be Α. No. plotted on a chart for an experimental study. So if we consider the data points that are on the graph and the data points that you calculated, does this graph show a pH of 3.7 to 3.9 is critical to stability? No, it doesn't, and for the same reasons, we see Α. adjacent value for assay decrease that are just as good, if not better, than the, for example, the pH 3.9 value. Can you give us an example of one of those, please? 0. THE COURT: Just hold on a second. So can you clarify again, because the way you're presenting it, it's a table from one document and a graph from the other, and then you are saying, I understand that the witness is saying

Are you saying that some of the data that was in the study that was reflected in the table data was omitted

1 from the graph? 2 MR. LASKY: So, Your Honor, what they've done 3 here --4 THE COURT: And maybe you can ask the witness to 5 elucidate. 6 MR. LASKY: Okay. 7 THE COURT: I've got to say, you know, I'm 8 confused. I'm going to be the fact-finder and it's probably 9 on me. I'm just confused. 10 MR. LASKY: I will ask some further questions. BY MR. LASKY: 11 12 So, Dr. Chyall, first of all, where was this graph 13 that you are showing here? How was that provided to the 14 Patent Office? 15 It was provided in a declaration by Dr. Kannan. 16 And you mentioned that there were some missing data 17 points on the chart. So where did you get the data from 18 that you used to calculate the missing value? 19 So the data points for the week zero and week four, Α. 20 those were included in the appendix of the declaration. Okay. And so where are you showing your calculation 21 0. in the table on the right side of this slide? 22 23 So the calculation that I'm showing, that percent 24 decrease calculation, so the numerical values on the right

are the ones that I calculated from the data that was in

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the appendix. And those values correspond to the data points on the chart on the left, but as I testified to earlier, not every data point is actually plotted on the chart. THE COURT: Okay. I'm sorry. I'm going to have to -- so, Doctor, DTX, the table that is on -- we're looking at DDX-4-17, which is a slide. THE WITNESS: Yes, Your Honor. THE COURT: Did you prepare the slide? THE WITNESS: Yes, I did. THE COURT: Okay. And on the slide it appears that there are featured two documents. That's what I'm reading. So the first and the primary one in terms of space is labeled DTX-10.2364. Is that pictorial -- that's a graph? THE WITNESS: Yes, Your Honor. THE COURT: Did you just pull that off of an exhibit or did you modify it? THE WITNESS: I --THE COURT: Or did you do both? THE WITNESS: I did both, Your Honor. The modifications are the blue blocks. THE COURT: Right. THE WITNESS: That has the values that I considered to be comparable and then the pink annotation is

1 the claimed range 3.7 to 3.9 and the bracket is also my 2 notation. 3 THE COURT: All right. Other than the bracket 4 with the annotation that says claimed range, the pink square 5 and the blue box which you added, correct? THE WITNESS: Yes, Your Honor. 6 7 THE COURT: Is it otherwise, that is a document 8 in evidence. You wouldn't know if it's in evidence, but that's a document that was labeled DTX. Is that right? 9 10 DTX-10? 11 THE WITNESS: Yes, Your Honor. 12 THE COURT: Okay. Now, the table that's to the 13 right of the slide or on the right side of it, which is 14 identified as DTX-7.1893-96. So there's a document called DTX-7.1893-96. 15 16 That column that has four columns, is that listed from 17 DTX-7? 18 THE WITNESS: No, Your Honor. THE COURT: Tell me what it is. 19 20 THE WITNESS: The values for the first three 21 columns --First of all, did you just make up 22 THE COURT: 23 this column or -- the column that has four columns that says 24 target pH, Vasopressin percent assay. The third column says 25 Vasopressin percent assay. The fourth column says decrease.

1 Did you create that table?

THE WITNESS: I created the table in this

format, but the columns -- you'll find in DTX-7, you'll find

comparable columns of data. I believe my memory is correct,

it's done a little bit differently, where they have all of

the week zero data points and then they have another part of

the column has the week four data points, so kind of all

like in a linear format.

What they don't have, Your Honor, is the mathematical calculation in order to get the percent decrease in assay, the fourth column.

THE COURT: That column that's percent decrease, that's your work product?

THE WITNESS: Yes, Your Honor.

THE COURT: And there's no such -- is there a column called percent decrease in the -- in DTX-7, which was submitted to the Patent Office?

THE WITNESS: From my memory, there's not, but I need to let you know, Your Honor, that those data points, the actual values, like the .5, .6, .4, those numerical values correspond to locations of the dots on the chart.

THE COURT: I got you. I got that much. I'm trying to figure out what's your work product versus what Par submitted to the PTO and that's what is confusing me, see. So it sounds like -- okay. Well, now it disappeared.

1 Okay. So this is a slide. The slide wasn't 2 given to the PTO. Right? 3 THE WITNESS: Correct, Your Honor. THE COURT: It sounds like the graph was given 4 to the PTO without the modifications you made. 5 6 right? 7 THE WITNESS: Yes. 8 THE COURT: And then I guess something else 9 labeled DTX-7, which was a table, was given to the PTO, and 10 you basically created your own table and had taken out data from that table? 11 12 THE WITNESS: Yes. If you don't mind, there was 13 just briefly on my screen, there was an actual chart of that 14 data that I used to prepare my table. MR. LASKY: James, can you pull that up? 15 This is from DTX-7, which is actually the full 16 17 file history of the '478. That is from that Vandse 18 declaration that's cited. This is the appendices for those. 19 THE COURT: Okay. So you summarized that or 20 excerpts from it in a chart? 21 THE WITNESS: Yes, Your Honor. THE COURT: Okay. Now, can we go back to the 22 23 So when you said there was stuff that was missing, slide? 24 there was a 3.5 data point was missing. Is that my 25 recollection?

1 THE WITNESS: Correct. 2 THE COURT: Okay. And so was that missing from 3 the graph in the Kannan declaration? Was it missing from the table that was in the Vandse declaration or both? 4 5 THE WITNESS: The data point is missing from the graph, and just to remind you, Your Honor, the percent 6 7 decrease values are not in the original exhibits, that table that we had just looked up. 8 9 THE COURT: Right. 10 THE WITNESS: So you won't find any of the 11 values on the far right column. 12 THE COURT: I'm good with that. Am I going to 13 be missing any of this other data, like in the red box where 14 you have target pH 3.5, Vasopressin 96.2 for week zero and 96.3 for week four, is that in the table that was submitted 15 16 to the PTO? 17 THE WITNESS: Yes. You'll find the 96.3, 96.2 You'll find those values in DTX-7. 18 and 96.5. 19 THE COURT: Okay. Anything other than your 20 notations on the graph in the fourth column that you've 21 added on this slide that was not submitted to the PTO? 22 THE WITNESS: No. 23 THE COURT: Okay. Thank you. BY MR. LASKY: 24 25 Just to reiterate, considering the data points that

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you testified are missing from this graph, does the graph show that a pH of 3.7 to 3.9 is critical stability? It's my opinion there's no showing of any critical stability between pH 3.7 and 3.9 from the data on this graph. All right. Can we move to the next slide, slide 18. This is another graph from the DTX-10, page 2 '365. What are you showing on this slide? So this is the assay decrease data. It's the same type of data we had on the previous slide, but now this is for stability at 40 degrees storage conditions. And the table on the right, did you prepare that in the same way we just described you prepared the table for the previous slide? Α. Yes, I did. And does this graph show that a pH of 3.7 to 3.9 is critical to stability? No, it doesn't. I just want to refer to some adjacent Α. values at pH 3.5 and 3.6. There, the percent decrease values of 2.2 and 2.5 percent, those values are actually lower than some of the values that were measured in the claimed range.

What does it mean to have a lower value there?

What it means, it's better, better stability with respect to preventing the degradation of the Vasopressin. 1

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at week zero.

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Q. So we just looked at all four of those graphs individually. If you consider them collectively, in your opinion, do they show that a pH of 3.7, 3.9 is critical stability? It's my opinion that when you look at the data as a whole, there's no showing of criticality. Q. Now, turn to slide 19. What are you showing on this slide? So this chart on the left is the impurity data for the impurities that are present at week four upon storage of 25 degrees C and here I have some annotations onto the chart as well. Can you describe your annotation? So what I've done is indicated with different colors that these two studies are actually done at different times. The charts are presented as one study, but there were actually two separate sets of experiments. One set was done November of 2015 and that's the low pH region, and then the March 2015 study is the high pH region. And what are you showing in this table on the right? On the right, this is a table where I summarized the data from DTX-7. These are the percent impurities measured at week four and then also the percent impurities measured

1 Q. Okay. Where did you take this data from?

A. This is from the Vandse declaration that was a part of the file history for the '478 patent.

- Q. Are these data points that are in those same appendices we just saw a little earlier?
- A. Yes.

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- Q. Now, how do you know that these would -- this data is from two separate studies conducted at different times?
- A. I reviewed the laboratory notebooks that Par scientists maintained the documents for these experiments.
- 11 Q. Okay. Moving to slide 20. Here you have some 12 excerpts from DTX-82. What is DTX-82?
 - A. This is Mr. Matthew Kenney's laboratory notebook, a

 Par scientist and inventor who documented in preparing the

 Vasopressin samples that were used for the stability study.
 - Q. At the top you have some excerpts from a particular page of DTX-82, page 39. What are you showing there?
 - A. Well, here I'm showing the preparation of the high pH studies that were done in March of 2015 and I've highlighted that the solid Vasopressin, the Vasopressin API lot number 056 was used, and it has a potency of 480 units per milligram with an expiry of October 21st, 2015.
- A. These -- these experiments were set up on March 17th, 25 2015.

1 Q. Okay. Now, if we move down to the bottom of your 2 slide here, slide 20, we see some excerpts from page 87 of 3 DTX-82. 4 So what formulations does it show the 5 preparation of? So the November 2015 notebook entry here shows the 6 7 documentation for the low pH formulation, 2.5 to 3.4. 8 And what was the lot number of the API that was used Q. 9 in that second study? 10 Here, it's lot number 19056, the same lot as the March Α. 11 7th. 12 Okay. And how does the date that the second study was Ο. 13 conducted compare with the original expiration date below, 14 the lot that was used? 15 The second study was done after the original Α. expiration date for the March study. 16 17 Now, did Mr. Kenney or -- did Mr. Kenney do anything to account for that? 18 19 Yes, yes, he did. You can see in the notebooks that Α. 20 now that lot number 19056, it has a different -- a different 21 potency. Now it's listed as 456 units per milligram. 22 a new expiration date is also assigned, December 31st, 2015. 23 Now, have you considered what the impact was of having

used -- having conducted these two studies at different

25 times?

A. Yes, I have.

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- Q. We move to slide 21. What are you showing on this slide?
- A. So here I'm showing the amount of the starting
 impurity on these pH stability experiments. So these are
 the impurity levels that were measured just upon
 formulation, the so-called week zero impurity. And the
 graph, the bar chart that I've prepared here shows a marked
 increase in impurities for the November 2015 study at the
 low pH region when you compare it to the level of the
- 12 Q. And is the data you're plotting here shown in the table on the right side of your slide?

starting impurity for the March 2015 study.

- A. Yes, it is.
- Q. And where -- excuse me. Go ahead.
- A. I was just going to say that the week zero impurity
 data here, I obtained that from DTX-7, which is one of the
 Vandse declarations for the '478 patent.
 - Q. Again, is that the same appendices we looked at a little earlier?
- 21 A. Yes, it is.
- Q. So what does this bar graph show with respect to the starting impurities for the two studies?
- A. Well, here we can clearly see that there's more impurity in the November 2015 study, starting impurity

compared to the March study.

- Q. And can the amount of starting impurities that are in a formulation impact the final amount of impurities at four weeks?
- A. Absolutely. If you start with a high level of impurities, those impurities are going to get continued through to week four and you wind up measuring those initial impurities along with the impurities that formed over time.
- Q. Now, has Dr. Kirsch asserted an alternative explanation to these differences in starting impurities?
- A. Yes. Dr. Kirsch has suggested that the higher impurities are due to the instability of Vasopressin just initially immediately upon putting them into that pH environment. And he points to the gradual slight increase in going from pH 3.4 to 2.5 as evidence of that.
- Q. Now, do you agree with Dr. Kirsch's conclusion?
- A. No, I disagree. You can see there's a very sharp break in the initial impurity, pH 3.4 compared to 3.5 pH and that distinct break in the amount of impurity falls right at the pH level. It divides the two different pH studies.
- Q. Now, given these differences in starting impurities, in your opinion was it appropriate to combine the data from the two studies?
- A. No, it wasn't. If you are only going to plot like impurities that are present at week four without any

consideration of the week zero impurities, that's an improper way to plot and analyze the data.

- Q. And is there anything that the inventors could have done to account for these differences in starting material?
- A. Yes. They could have done a process called
 normalization, which is just to subtract the initial
 impurities from the final impurities and then they would
 know how much formed over time and we saw a similar type of
 analysis for the assay chart that we were looking at
 earlier.
 - Q. Okay. Let's move to slide 22. So over here on the left we again see one of the -- the 25-degree impurity data that was in the Kannan declaration. Does that show -- well, what does that show?
 - A. Well, here what I've shown with my vertical line right at the differences in color, which define the differences in the chronology, in the experiment, I am showing there's a distinct increase in impurity at the pH 3.4 measurement compared to the 3.5 measurement, and this is impurities on storage for the four-week period here.
 - Q. And what is the extent of that break?
- A. It looks like it's on the order of around one percent or so, a little less than one percent.
 - Q. Okay. And that's one percent of the total impurities at week four?

1 A. Correct.

- Q. Okay. Now, you've got a different graph shown on the right. First of all, which annotations have you added to that graph just to clear the record?
- A. The graph on the left, all of the color annotations
 are mine, so that would include the shading for the pH
 ranges in the red box -- I'm sorry, with the red line, the
 rectangle and the shaded area for the claimed range, and
 then the annotation about the claimed range below.
- Q. Okay. So the document on the right, what are you showing there?
- A. This document on the right is actually normalized impurity data. This is a chart that Par has prepared.
- 14 Q. Do you know where this normalized graph came from?
- 15 A. Yes. The graph, and then setting aside the color 16 annotation, the graph was from an e-mail correspondence 17 between Dr. Kannan and Mr. Kenney.
- 18 Q. Can you turn in your binder to DTX-66, please.
- 19 A. Okay.
- Q. Is DTX-66 a document you considered in providing your opinions in this case?
- 22 A. Yes, it is.
- 23 Q. And what do you understand DTX-66 to be?
- 24 A. This is an e-mail between Dr. Kannan and Mr. Kenney.
- 25 Q. And this is actually a document that we saw Dr. Kannan

1 | talk about during his deposition testimony; right?

A. That's correct.

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- 3 Q. Okay. And what's the date on this e-mail?
- 4 A. March 22nd, 2016.
- Q. Okay. And you mentioned that the normalized data
 was -- well, strike that.
- What does Dr. Kannan say to Mr. Kenney in this e-mail?
- 9 A. Here, he asks, please see if these make sense. If
 10 yes, we can clean up and send to Gina.
- 11 Q. Okay. And were there some attachments to the e-mail?
- 12 A. Yes.
- 13 Q. And the graph that we saw on your previous slide,
- 14 DTX-67, was that one of the attachments to the e-mail?
- 15 A. Yes. That was Vasopressin rate RF 25 degrees C.
- Q. And were you in the courtroom when Dr. Kannan's
- 18 A. I was.

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19 Q. Do you understand who Gina is?

testimony was played?

- A. Yes. My understanding is she's a Par employee that worked in the legal department.
- Q. Okay. In your review of the file histories for the patents, have you seen that normalized graph, DTX-67?
- 24 A. I'm sorry. I didn't hear your question.
- 25 Q. Yes. In your review of the file histories for Par's

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patents, have you seen the normalized graph, DTX-67, was never disclosed to the Patent Office? No, it wasn't disclosed. Α. Okay. If we can go back to the slide. 0. Now, are the two graphs you're showing on your slide 22, are they describing the same study? Yes, they are. Just the data was plotted in a Α. The data on the left was -- data that was different way. not normalized and the data on the right was the normalized impurity data. And did the normalized and non-normalized data from Q. the study tell the same story about --THE COURT: Was there one study or two studies? MR. LASKY: Well, sorry. Let me clarify the question. THE COURT: I just want to know. I thought it was two studies. MR. LASKY: It's two studies. What I meant was the 25-degree impurity, so I will clarify that in the record. BY MR. LASKY: Now, the normalized graph on the right, is that 0. showing the data from the same two studies of the normalized -- that the non-normalized -- let me start again. Is the non-normalized data on the right showing

1 | the same two studies as the non-normalized data on the left?

A. Correct.

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- Q. Okay. And is it showing the same temperature?
- A. Yes. This is the 25-degree data.
- 5 Q. Now, and also, are they both plotting impurities at
- 7 A. Correct.

four weeks?

- Q. Now, the two, the non-normalized table on the left, does it tell the same story about the stability of the a Vasopressin as the normalized graph on the right?
 - A. No, it doesn't. The normalized data indicate the broader range of stability and that is what I've meant to highlight here with these two blue boxes.

You can see on the chart on the right, when you actually account for the initial impurities that are present in these Vasopressin formulations, you can see impurity levels after the storage condition that are in the same range as the values that are between 3.7 and 3.9. So it's just a broader region of stability once normalization is done to -- this is really data processing.

- Q. Now, in your opinion, was there a rational scientific basis for the inventors not to submit the normalized impurities data to the Patent Office?
- 24 A. No.
- 25 Q. And does the normalized impurities graph show that a

1 pH range of 3.7 to 3.9 is critical to stability?

A. No, it doesn't.

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- 3 Q. Okay. If we turn to slide 23, here we have an excerpt
- 4 | from DTX-10, page 2370. What is -- sorry. Page 2369 to 70.
- 5 What is this document?
- A. So this is an excerpt from the Kannan declaration that
 was put in in support of the '239 patent.
- Q. And what did Dr. Kannan represent to the Patent
 Office?
- 10 A. Here, I've highlighted in paragraph 32 that he's
 11 telling the Patent Examiner that pH was the only variable
 12 that was not normalized.
- Q. In your review of the file history, did you see an indication as to why Dr. Kannan was telling the Patent

 Office that pH was the only variable not normalized?
 - A. Yes, because the Patent Examiner identified that same break in the data that I identified with respect to the impurities at pH 3.4 and 3.5.
 - Q. And so why was what Dr. Kannan was saying there about the pH being the only variable that was not normalized relevant to that question?
 - A. Because the Examiner was interested in knowing whether that break was due to the fact that the studies were done at different times, because the Examiner appreciated that there were two different time frames for these studies.

Chyall - direct 1 0. And based on his representation that pH was the only 2 variable that was not normalized, what did Dr. Kannan tell 3 the Examiner was the reason for that break? He told the Examiner that the differences in 4 5 impurities were attributable to changes in the pH and he was not aware of any other factors that would account for the 6 7 differences in the result for each formulation. And so then what did he conclude in his statement to 8 9 the Examiner here? 10 So at the end he concludes that the data presented 11 above are attributable to pH and not to the fact that the 12 data were collected on different days. 13 Now, starting with the first statement that you've 14 underlined here, pH was the only variable -- well, let's 15 back up a step. When you said pH was the only variable that was 16 17 not normalized, what other variables are described in this 18 paragraph of Dr. Kannan's declaration?

- The other variables would be the impurities, the Α. differences in assay and Vasopressin remaining.
- 0. Now, was it true, was this statement that pH was the only variable that was not normalized true?
- The impurities were not normalized. Α. No.

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24 And what is the significance again of the impurities 25 not having been normalized?

A. Because if you don't take into account your starting level of impurities, then your final measured impurities would also include those starting materials.

Q. So what is your conclusion having seen the normalized data regarding whether the results of the data presented

A. The results presented were attributable to also the fact that you started with more impurities in certain experiments.

- Q. Now, when was Dr. Kannan's declaration signed?
- 11 A. On May 22nd, 2017.

were attributable to pH?

- Q. And how does that compare with the date of the e-mail
 we saw where Dr. Kannan was sending the normalized data to
 Mr. Kenney?
- 15 A. It's later.

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- Q. Now, in order to have generated a normalized impurities graph like DTX-67 from the data that was presented to the Patent Office, what would the Examiner have had to have done?
 - A. Well, she would have had to wondered whether these things that were represented were accurate and then she would have had to go into the appendix and find the raw data like I did. And then she would have had to do that subtraction of the impurities in order to get the change in impurity. And then she would have had to have taken that

data and then make her own chart of the normalized impurity and then compare that chart to the chart that was provided in a declaration.

- Q. Now, if we move to slide 24. Here we see another graph, DDX-79. What is that graph chart?
- A. This chart, without my annotations, was also an attachment to the e-mail correspondence from Dr. Kannan to Mr. Kenney. This chart here is the normalized impurity data for 40 degrees C temperature.
- Q. Was this chart disclosed to the Patent Office?
- 11 A. No, it wasn't.

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- Q. Now, what is your understanding of how Dr. Kirsch
 weighs this 40 degrees normalized data compared to the
 degrees normalized data we saw in DTX-67?
 - A. It's my understanding from Dr. Kirsch's report that he places more emphasis on the 40-degree data because that provides more of a change in assay and impurity.
- Q. And do you agree with Dr. Kirsch's prioritization of the 40 degrees in that way?
 - A. No. It's my opinion that all of the data should be considered and it's not fair to the data to just emphasize one set over another. This is especially true when you consider that Vasopressin formulations are going to be stored at room temperature and below.
- Q. Okay. Considering all the data that you've seen from

the pH studies, if your view, does that support criticality of the claimed range?

A. It is my opinion that it does not.

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- Q. Okay. Let's skip ahead to slide 26. What other data does Dr. Kirsch rely on for criticality?
- A. Dr. Kirsch makes comparisons of the reformulated

 Vasostrict product to other Vasopressin products to show an

 increase in performance that he says is evident to

 criticality.
- Q. What does Dr. Kirsch compare reformulated Vasostrict to in that analysis?
- A. So he compares reformulated Vasostrict to the original
 Vasostrict product and he also compares the reformulated
 Vasostrict to Eagle's ANDA product.
- Q. And what is reformulated Vasostrict intended to represent in that comparison?
- 17 A. An example of the claimed invention.
- Q. And what are original Vasostrict and Eagle's products intended to represent in that comparison?
 - A. So for this comparison, those are examples that are outside of the claimed invention.
- Q. Okay. In your opinion, do those comparisons show criticality of the claimed range?
- 24 A. In my opinion, they don't.
- Q. Okay. Moving to slide 27, this is DTX-111 you are

showing here. What is that document?

- A. This is from the prescribing information of the reformulated Vasostrict product.
- Q. And what are you showing here in your annotation?
- A. Here, I've highlighted that this product is adjusted to a pH of 3.8.
 - Q. And how does that compare to the full scope of the pH limitations of the claims as Par is interpreting them for infringement?
 - A. It's narrow for at least a couple of reasons. The first is that the claimed range is 3.7 to 3.9 and then this is just 3.8, so it's just within that, that broader range.

Also, this reformulated product is an example of a product that is targeted to, adjusted, I should say, to have a pH of 3.8, which is in the claims. It doesn't concern a product that would have an adjusted pH that's outside of the claimed range and then drift into the claimed range at sometime later on.

- Q. Okay. Moving to slide 28. How does the formulation of reformulated Vasostrict compare to the formulation of the original Vasostrict product and the Eagle product that Dr. Kirsch is comparing it to?
- A. Well, there are differences in more than just the pH.

 The reformulated product does not contain any chlorobutanol,
 but that excipient is present in the original Vasostrict as

well as Eagle's product.

The reformulated product has sodium acetate buffers that's added and then that buffer is not included in the original product or Eagle's product.

pH adjusting is different. Sodium hydroxide and hydrochloric acid are used for the reformulated product, but pH adjustments for original Vasostrict in Eagle's product involves acetic acid.

- Q. And can those differences other than pH have an impact on the stability of the formulation?
- A. Yes. Well, here's an example. We have more than one thing going on, so it's not possible to attribute any difference in performance of just pH, because there are other variables in these products.
- Q. And were you in the courtroom when Mr. Kenney's deposition testimony was played?
- 17 A. Yes, I was.
- 18 Q. And did he have any testimony relevant to that point?
 - A. Yes, he did. His testimony was consistent with the opinion I just offered in that it's not possible to note the decrease in performance is due to one particular thing when there's no one thing that's being changed, such as chlorobutanol.
 - Q. And we've also heard some testimony about Par having data to support a longer shelf life for reformulated

1 | Vasostrict. Did you hear that testimony?

- 2 A. Yes, I did.
- 3 | Q. I think there was also a suggestion that reformulated
- 4 | Vasostrict might have lower impurities than original
- 5 | Vasostrict. Did you hear that in the testimony?
- 6 A. Yes.
- 7 | Q. Now, first of all -- well, has reformulated Vasostrict
- 8 been given a longer shelf life than original Vasostrict?
- 9 A. It's my opinion, it's my understanding it has not.
- 10 The shelf lives are the same.
- 11 \ \Q. And if we skip ahead to slide 30, have reformulated
- 12 | Vasostrict been given a broader impurity specification than
- 13 original Vasostrict?
- 14 A. No. The release impurities and the shelf life
- 15 | impurities of five percent and 17 percent respectively,
- 16 those are the same impurities for the original Vasostrict or
- 17 the reformulated Vasostrict product.
- 18 Q. Now, if there is data to support a longer shelf life
- 19 | for reformulated Vasostrict, can we conclude that that is
- 20 due to the pH of the product?
- 21 A. We can't reach that conclusion because there are other
- 22 differences other than pH.
- 23 \ Q. Now, have you looked at the pH, sorry, yes. Have you
- 24 | looked at the pH specifications for reformulated Vasostrict
- 25 | in forming your opinion?

A. Yes, I have.

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- 2 \ \oldots \oldots \ \oldots \oldots \ \oldots \ \oldots \ \oldots \ \oldots \ \oldots \ \oldot
- 3 specification for pH for reformulated Vasostrict?
 - A. Anywhere between pH 3.6 and 4.0.
- Q. And for reformulated Vasostrict, what's the stability
- 6 specification for pH?
- 7 A. It can be anywhere between 2.5 and 4.5.
- 8 Q. What is the document you're relying on for that, for
- 9 those numbers?
- 10 A. This is DTX-72. It's the NDA specification for the
- 11 reformulated Vasostrict product.
- 12 Q. So does a batch of reformulated Vasostrict in order to
- 13 meet the specification need to have a pH in the claimed
- 14 range on release?
- 15 A. No, it doesn't. It can either be above or below the
- 16 claimed range, as long as it's within 3.6 to 4.0.
- 17 Q. Does a batch of reformulated Vasostrict need to have a
- 18 pH within the claimed range at any time during its shelf
- 19 | life?
- 20 A. No, it doesn't. It's really secondary. Anywhere
- 21 between 2.5 and 4.5.
- 22 \parallel Q. So how, if at all, does that impact your criticality
- 23 analysis?
- 24 A. My opinion is that there's no showing of criticality
- 25 when the release spec and stability spec would allow for the

product to be released and have a shelf life that never falls within the claimed pH region.

- Q. Now, if we move forward to slide 32, what other opinion will you be providing the Court today?
- A. I also plan to testify that Par's pH study could not have addressed criticality when compared to the April 2014

 Vasostrict label formulation.
- 8 \ Q. And why is that issue significant in this case?
 - A. Well, once the Vasostrict label was removed from consideration as prior art, then the focus was on criticality with respect to the PPC, the pharmaceutical partners in Canada, prior art product, which had a broad range of 2.5 to 4.5.
- Q. Now, during prosecution of which patent was the 2014

 Vasopressin label disqualified as prior art?
- 16 A. I believe it was during the '239 patent.
- Q. And how does the '239 patent relate to the patents-in-suit in this case?
- 19 A. It's in the same family as the asserted patent.
 - Q. And if you consider the date when the Examiner
 withdrew her rejection and disqualified the label as prior
 art in prosecution of the '239 patent, when would the
 applications for the patents-in-suit filed relative to that?
- 24 A. Later on.

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25 Q. And was the Examiner the same for the asserted patents

as for the '239 patent?

A. Yes. Christina Bradley.

prosecution of the asserted patents?

- Q. Okay. If we move to slide 33. Was the 2014

 Vasostrict label ever cited to the -- by the Examiner during
- A. It was referenced, but not as a prior art. It was just referenced as an evidentiary reference. The Examiner used it to relate the weight of Vasopressin in milligrams to the units of a potency of Vasopressin.
 - Q. If we skip forward to slide 35 in the interests of time.

If the Examiner had considered the 2014

Vasostrict label as prior art during prosecution of the asserted patents, in your view, would the stability data that the inventors submitted to be sufficient to address criticality over that label?

- A. In my opinion, it could not have been shown to be critical.
- Q. And why is that?
 - A. The original Vasostrict product is FDA approved product with a narrow pH range, known stability, and the last studies that were done for the criticality declarations and then put into the claims of the asserted patents, those were a four-week study that was done over a broad pH range. It wouldn't have the amount of scientific rigor needed to

show criticality when you have this FDA approved product with a pH range that's adjacent to the pH ranges that are being claimed.

MR. LASKY: Your Honor, I pass the witness. No further questions.

THE COURT: All right.

CROSS-EXAMINATION

8 BY MR. BLACK:

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- Q. Dr. Chyall, good afternoon.
- 10 A. Good afternoon.
- 11 Q. You were here for Dr. Park's testimony? You were here
- 12 for Dr. Park's testimony, I assume?
- 13 A. Yes.
- Q. And you heard him describe a person of skill in the
- 15 art as someone with a background in pharmaceutical
- 16 formulations with expertise in peptide formulations. Do you
- 17 recall that?
- 18 A. Yes.
- 19 Q. You are not actually a POSA, are you?
- 20 A. Well, I have some limited experience with peptide
- 21 stability. I worked on a project when I was at SSCI that
- 22 involved stability of peptides.
- 23 \ Q. How long ago was that?
- 24 A. I think it was in the -- probably around 2010 or so.
- 25 No. Earlier than that. I think 2005 or 2007 time frame.

1 Q. Right. You knew I was going to ask that question.

2 | That's the best peptide experience you could come up with?

- A. It is my peptide experience with respect to studying a
- 4 peptide.

- 5 Q. You do have a lot of expertise in testifying in cases
- and at deposition and hearings; is that correct?
- 7 A. I definitely have served as an expert witness before.
- 8 Q. Yes. And by before, you mean you testified in, at
- 9 deposition over 50 times; right?
- 10 A. Yes. Not for 50 different matters, but I have sat for
- 11 a lot of depositions.
- 12 \parallel Q. And that's what you told us at your deposition in
- 13 | February of 2020. It has been a year-and-a-half. I'm
- 14 wondering how Covid was for the deposition, virtual
- 15 deposition business. How many depositions have you taken
- 16 since our deposition in this case?
- 17 A. I believe just one other.
- 18 Q. Okay. All right. So the state of the prior art at
- 19 the time included, as I believe you said that the
- 20 | formulation of the pH for original Vasostrict was well-known
- 21 as 3.4 to 3.6; correct?
- 22 A. Yes. That's the adjusted pH that is on the label that
- 23 | I showed.
- optimized at 3.5 at the time of the priority date of the

patent; is that correct?

- A. Well, my opinion is that there's a broad range of stability and that's shown by the work that Par scientists did and others, other work, but that is a broad range that would include pH 3.5.
- Q. You had a section of your report that said that the pH for Vasopressin formulations had already been optimized.

Do you recall that?

- A. Yes.
- Q. All right. And it's your opinion, is it not, that the pH of Vasopressin formulations had already been optimized as of the priority date of the patents-in-suit; correct?
- A. Optimized within -- within a range of around 3.5. I think in my report I mentioned that that range is rather broad.
- Q. Actually, I think what you said in your report, and you can correct me if you don't recall, is that by Bi and Singh determined the stability of Vasopressin solutions under various pH environment and found favorable stability at a pH of around 3.5. Those were your words.

Do you recall that?

- A. Yes. Around 3.5 is important.
- Q. And when someone says around 3.5 to a scientist in this space, especially in light of Bi and Singh, that might tell somebody 3.4 to 3.6?

A. No. I disagree. With respect to Bi, my recollection of that study is it was done at rather coarse pH units. It wasn't done at the tenth of a pH unit that we're seeing, for example, that Par cited. I think it was 3.5, 3.75, 3.4.

- Q. But at a minimum, your opinion was that those of skill in the art believed that the pH had already been optimized on the priority date; is that correct?
- A. Optimized within a rather broad range.
 - Q. You didn't say the words within a rather broad range in your report or deposition, did you? That's something that you have now added to your testimony today; isn't that correct?
 - A. I disagree. In my report I do remember talking about the broad range of stability of Vasopressin. That's I think consistent with my testimony here.
 - Q. Let's take a look at your report. Paragraph 46. You had a whole section on page 15 titled pH for Vasopressin formulations had already been optimized; correct?
- 19 A. Yes.

- Q. And then at the top of the page, page 16, you wrote, in earlier studies, Bi and Singh determined the stability of Vasopressin solutions under various pH environments and found favorable stability of Vasopressin at a pH of around 3.5; is that correct?
- 25 A. Correct.

Q. You had some testimony during the case about the normalization of data.

Do you recall that?

- A. Yes, I do.
- Q. And you spoke most of the time during your testimony about the data that is at 25 degrees Centigrade; is that correct?
- 8 A. Correct.

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- Q. Let me just back up a second because the presentation was so long and there are so many things thrown around, I think we maybe need to clarify a few things.
 - So the inventors studied the pH at different levels, determined impurity level and determined the assay, which is the amount of Vasopressin left in the formulation at two different temperatures, 25 degrees Celsius and 40 degrees Celsius; is that correct?
- 17 A. Correct.
- 18 Q. Okay. And how long were those studies?
- 19 A. Four weeks.
- Q. Four weeks. So the 25 degrees study at four weeks,
- 21 **25** degrees Celsius is roughly room temperature; is that
- 22 correct?
- 23 A. Yes.
- 24 Q. And in the room temperature data you don't really see 25 a signal in your view that there's anything really important

going on with respect to reducing degradation; is that correct?

- A. The degradation is occurring, but the amount of degradant is less at 25 degrees C than 40 degrees.
- Q. Right. You saw the curve that went kind of like this and it didn't in your view demonstrate criticality of the range; is that correct?
- A. Over, over the range above, and depending on whether you look at assay or impurity, but certainly, a range of above around let's say 3.4 or so. Depending on whether it's normalized or not, you could say that it's stable depending on really what data you look at.
- 13 0. **Now --**

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- A. Now, at the lower end, pH of 2.5, there's definitely an increase in impurity and a decrease in assay even at room temperature.
 - Q. I think I'm making your point, that you don't believe that the 25-degree Celsius room temperature data supported the criticality of the claims; correct?
- A. That's correct.
- Q. Correct. Now, the inventors, and as you know, Dr.

 Kirsch, believed, and Dr. Kannan testified by deposition

 that what was important here was the 40-degree data; is that

 correct?
- 25 A. Correct.

Q. And the 40-degree data is much more important here
because that is roughly 104 degrees, I think? Forty degrees
would be 104, I think?

- A. Oh, the conversion to Fahrenheit?
- 5 | Q. Yes. Plus 40 minus ten percent plus 32?
- A. Unless you're talking about weather, I just think in Celsius.
- 8 | Q. Okay.

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- 9 A. So I'm not sure.
- Q. All right. It's over a hundred degrees, and they do
 that to try to accelerate -- they call that accelerated
 stability so that they could see more quickly if there's an
 effect; is that correct?
 - A. Well, accelerated stability studies will enhance the degradation, but it's important to always to do both at room temperature and accelerated.
 - Q. Sure. You might not see a signal for a particular thing of interest at room temperature in a short study, but you might see that signal in a 40-degree accelerated study, which we tell you what's likely to happen months and months into the product; isn't that right?
 - A. Well, we do have evidence with respect to the Vasopressin stability experiments. We do have evidence of degradation at 25 degrees C at the low pH values.
- Q. Okay. Sir, you testified at deposition 50 times. You

know how this, works. My question: Isn't it true that accelerated stability studies are done at 40 degrees Celsius because the higher temperature accelerates the ability to see a signal in the data over room temperature? Isn't that the purpose of those studies?

- A. It will accelerate and you will see an enhancement in any degradation but I need to qualify that and say it's always important to relate to room temperature data because there could be a change in mechanism of the degradation pathway, and it's important here because you could have a pH stability experiment that gives you a pH of 3.8 at 40 degrees C, but that pH 3.8 may not be optimum at room temperature.
- Q. It's important to look at all the data; right?
- 15 A. Yes, sir.

- Q. And to make a judgment about whether that data supports criticality of the particular pH range; is that correct?
- 19 A. I agree.
- 20 Q. And when looking at all the data from the standpoint
 21 of a POSA in the peptide field, that is a judgment that you
 22 never really had to make yourself, is it?
 - A. Well, I certainly have studied stability of organic compounds and with respect to understanding the mechanisms of degradation, I understand how Vasopressin is falling

apart, if you will, under these conditions. So I know the significance of the data for this. I feel like I'm qualified to offer the opinions I'm offering here.

- Q. You feel comfortable to give opinions about what a POSA would consider valid data, valid conclusion based on general knowledge about pH and pharmaceutical, but even though you are not a POSA yourself. Correct? Is that fair?
- A. Well, I disagree with you that I'm not a POSA. I have the relevant experience in studying stability of organic compounds and pharmaceutics.
 - Q. Would you at least agree that Dr. Kirsch has more experience than you do?
 - A. Dr. Kirsch has a lot more experience with peptides than I do.
 - Q. And would you agree that Dr. Park has a lot more experience with peptides than you?
 - A. Yes.

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- 19 Q. But Dr. Park didn't testify on this issue, did he?
- 20 A. No, he didn't.
- 21 Q. Now, there was a lot of data and confusion about the slides that you presented and if we can put up slide 18, please.
 - This is one of the slides you used. This is the assay data for 40 degrees Celsius. I think you spent a few

1 seconds on this slide and kind of blew through it. 2 you are trying to show here is the graph and relate the 3 graph to the data that's in the top of the table; is that correct? 4 5 Correct. Α. Okay. Now, all the actual values, the pH values, the 6 7 assay percentage and the other assay percentage, those values, although not the percentage on the far right, all of 8 9 that data was submitted to the Examiner; right? 10 Α. The first three columns, but not the fourth. 11 Right. You didn't mean to imply that the Examiner 12 didn't have the actual values that were sent by the 13 patentee; is that correct? 14 Correct. Didn't have the decrease percent value. That's what I calculated, but she had everything else. 15 16 Right. But now with the 40 degree C data, you 17 complained that it had not been normalized; correct? 18 The impurity data was not normalized for both 25 and Α. 19 40 degrees C. 20 Right. And you showed us slide 22, if we can pull 21 that up. And you showed on the left-hand side the 25 C data in original form and normalized form; correct? 22

A. Correct.

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Q. And the 25 C data on the left doesn't show the critical range and the 25 C data on the right doesn't show

1 | the critical range either; is that correct?

- A. I don't follow the question. With respect --
- 3 \parallel Q. Neither of these slides show critical range of 3.7 to
- 4 3.9, is that right, or do you disagree? Maybe it does.
- 5 A. The critical range -- so what I've put in the pink is
- 6 relating to the data point for the pH back to the claimed
- 7 value.

- 8 \ Q. Yes. Well, you put that same box around 3.7 to 3.9;
- 9 | is that right?
- 10 | A. Yes.
- 11 Q. And that is the pH of the patent in this case?
- 12 A. Correct.
- 13 Q. Do you believe that the data on the left supports a
- 14 claim for a critical range?
- 15 A. So by the data on the left, if you mean the low, lower
- 16 | than 3.7, it's my opinion that the data -- oh, I'm sorry. I
- 17 | think your question is referring to the non -- the plot on
- 18 the left?
- 19 **Q. Yes?**
- 20 A. Yes. It's my opinion here with respect to the high pH
- 21 value, I think I testified there were values above 3.9. You
- 22 can see from the chart it's leveling off at this region.
- 23 Q. Okay. Do you believe that that data supports
- 24 criticality or not?
- 25 A. It does not support criticality.

Chyall - cross 1 Q. Do you believe the normalized data on the right 2 supports criticality or not? 3 Even less so, because the range of stability is 4 broader. 5 Okay. Now, you also did the same thing with the 40-degree data, but very curiously, after accusing -- strike 6 7 that. 8 But very curiously, on slide 25, your counsel 9 skipped over this slide. And to normalize the data on the 10 defendant is the non-normalized data; correct? 11 MR. LASKY: Objection. Mischaracterizes. 12 MR. BLACK: Well, if I'm wrong and I don't 13 understand the slide, that's okay, the witness can tell 14 me. 15 BY MR. BLACK: 16 Is the data on the left non-normalized data and the 17 data on the right normalized data? Both of these are normalized. The data on the left is 18 Α. a normalized assay. 19 20 I see. The data to the right is normalized 21 impurities; right? 22 Α. Correct. 23 Okay. I was wrong. But the data on the right looks like the values in the critical range are lower than the 24

other values, so normalization would support criticality,

wouldn't it?

A. It's my understanding there has to be a difference in kind and performance and what I'm showing here for both charts is that we -- we see if there's any difference of the values outside the claimed range, they're just slightly higher than the values that are in the claimed range, and I think equally as important, more important, some of the measured value at pH 3.5 and 3.6 for the assay, those values are lower than what's in the claimed range. And then on -- so that was with respect to assay.

With respect to impurities, we see examples of that on the high end of the claimed range.

Q. You gave an opinion about whether something was scientifically reasonable in withholding data.

Do you recall that?

- A. Yes, I do.
- Q. But, of course, there are a lot of reasons why data might not be sent to the Patent Office. Somebody might have a view that doesn't need to be sent to the Patent Office. Somebody could make a mistake. Somebody at the company could forget to send it along. There are a lot of human reasons why someone would not send data, which would explain why you're sending data to the Patent Office; is that correct?
- A. With respect to when data is sent, yes, but also I'm

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Chyall - cross

looking at the declaration and what was said by the inventor in forming my opinion. I understand that. But you would agree that there's 0. no evidence that the inventors were somehow deliberate in putting the wrong data in front of the Patent Office. have no evidence for that, do you? I believe the inventors had the normalized data and Α. didn't submit it. But you have previously testified that you didn't have any information, any evidence that the inventors had deliberately withheld data; is that correct? Α. Correct. I pass the witness. MR. BLACK: THE COURT: Thank you. Any redirect? MR. LASKY: No redirect, Your Honor. THE COURT: I just have a quick question for you. So when I look at the slides that you said you prepared --THE WITNESS: Yes. THE COURT: -- and I didn't see a slide -- well, first of all, I thought you testified that the Patent Examiner was aware that there was a November and then the March studies. THE WITNESS: Correct.

Chyall - redirect

1 THE COURT: So nobody hid that from the Patent 2 Examiner? 3 THE WITNESS: That's correct. THE COURT: Okay. And so in terms of 4 5 recognizing that these two studies were conducted at two dates, the Patent Examiner would have been aware of that? 6 7 THE WITNESS: Yes, she was. 8 THE COURT: Okay. Thank you. 9 MR. LASKY: May I ask a few questions to follow 10 up on that? 11 THE COURT: Yes. 12 BY MR. LASKY: 13 Although the Examiner was aware that there were two 14 studies conducted -- well, what did the inventors say was the reason for the break when she questioned whether that 15 16 was due to the fact that there were two studies or something 17 else? 18 The inventor said that the break in the pH 3.4 and 3.5 19 was due to the pH and not due to the different studies. 20 And what was the representation Dr. Kannan relied upon 21 to represent that? He represented that the only thing that was not 22 Α. 23 normalized was the pH. 24 And was that statement true? Q. 25

Α.

No, it wasn't.

Chyall - redirect 1 Q. And was it true that the reason for the break was due 2 to pH? 3 No, because there was a huge amount of impurities, Α. 4 initial impurity at the pH 3.4 value that got carried into 5 the result that was --Were you here for the testimony of Dr. Kannan? 6 7 Α. Yes, I was. 8 And what did Dr. Kannan testify during his deposition 9 that the reason for the break was? 10 He testified that the reason for the break was due to Α. 11 the pH. 12 But in his deposition, what did he testify to? 0. 13 Oh, the impurities. At deposition he testified that 14 the impurities were higher and that could account for the 15 break. 16 Right. And was that consistent with what he told the 17 Examiner during prosecution? 18 Α. No. He told the Examiner that the break was due to pH 19 only. 20 MR. LASKY: Thank you. All right. Mr. Black, you retread 21 THE COURT: 22 If you want another retread opportunity, I will give that. 23 it to you.

MR. BLACK: It's late. Not necessary, Your Honor.

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where we are? Okay.

THE COURT: All right. Thank you. Thank you very much. You're excused. Thank you. THE WITNESS: (Witness excused.) MS. WACKER: Your Honor, I have a housekeeping matter with respect to Dr. Park. I accidentally said DTX-47. It was supposed to be DTX-45. The parties are getting a list together. it's okay with Your Honor, we'll submit that tomorrow They just want to get an e-mail so they can verify morning. all of the data. THE COURT: Okay. So then let's just do this since my deputy clerk had to leave. So we can offer for admission tomorrow morning the exhibits for Dr. Park and for Dr. Chyall. All right. Any other housekeeping measures we need to address? MR. BLACK: Time, Your Honor. How much time? THE COURT: I don't have a count right now. MR. BLACK: Okay. THE COURT: Wait. Maybe we do. We have the parties, a total of 153 minutes today and that Eagle used over 291. MR. BLACK: And is that sufficient to tell us

Great.

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THE COURT: I mean, frankly, I will say this, and this is not -- I don't mean to impugn Dr. Chyall at all, but I think other than the last two hours, of course, it's tedious to listen to deposition time. I know you have to do it, play depositions, but frankly, I think counsel have been very good about their use I don't think we're going to have a problem. of time. We're going to finish tomorrow. Right? MR. BLACK: We will finish tomorrow. MR. HALES: Yes. We will finish tomorrow. Ι have a question I quess in relation to our discussion yesterday about whether there are going to be closings. THE COURT: So we're not going to have closings tomorrow. MR. HALES: Okay. THE COURT: We're going to have briefing pursuant to the expedited schedule that I already set, and then what I'm thinking of is bringing you back for an oral argument soon. And we may have -- I mean, you should be prepared to argue your case tomorrow. Well, let me see. What did you begin the day with? I know you all asked my deputy clerk for the time. MR. HALES: Four -- I think I have it, Your Honor, if you want it? THE COURT: Yes. What do you think you have

1 left? 2 MR. HALES: So we started with four hours, 3 21 minutes this morning. 4 THE COURT: Okav. 5 MR. HALES: And then --6 MR. MOORE: No, no, no. Yesterday plaintiffs 7 used four hours and 21 minutes, I believe, and defendant used three hours and 37 minutes. 8 9 THE COURT: Okay. MR. MOORE: So with the minutes we've gotten 10 11 today, we need to make the calculation. 12 THE COURT: Why don't you do it. Take your 13 time. 14 MR. BLACK: While we're doing that, Your Honor, two other housekeeping things. 15 Defendants sent in a definition list in a 16 17 letter. 18 THE COURT: Yes. 19 MR. BLACK: We were about to send our own in, 20 but the definitions were actually fairly similar, so we're 21 going to adopt most of them, if not all of them, maybe. 22 THE COURT: Okay. 23 MR. BLACK: We're just going to look at it 24 tonight and send you some comments tomorrow. 25 THE COURT: I did notice like in the Yes.

1	middle of the either second or third page there were a bunch
2	of quotes with no authority cited for where they came. I
3	think maybe it was compounding.
4	MR. BLACK: Okay.
5	THE COURT: I would be interested where that
6	definition came from. That wasn't you.
7	MR. BLACK: Okay. So the thought, Your Honor,
8	is that we maybe have some back and forth tomorrow at the
9	end, not formal closings, and then we would do briefing and
10	you might bring us back for closing after that?
11	THE COURT: Yes, but I would like to hear the
12	time. Can I get the time?
13	MS. WACKER: The defendants have used
14	508 minutes, I believe, and Par has used 414.
15	THE COURT: What do you all have left?
16	MR. HALES: 92. Right? I think we had was
17	it ten or ten-and-a-half. I can't remember. Ten hours.
18	THE COURT: Right.
19	MR. HALES: Ten hours.
20	THE COURT: You have an hour-and-a-half left.
21	MR. HALES: Correct.
22	THE COURT: Something like that. What does
23	plaintiff have left?
24	MR. BLACK: We have three-and-a-half.
25	MR. HALES: No.

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THE COURT: Around three. MR. BLACK: Three hours and six minutes. hours basically, and we have some depositions tomorrow. the order of battle for tomorrow is I believe -- well, your case. Yes. We have a few more deposition MR. HALES: clips to play. We trimmed some of those last night. We'll see if they are really needed or not. MR. BLACK: Amneal. MR. HALES: Yes. I'm just saying, in terms of our case, our witnesses, we would be turning it over to Amneal at that point. MS. WU: Your Honor, we have two witnesses, Dr. He'll be up tomorrow morning because it's pretty Winter. late now in Brussels followed by Dr. Marais. THE COURT: And you're going to get all of that done in and hour-and-a-half? MR. HALES: Well, it's not going to be easy. We've tried to be very efficient, but it has not been easy. MS. WU: Again, as we talked about previously, Dr. Winter's testimony is going to be very discrete issues. One of the more important things he's doing is presenting some testimony to support Dr. Marais's statistical analysis. THE COURT: Doctor -- hold on. I have to go

Dr. Marais's statistical analysis.

1 MS. WU: Dr. Marais is a statistician. 2 THE COURT: He hasn't testified yesterday? 3 MS. WU: No. THE COURT: I'm thinking, I don't remember the 4 5 statistical analysis. Okay. 6 MS. WU: We're previewing. 7 MR. BLACK: We do have some significant issues 8 with Dr. Winter going beyond the scope. I don't know 9 what they are planning for sure, but many of the exhibits in 10 his binder were not things that were mentioned in his 11 report. 12 THE COURT: All right. MR. BLACK: So hopefully, that will resolve over 13 14 the evening. 15 I thought we --MS. WU: All right. Well, I guess --16 THE COURT: 17 MR. BLACK: Not resolved. The document is not 18 referred to in his report. You can assume I'm going to 19 object. 20 THE COURT: Okay. Well, we should finish 21 tomorrow, that's for sure. I guess the only thing I would say to you is, I mean, I encourage you all to think about 22 23 transition statements, you know. 24 It's hard to follow some of this stuff and 25 what's the relevance of it and you do have to think about

how you present stuff to me.

That was really confusing, the stuff about slides and the documents and what's in evidence, and then thinking about, so is this inequitable conduct this is about, is it about obviousness, is it about criticality? You know, you expect a lot and that's on me, I guess, the expectations, but I just can't keep up with it all.

Just for what it's worth, you ought to think about that. I left on lunch. I'm working on two other cases. I was here writing a Markman opinion before the case started. Then we had an 8:00 o'clock telephone call. It's hard to juggle it all and I might want to think about it.

MR. BLACK: Yes.

THE COURT: And also think about picking your spots and your arguments, you know. I always say you lose low hanging fruit. You bring low hanging fruit, my attention is already gone.

Anyway, a short way of saying, or maybe a long way of saying think about that. Transition statements often help.

MR. BLACK: Yes.

THE COURT: I'm about to move into this area because it's going to relate to blank, then I'm like, okay, I kind of get it.

MR. HALES: Noted. I appreciate it.

THE COURT: And so, Mr. Black, one thing I was thinking is, you know, it does help sometimes if you want to make some point tomorrow before you leave when stuff is fresh in my mind, especially since you've got some time.

MR. BLACK: Yes.

THE COURT: You know --

MR. BLACK: I will be happy especially if he doesn't get a response.

THE COURT: I try to be fair.

MR. BLACK: We'll keep it in mind having a short presentation. As you can see, Your Honor, this case, there are a lot of issues that unusual even for us and we do this all the time and some of the prior art issues and the way the art is being applied, the infringement issue, the FDA complexity.

THE COURT: I think the FDA thing is very, very interesting and, you know, and I'm not faulting anybody, but just -- I mean, part of me thinks we really ought to have an FDA person here who would say, this is what happened. I just found quickly online like a two-minute search, FDA guidance about how you define the RLD and it looks like it's kind of complex.

There's a reference in one of the legal filings with the ANDA saying I guess there was a citizen's petition to find out whether their withdrawal was for safety or

efficacy, and that seems to impact whether something can be deemed an RLD. I have not studied the statute. I don't even know. But I do find that interesting.

And then I find them very interested in how

Eagle filed -- I mean, frankly, this occurred to me, which

is, well, if the ANDA is changed -- sorry, if the NDA is not

original, right, it's reformulated, but the original is

withdrawn before Eagle files its ANDA, like, why are you

filing an ANDA?

I mean, part of me says, do you file an -- I
mean, in other words, if your claim is you're not -- the RLD
is not the reformulated ANDA, then why do you file an ANDA?
But I think it's more complicated than that.

MR. BLACK: It's a very complicated topic.

THE COURT: All right.

MR. HALES: It is complicated. I think maybe the simple thing is, and I'm not an expert in this area, but unless there's a safety or efficacy based reason that one of those would be withdrawn, those various labels still exist.

THE COURT: I think it's more complicated than that. I think there's something in the statute that there are like maybe four or five paragraphs that say when something is no longer an RLD. I don't know. You all referred to it as the RLD. That seems every case I've read, everything says the RLD, yet it seems like we have two RLDs?

MR. BLACK: We have one RLD with different versions. It's not different enough to be a separate RLD. It's different enough, the pH changed.

THE COURT: That's what your side is.

MR. BLACK: Those are facts. The difference between the two products are facts, but it's still the same RLD, I think. Anyway, it is what it is.

THE COURT: It is. I'm just saying there's a lot of complexity.

MR. BLACK: Yes.

THE COURT: What I find with the lawyers sometimes, a lot of assumptions. I am dealing with an ANDA case where the parties just briefed all of this stuff and nobody thought to kind of talk to me and say, well, whether the pre-AIA or post-AIA and whether you ought to apply Section 120 and it's a different version, and nobody briefed it, nobody thought about it. It was all just kind of treated.

And I've got to write the opinion. I want to explain it. If I am going to write something, I want to explain it to somebody who is like me, high school, and when you try to do that, it is really hard when the lawyers haven't jointed the issues. You can only join so many issues.

I feel like that was it. I feel like there was

a big thing about --

MR. BLACK: Yes, I think you're right, Your Honor. We had a an FDA expert in the Amneal case, but they objected. That is water under the bridge. What I was thinking, I was wondering last night actually whether or not we should have had like an FDA tutorial about some of these things where we could have had more back and forth and colloquy, because that's really about what the law is and how things work. The testimony is kind of tedious and awkward.

THE COURT: It's not only tedious. Here's what I think. I think both sides, when it suits them, will say here's what the FDA will do. When it doesn't suit them, they'll say I don't know what the FDA does.

I'm thinking I bet the FDA would know.

Actually, having worked with the FDA as U.S. Attorney and seeing how things work, I'm not sure the FDA would come up with a straight answer. Okay. But anyway, we'll see you all tomorrow.

MR. HALES: Very quick question. I just want to understand.

Does that mean if I picked up on you, if we have any kind of closing remarks, is that within the time or is that going to happen the on fly?

THE COURT: What I would say to you, you should

1 count on closing your case, an hour-and-a-half, without 2 closing arguments and then we'll just see. 3 MR. HALES: Okay. THE COURT: You kind of get what you get. 4 5 all got the hours. MR. HALES: I understand that. The last 6 7 clarification --8 THE COURT: I would be ready to answer questions 9 if you have the extra time. 10 MR. HALES: Of course. I understand. 11 Just to clarify, I know the Court takes care of 12 Are these numbers already that you've given us, they this. 13 include, like, sometimes you have the sidebars and the like. 14 Is that adjusted? 15 THE COURT: You know what, I don't know. MR. BLACK: There's no appeal from that. 16 17 THE COURT: I don't think they fully were, which 18 is why I will be a little bit flexible. I do think you've 19 all, you know, done your part. 20 Yes? 21 MS. WU: Your Honor, one more thing, because every minute will be precious tomorrow. We have disclosed 22 23 the witness order as to plaintiffs, but I'm wondering if we 24 can switch it up and start with Dr. Winter first thing 25 tomorrow morning.

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I know it's the first time you're hearing this. I'm just springing it on you, but because there's going to be tech setup for Dr. Winter, it might be helpful to have him go first so that tech setup can happen even before you arrive and he's all set. We don't have to transition after the video for Dr. Winter? MR. BLACK: It's your case. If they want to end with the Kannan deposition clip. THE COURT: However you want to do it. I think it makes sense in terms of the tech stuff. MR. BLACK: Yes. THE COURT: Of course, I told you, I don't know exactly when I get out of my meeting. All right. Everybody, thank you very much. (Court recessed at 5:28 p.m.)

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